

Ministry of Higher Education and Scientific Research

Scientific Supervision and Scientific Evaluation Apparatus

Directorate of Quality Assurance and Academic Accreditation Department



Academic Program and Course Description Chemical Engineering Department



Introduction:

The educational program is a well-planned set of courses that include procedures and experiences arranged in the form of an academic syllabus. Its main goal is to improve and build graduates' skills so they are ready for the job market. The program is reviewed and evaluated every year through internal or external audit procedures and programs like the External Examiner Program.

The academic program description is a short summary of the main features of the program and its courses. It shows what skills students are working to develop based on the program's goals. This description is very important because it is the main part of getting the program accredited, and it is written by the teaching staff together under the supervision of scientific committees in the scientific departments.

This guide, in its second version, includes a description of the academic program after updating the subjects and paragraphs of the previous guide in light of the updates and developments of the educational system in Iraq, which included the description of the academic program in its traditional form (annual, quarterly), as well as the adoption of the academic program description circulated according to the letter of the Department of Studies T 3/2906 on 3/5/2023 regarding the programs that adopt the Bologna Process as the basis for their work.

In this regard, we can only emphasize the importance of writing an academic programs and course description to ensure the proper functioning of the educational process.

Concepts and terminology:

<u>Academic Program Description</u>: The academic program description provides a brief summary of its vision, mission and objectives, including an accurate description of the targeted learning outcomes according to specific learning strategies.

<u>Course Description:</u> Provides a brief summary of the most important characteristics of the course and the learning outcomes expected of the students to achieve, proving whether they have made the most of the available learning opportunities. It is derived from the program description.

Program Vision: An ambitious picture for the future of the academic program to be sophisticated, inspiring, stimulating, realistic and applicable.

<u>Program Mission:</u> Briefly outlines the objectives and activities necessary to achieve them and defines the program's development paths and directions.

Program Objectives: They are statements that describe what the academic program intends to achieve within a specific period of time and are measurable and observable.

Curriculum Structure: All courses / subjects included in the academic program according to the approved learning system (quarterly, annual, Bologna Process) whether it is a requirement (ministry, university, college and scientific department) with the number of credit hours. Learning Outcomes: A compatible set of knowledge, skills and values acquired by students after the successful completion of the academic program and must determine the learning outcomes of each course in a way that achieves the objectives of the program.

Teaching and learning strategies: They are the strategies used by the faculty members to develop students' teaching and learning, and they are plans that are followed to reach the learning goals. They describe all classroom and extracurricular activities to achieve the learning outcomes of the program.

Academic Program Description Form

University Name: University of Diyala Faculty/Institute: College of Engineering

Scientific Department: Chemical Engineering Department

Academic or Professional Program Name: Bachelor

Final Certificate Name: Bachelor of Science in Chemical Engineering

Academic System: Course

Description Preparation Date: 2025

File Completion Date: 14 / 4 /2025

Signature: مرا

Head of Department Name:

Lect. Dr. Muwafaq Mahdi Abd

Date: 14 / 4 / 2025

Signature:

Scientific Associate Name:

Prof. Dr. Jabar Qasim Jabar

Date: 14/4/2025

The file is checked by: Assist. Prof. Dr. Salah N. Farhan

Department of Quality Assurance and University Performance

Director of the Quality Assurance and University Performance Department: Date:

Signature:

Approval of the Dean

Prof. Dr. Anees Abdullah Khadom

1. Program Vision

The vision of the Chemical Engineering Department is to be recognized as one of the distinguished departments in its education, research and outreach programs.

- Hopping to be a world-renowned department, advancing the contributions of chemical engineering through innovation, research, education, and social responsibility.
- Making every effort to provide the student with the foundations of modern knowledge and scientific research methods in the fields of chemical engineering.
- Working to develop the students' personality to make them capable of innovation, leadership, self-learning, and teamwork.
- Developing curricula periodically and according to local and international standards.
- Opening horizons of cooperation between the Department of Chemical Engineering and the departments of faculties of the University of Diyala and the corresponding departments in Iraqi universities.

