Extraction of Gold By Dr. Ahmed Ameed



Characteristics and uses of gold

Density: 19.3 g/cm^3 , T_m :1064 ° C

Shinny: for Jewelry

Durable: does not tarnish or corrode easily, sometimes used in dentistry to make the crowns for teeth.

Malleable and ductile: can be bent & flattened . For this reason it is used to make fine wires and thin, flat sheets

 Good conductor for heat & electricity: used in transistors, computer circuits & firefighting cloths.

Types of ores

Gold occurs principally as a <u>Native metal</u>, usually <u>alloyed</u> with <u>silver</u> (as <u>*Electrum*</u>), or with <u>mercury</u>(as an <u>*Amalgam*</u>). Native gold can occur as sizeable nuggets, flakes, grains or microscopic particles embedded in other rocks.

Ores in which gold occurs in chemical composition with other elements are comparatively rare. They include <u>calaverite</u>, <u>sylvanite</u>, <u>nagyagite</u>, <u>petzite</u> and <u>kren</u> <u>nerite</u>

Gold mining

 Hard rock mining – used to extract gold encased in rock. Either open pit mining or underground mining.



 Panning (الفصل) – sand and gravel (حصى) containing gold is shaken around with water in a pan. Gold is much denser than rock, so quickly settles to the bottom of the pan.



Gold mining

 Sluicing – water is channelled to flow through a sluice-box with riffles (تموجات) at the bottom which create dead-zones in the water current which allows gold to drop out of suspension.



 Sluicing and panning results in the direct recovery of small gold nuggets (خامات الذهب) and flakes.

<u>Gold ore processing</u> Gold cyanidation:

- The most commonly used process for gold extraction.
- Used to extract gold from low-grade ore.
- Gold is oxidised to a water-soluble aurocyanide metallic complex Au(CN)₂.
- In this dissolution process, the milled ore is agitated with dilute alkaline cyanide solution, and air is introduced:

 $4Au + 8NaCN + O_2 + 2H_2O \rightarrow 4NaAu(CN)_2 + 4NaOH$

<u>Gold ore processing</u> Gold cyanidation:

- At a slurry concentration of around 50% solids, the slurry passes through a series of agitated mixing tanks with a residence time of 24 hrs.
- The gold-bearing liquid is then separated from the leached solids in thickener tanks or vacuum filters & the tailings (مخلفات المعالجة) are washed to remove Au and CN⁻ prior to disposal.

Gold ore processing Heap Leaching

- Is an alternative to the agitated leaching process.
- Drastically reduced gold recovery costs of low grade ore.
- Ore grades as low as 0.3 g per ton can be economically processed by heap leaching.



Gold ore processing Heap Leaching

- Generally requires 60 to 90 days for processing ore that could be leached in 24 hrs in a conventional agitated leach process.
- Au recovery is around 70% as compared with 90% in an agitated leach plant.
- BUT, has gained wide favour due to vastly reduced processing costs.
- Frequently, mines will use agitated leaching for highgrade ore & heap leaching for low grade ores that would otherwise be considered waste rock.

<u>Gold ore processing</u> Gold cyanidation:

Merrill-Crowe process

- Traditional method for Au recovery from pregnant cyanide solutions.
- Once dissolution of Au is complete, the remaining rock pulp if filtered off through various filters to produce a sparkling (لامع) clear solution.
- O₂ is removed from the clarified solution by passing the solution through a vacuum deaeration column.

<u>Gold ore processing</u> Gold cyanidation:

Merrill-Crowe process

- Zinc dust is then added to the cyanide solution to chemically reduce the gold to the metal.
 2Au(CN)₂ + Zn → 2Au+ Zn(CN)₄
- The metallic gold is then filtered out & refined.

<u>Smelting</u> of resultant powder (rich Au) into steel molds and the slug is remove from the top furnace.

Electrolysis is done to get high gold grade 99.999% purity.

Cyanide process of gold production

