Republic of Iraq

The Ministry Of Higher Education

& Scientific Research

بسم الله الرحمن الرحيم



University: Diyala College: Engineering Department: Chemical Stage: First Lecturer name: Anees A. Khadom Qualification: PhD Place of work: Diyala University

Flow up of implementation celli pass play

Course Instructor	Anees A. Khadom				
E-mail	aneesdr@yahoo.com				
Title	Principles of C	hemical Engine	ering (I)		
Course Coordinator	Annually				
Course Objective	This course is intended to serve as an introduction to the principles and basics of material balance and how to deal with materials changes.				
Course Description	 Introduce the principles and calculation of material balance. Assist in methods of problems solving. Review certain principles of applied physical chemistry. Study the behavior of gases, liquids, and solids. Units and dimensions. 				
Textbook	Himmelblau David M. "Basic Principles and Calculations in Chemical Engineering". ^V th Ed. ^Y .				
	Term Tests	Laboratory	Ouizzes	Project	Final Exam
Course Assessments	۳۰٪	•	1.%	-	٦.٪
General Notes	This subject is of chemical eng	very important i gineering.	n understanding	the principles a	nd calculations

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Course Weekly Outline

We	Date	Topes Covered	Lab. Experiment	Notes			
ek			Assignments				
١	۱۷ and ۱۸, Nov. ۲۰۱۶	General review of material balance, Dimensions Units and Their					
		Conversion					
٢	۲٤ and ۲۰, Nov. ۲۰۱٤	Moles, Density, and Concentration, Choosing a Basis.					
٣	• 1 and • 7, Dec. 7 • 1 ٤	Temperature, Pressure.					
٤	\cdot^{Λ} and \cdot^{q} , Dec. \cdot^{12}	Introduction to Material Balance					
0	1° and 17, Dec. 7.12	General Strategy for Solving Material					
		Balance Problems.					
٦	ττ and ττ, Dec. τ. 1 ε	Solving Material Balance Problems					
		for Single Units without Reaction,					
٧	Υ^{9} and Υ^{\bullet} , Dec. $\Upsilon^{\bullet} \Upsilon^{\varepsilon}$	The Chemical Equation and					
		Stoichiometry.					
٨	•• and •٦, Jun. 7.12	Material Balances for Processes					
		Involving Reaction.					
٩							
۱.							
11							
١٢							
١٣							
١٤							
10							
١٦							
Half	Half – year break						

) V	17 and 17 , Feb. 7.10	Material Balance Problems Involving Multiple Units.	
14	Υ and Υ , Feb. Υ .	Material Balance Problems Involving Multiple Units.	
١٩	• τ and • τ , Mar. τ • 10	Material Balance Problems Involving Multiple Units.	
۲.	• 9 and 1 •, Mar. ۲ • ۱ •	Recycle and the Industrial Application of Material Balances.	
71	γ and γ , Mar. γ , γ	Bypass and the Industrial Application of Material Balances.	
22	τ and τ , Mar. τ	Purge and the Industrial Application of Material Balances.	
۲۳	• \uparrow and $\bullet \lor$, April, $\uparrow \bullet \lor \bullet$	Ideal Gases	
7 5	1° and 1° , April $7 \cdot 1^{\circ}$	The Ideal Gas Law	
20	$\tau \cdot$ and $\tau \cdot$, April $\tau \cdot \cdot \circ$	Ideal Gas Mixtures	
22	$\gamma\gamma$ and $\gamma\Lambda$, April γ .	Partial Pressure	
۲۷	$\cdot \epsilon$ and $\cdot \circ$, May, $\gamma \cdot 1 \circ$	Critical pressure and temperature	
۲۸	11 and 17 , May, 7.10	Real gases and compressibility factor	
29	1^{A} and 1^{9} , May, $7 \cdot 10^{\circ}$	Material Balances Involving Ideal Gases	
۳.	۲۰ and ۲٦, May, ۲۰۱۰	Material Balances Involving Ideal Gases	
31	\cdot and \cdot \cdot , June, \cdot \cdot \cdot		

INSTRUCTOR Signature:

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