2. Program Mission

- Preparing engineers with competence and scientific knowledge in the of chemical engineering and its technological developments.
- Enabling the graduate student to possess the skills in designing production units, oil, petrochemical, food and pharmaceutical industries.
- Preparing engineers capable for operate and manage factories related to chemical engineering specializations by focusing on the theoretical aspects and linking them to the practical aspect.
- Preparing the graduates to continue postgraduate studies in various fields of chemical engineering.
- Study the market needs for new and necessary branches of chemical engineering and

implement it.

- •Make contact with the community's needs for chemical engineering specializations by preparing highly qualified graduates.
- Working to develop teaching and learning methods and adopting modern methods in addition to traditional one.
- Contributing to providing academic and scientific consultations and developing services in Diyala Governorate in particular and Iraq in general.

3. Program Objectives

- Graduating effective scientific cadres who are distinguished scientifically and practically and are characterized by sound professional ethics and honesty.
- Promoting scientific research and encouraging creativity and innovators in the fields and applications of chemical technology.
- Providing an environment for stimulating the scientific thoughts.
- acquiring the local and international academic accreditation.

4. Program Accreditation

The department submitted an application to obtain program accreditation from the Iraqi Council for Engineering Accreditation

5. Other external influences

All relevant ministries in dealing with this program, such as the Ministry of Oil, Industry, Environment, and others

6. Program Structure

| Program Structure | Number of Courses | Credit Hours | Percentage | Reviews |
|-----------------------------|----------------------|-------------------|------------|---------------------|
| Institution Requirements | 7 | 13 | 13.8% | |
| College Requirements | 2 | 8 | 8.5% | |
| Department Requirements | 19 | 73 | 77.7% | |
| Summer Training | 1 month | Without credit | - | Compulsory training |
| Others | | | | |

^{*} This can include notes whether the course is basic or optional.

| 7. Program Description | | | | | | | | | | |
|------------------------|-------------|------------------------------------|-------------|-----------|--|--|--|--|--|--|
| *** | G 1 | G V | Credit 1 | Hours | | | | | | |
| Year | Course code | Course Name | Theoretical | Practical | | | | | | |
| | E 101 | Mathematics I | 4 | 0 | | | | | | |
| | CHE 101 | Organic Chemistry | 2 | 2 | | | | | | |
| First Year | CHE 102 | Principles of Chemical Engineering | 4 | 0 | | | | | | |
| /Semester 1 | CHE 103 | Engineering Mechanics | 3 | 0 | | | | | | |
| | U 101 | Human Rights and Democracy | 2 | 0 | | | | | | |
| | U 103 | Computer skills | 1 | 2 | | | | | | |
| | E 102 | Mathematics II | 4 | 0 | | | | | | |
| | CHE 104 | Analytical Chemistry | 2 | 2 | | | | | | |
| Einst Voor | CHE 105 | Material Balance | 4 | 0 | | | | | | |
| First Year /Semester 2 | CHE 106 | Engineering Drawing | 1 | 3 | | | | | | |
| /Semester 2 | U 104 | English Language | 2 | 0 | | | | | | |
| | CHE 107 | Engineering Workshops | 0 | 3 | | | | | | |
| | UD 02 | Arabic Language | 2 | 0 | | | | | | |
| | CHE 201 | Engineering Mathematics I | 3 | 0 | | | | | | |
| | CHE 202 | Energy Balance | 4 | 0 | | | | | | |
| Second Year | CHE 203 | Fluid Flow I | 3 | 0 | | | | | | |
| /Semester 3 | CHE 204 | Physical Chemistry | 3 | 2 | | | | | | |
| | CHE 205 | Petroleum and Natural Gas | 3 | 2 | | | | | | |
| | CHE 206 | Computer Programming I | 2 | 2 | | | | | | |
| | UD205 | Extinct Ba'ath Party Crimes | 2 | 0 | | | | | | |
| | CHE 207 | Engineering Mathematics II | 3 | 0 | | | | | | |
| | CHE 208 | Fluid Flow II | 3 | 2 | | | | | | |
| Second Year | CHE 209 | Pollution and Industrial Safety | 3 | 0 | | | | | | |
| /Semester 4 | CHE 210 | Engineering Statistic | 1 | 0 | | | | | | |
| | CHE 211 | Computer Programming II | 2 | 2 | | | | | | |
| | CHE 212 | Materials Engineering | 3 | 2 | | | | | | |
| | UD206 | English Language | 2 | 0 | | | | | | |
| | UD207 | Arabic Language | 2 | 0 | | | | | | |

| A Unovelodes | . T. 1 1 | | | | | | | | |
|---|--|---|--|--|--|--|--|--|--|
| A- Knowledge | A- Kilowieuge | | | | | | | | |
| | ➤ Knowing the facts, concepts, principles and theories of chemical engineering, and understanding the determinant and constraints facing the engineer's work for the purpose of making the right decision. | | | | | | | | |
| 1- Knowledge and | | | | | | | | | |
| understanding | of making the right decision.Understanding basic mathematical derivations and linking | | | | | | | | |
| | - | a with equations and laws to determine overn the industrial unit. | | | | | | | |
| | _ | w the optimal conditions for industrial | | | | | | | |
| 2- Awareness and understanding | known or unknowr | | | | | | | | |
| | Analyze and discuexperiments to obtain | ain more data. | | | | | | | |
| | ➤ Design units and improvements. | processes and make the necessary | | | | | | | |
| 3- Ability to apply | ➤ The ability to appl jurisdiction. | y new technologies within the general | | | | | | | |
| | | nensive view of industrial engineering | | | | | | | |
| | problems, taking into account cost, safety and quality | | | | | | | | |
| Skills | Skills | | | | | | | | |
| 1- The ability to use | e a variety of sources | ➤ Using multiple techniques and | | | | | | | |
| of understanding. devices related to the specialty. | | | | | | | | | |

- 2- Conduct successful laboratory experiments or design a safe experiment. and extract important data.
- 3- Work ethically and have the ability to identify and identify risks.
- Using laboratory equipment to find data.
- ➤ Develop and provide a safe work environment by selecting the most appropriate devices and equipment.

Ethics

- 1- Professional work, taking into account costs and occupational safety.
- 2- Working in the spirit of one team and ensuring human victory
- 3- Anticipating problems and finding appropriate solutions to them.
- Ethics and professionalism of the profession.
- ➤ The impact of industrial activities on society, both negatively and positively.
- ➤ Compatibility with environmental issues and environmental preservation

9. Teaching and Learning Strategies

- 1. Theoretical lectures with the use of illustrations.
- 2. Practical laboratory application of concepts taught theoretically.
- 3. Assigning students to perform seminars by assigning them a topic to be discussed with their colleagues.
- 4. Solve problems, discuss them, and assign students some homework and reports through the e-learning platform.

10. Evalution Method

- Sudden exams (5) marks.
- Monthly exams (25) marks.
- Reports assigned to them (5) degrees.
- Homework assignments (5) marks.
- A final examination of the curriculum (60 marks).

11. Faculty

| Academic Rank | Specializat | Specialization | | | Tea | nber of ching |
|---------------------|---------------------------|------------------------------|--|--|-------|---------------|
| | General | Special | | | Staff | lecture |
| professor | Chemical Engineering | Corrosion | | | 1 | |
| professor | Chemical Engineering | Mass transfer | | | 1 | |
| Assistant Professor | Chemical Engineering | Biochemical Engineering | | | 1 | |
| Assistant Professor | Chemical Engineering | Electrochemistry | | | 1 | |
| Assistant Professor | College of Languages | Hebrew language | | | 1 | |
| Assistant Professor | Mechanical Engineering | Thermal engineering | | | 1 | |
| Assistant Professor | Civil Engineering | Environmental Engineering | | | 1 | |
| Assistant Professor | Communication engineering | Image processing | | | 1 | |
| Lecturer | Chemical Engineering | Unit operation | | | 1 | |
| Lecturer | Chemical Engineering | Fluid Flow | | | 1 | |
| Lecturer | Chemical Engineering | Mass transfer | | | 1 | |
| Lecturer | Chemical Engineering | Reactor Design | | | 1 | |
| Lecturer | Chemical Engineering | Corrosion | | | 1 | |
| Lecturer | Chemical engineering | Oil Refinery | | | 1 | |
| Lecturer | Nuclear engineering | Environmental Engineering | | | 1 | |
| Lecturer | Mechanical Engineering | Mechanical Design | | | 1 | |
| Lecturer | Science of Chemistry | Organic Chemistry | | | 1 | |

| Lecturer | Electrical Engineering | Power Converters | | 1 | |
|--------------------|------------------------|----------------------------|--|---|--|
| Assistant Lecturer | Petroleum Engineering | Drilling wells | | 1 | |
| Assistant Lecturer | Science of Chemistry | Physical Chemistry | | 1 | |
| Assistant Lecturer | General Law | Human Rights | | 1 | |
| Assistant Lecturer | Chemical Engineering | Biochemical Engineering | | 2 | |
| Assistant Lecturer | Chemical Engineering | Mass Transfer | | 2 | |

Professional Development

Orienting new faculty members

New teaching staff are developed by putting them in central development courses organized by the university, as well as by interacting with senior staff during periodic meetings in the department for the purpose of introducing them to the work contexts and informing them of directives and instructions, along with giving advice, daily guidance and continuous follow-up.

Professional Development of faculty members

Professional development for faculty members takes place through the Divisions of Continuing Education and Academic Affairs in the Deanship of the College and its corresponding departments in the University, which constantly work to hold discussion circles and specialized scientific seminars, while reviewing what is published on the Internet sites of books and periodicals in various scientific specializations.

12- Acceptance criterion

Admission is centralized by the Ministry of Higher Education and Scientific Research according to the grade point average of the students obtained in the sixth scientific stage.

13- The most important sources of information about the program

Diyala University website / College of Engineering / Department of Chemical Engineering Website of the Ministry of Higher Education and Scientific Research

14- Program development plan

- ➤ Development is carried out by focusing on the advanced scientific staff in the department and through the committees formed annually, especially the Scientific Committee and the Quality Assurance and Academic Accreditation Committee.
- ➤ By preparing evaluation studies to prepare and develop senior leadership cadres in all aspects of the educational institution.
- ➤ Equipping scientific laboratories with modern equipment and qualifying their cadres in order to improve the most efficient performance.
- > Develop future plans and work to implement them.
- ➤ Creating a kind of competition among researchers, honoring the distinguished ones and motivating them to give more.
- ➤ Working to create a kind of financial income for the department to sustain and develop the work
- Supporting the department's first-in-class admission program annually and enrolling them in postgraduate studies.
- ➤ Conducting a twinning process with advanced universities and providing training opportunities for teaching staff in those universities.

13

| | Program Skills Outline | | | | | | | | | | | | | | |
|----------------------------|------------------------|------------------------------------|--------|-----------|-----|-----------|-----------|----------|----------|----------|-----------|-----------|-----------|----------|-----------|
| | | | | | | | prog | gran | n Lea | arni | ng (| outc | ome | es | |
| Year/ | Course CourseName | Basic/ | Kr | owl | edg | e | | Sk | ills | | | Etł | nics | | |
| Level | Code | 5541.551.4411.5 | Option | A1 | A2 | A3 | A4 | B1 | B2 | В3 | B4 | C1 | C2 | С3 | C4 |
| | E 101 | Mathematics I | Basic | $\sqrt{}$ | √ | | | √ | | | | | | V | |
| | CHE 101 | Organic Chemistry | Core | $\sqrt{}$ | 1 | | | √ | | | | | | V | |
| First Year / | CHE 102 | Principles of Chemical Engineering | Core | √ | 1 | | | V | √ | √ | | V | V | V | |
| Semester 1 CHE 103 | CHE 103 | Engineering Mechanics | Basic | √ | 1 | √ | | V | V | | | V | √ | V | |
| | U 101 | Human Rights and Democracy | Basic | √ | 1 | √ | | V | √ | | | V | √ | V | |
| | U 103 | Computer skills | Basic | √ | 1 | √ | | V | V | | | V | √ | V | |
| | E 102 | Mathematics II | Basic | √ | 1 | √ | | V | √ | | | V | V | V | |
| | CHE 104 | Analytical Chemistry | Core | √ | 1 | √ | | V | V | | | V | √ | V | |
| | CHE 105 | Material Balance | Core | √ | 1 | √ | | V | √ | | | V | √ | V | |
| First Year / Semester 2 | CHE 106 | Engineering Drawing | Basic | √ | 1 | √ | | V | √ | | | $\sqrt{}$ | √ | V | |
| | U 104 | English Language | Basic | √ | √ | | | V | √ | √ | | $\sqrt{}$ | √ | V | |
| | CHE 107 | Engineering Workshops | Basic | √ | 1 | V | | V | √ | √ | | | √ | V | |
| | UD 02 | Arabic Language | Basic | √ | 1 | | | V | √ | | | V | √ | V | |

| | CHE 201 | Engineering Mathematics I | Basic | √ | √ | √ | $\sqrt{}$ | V | √ | $\sqrt{}$ | √ | $\sqrt{}$ | |
|-----------------------------|---------|---------------------------------|-------|-----------|----------|----------|--------------|----------|----------|--------------|----------|--------------|--|
| CHE 202 | | Energy Balance | Basic | √ | V | V | | V | √ | | V | V | |
| | CHE 203 | Fluid Flow I | Basic | 1 | V | √ | \checkmark | √ | V | | V | $\sqrt{}$ | |
| Second Year / Semester 1 | CHE 204 | Physical Chemistry | Basic | V | V | V | | V | V | | V | $\sqrt{}$ | |
| | CHE 205 | Petroleum and Natural Gas | Basic | V | V | √ | \checkmark | V | V | \checkmark | V | \checkmark | |
| | CHE 206 | Computer Programming I | Basic | V | V | V | \checkmark | V | √ | \checkmark | V | $\sqrt{}$ | |
| | UD205 | Extinct Ba'ath Party Crimes | Basic | V | V | V | \checkmark | 7 | V | \checkmark | 7 | $\sqrt{}$ | |
| | CHE 207 | Engineering Mathematics II | Basic | V | V | V | \checkmark | V | √ | \checkmark | V | $\sqrt{}$ | |
| | CHE 208 | Fluid Flow II | Basic | V | V | V | \checkmark | 7 | V | \checkmark | 7 | $\sqrt{}$ | |
| | CHE 209 | Pollution and Industrial Safety | Basic | $\sqrt{}$ | V | | \checkmark | 7 | V | \checkmark | 7 | \checkmark | |
| Second Year / | CHE 210 | Engineering Statistic | Basic | V | V | V | \checkmark | 7 | √ | \checkmark | 7 | $\sqrt{}$ | |
| Semester 2 | CHE 211 | Computer Programming II | Basic | $\sqrt{}$ | V | V | \checkmark | 7 | V | \checkmark | 7 | $\sqrt{}$ | |
| | CHE 212 | Materials Engineering | Basic | V | V | V | \checkmark | V | √ | \checkmark | V | | |
| | UD206 | English Language | Basic | V | V | V | | V | V | | V | V | |
| | UD207 | Arabic Language | Basic | 1 | √ | √ | \checkmark | √ | √ | \checkmark | √ | $\sqrt{}$ | |

Please tick the boxes corresponding to the individual program learning outcomes under evaluation.

FIRST YEAR (SEMESTER 1)



Ministry of Higher Education and Scientific Research - Iraq University of Diyala College of Engineering Department of Chemical Engineering



MODULE DESCRIPTOR وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | | | | | | |
|---|-------------|-------------------------|-------------------------|----------|-------------------------------|--------------|--|--|--|--|--|
| Module Title | Матнем | IATICS I | | M | Module Delivery | | | | | | |
| Module Type | BASIC | | | | | | | | | | |
| Module Code | E 101 | | | | Theory Lecture Tutorial | | | | | | |
| ECTS Credits | 6 | | | | | | | | | | |
| SWL (hr/sem) | 150 | | | | | | | | | | |
| Module Level | | 1 | Semester | (s) offe | ered | 1 | | | | | |
| Administering Department | | Chemical Engineering | College | Engin | eering | | | | | | |
| Module Leader | Ali I. Abda | lla | e-mail | alialn | uaimmy@uod | iyala.edu.iq | | | | | |
| Module Leader's Title | Acad. | Lecturer | Module Lo Qualificat | | | MSc | | | | | |
| Module Tutor | | | e-mail | | | | | | | | |
| Peer Reviewer N | lame | | e-mail | | | | | | | | |
| Review Commit Approval | ttee | | Version N | umber | 1.0 | | | | | | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | | | | | | |
|--|--|--|---|--|--|--|--|--|--|--|
| Prerequisite module | None None | Semester | - | | | | | | | |
| Co-requisites module | None | Semester | | | | | | | | |
| - | - | | | | | | | | | |
| Module Aims, Learning Outcomes, Indicative Contents and Brief Description أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | | | | | | | |
| Module Aims أهداف المادة الدر اسية | This module aims to provide students with a competence in the use of, mathematical technique solution of engineering problems. It will also foundation from which to develop solutions to a engineering problems that they will encoundergraduate engineering program of study. | es that are releventso give studen wider and deepounter through | vant to the lets a firm er range of lout their | | | | | | | |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | Preliminaries: Explain mathematical coordination, slope of line, shifting of lines Vectors: Demonstrate an understanding of vectors: Demonstrate an understanding variables, range and domain of function, type graphs. Limits and Continuity: Demonstrate an fundamental concepts of calculus including differentiability. Derivatives: Apply the techniques of different functions including transcendental functions Applications of derivatives: Apply the techniques of change, line mean value theorem and Initial value problem. Complex numbers: Demonstrate an understand with basic operations and their mather representations including Euler's Formula | etors in plane and of function are of functions understanding limits, continuation at different arization, curve anding of comples | d space. nd related and their g of the nuity, and nt types of ntiation to sketching, x numbers | | | | | | | |
| Indicative Contents المحتويات الإرشادية | The topics listed under the indicative content be areas of knowledge and understanding that successful completion of the module. The illustrated in the context of relevant engineering • Preliminaries Cartesian coordinates, polar angle of inclination. • Functions, types of functions, graph of the formation • Review of trigonometric function: graph range and domain of trigonometric functions | will be obtain mathematical to scenarios. coordinates, slo functions, domain of trigonometri | ined from topics are pe of lines, and range | | | | | | | |

continuity. **Transcendental functions**: Inverse function, graph of inverse function, Logarithmic and exponential functions, inverse trigonometric functions, hyperbolic functions, inverse hyperbolic functions. **Derivatives:** Definition, rules of derivative, Implicit differentiation, L hospital's rule, derivative of inverse functions Applications of derivatives: rate of change problems, Relative maximum and relative minimum, Curve sketching with 1st derivative, Linearization, Mean value theorem, Initial value problem,. **Complex numbers:** Basic definitions. The geometric representations of the complex numbers, argand diagram, Basic operations with complex numbers, Euler's Formula **Vectors**: Introduction to vectors This course lays the foundation for a robust understanding of mathematical concepts that underpin the various disciplines within engineering. It covers a breadth of topics ranging from coordinate systems, slopes of lines, and angles of inclination to the introduction of two- and three-dimensional coordinate systems. A focus is also given to the understanding and manipulation of functions, including domain and range determination and function composition. **Course Description** The course incorporates a substantial overview of trigonometry, limits, continuity, derivatives, including their applications in real-world engineering contexts in addition to complex numbers and their mathematical representation. By the end of the course, students will have a sound understanding of these principles, preparing them for more advanced engineering courses in their respective fields. **Learning and Teaching Strategies** استراتيجيات التعلم والتعليم Begin In Mathematics I, then employ a range of teaching strategies to ensure first-year engineering students fully grasp the various mathematical concepts. Instructional methods include interactive lectures, where core mathematical principles are explained in detail, and practical problem-solving sessions to provide hands-on learning experiences. Collaborative group work encourages peer-to-peer learning and reinforces understanding through shared insights. Regular formative assessments will be conducted to monitor students' **Strategies** understanding of the material, and feedback will be promptly given to guide their learning process. Instructors will maintain office hours for personalized support, and online resources will be available to supplement classroom instruction. Emphasis will be placed on relating mathematical concepts to realworld engineering applications to make the learning experience more relevant and engaging. These strategies aim to develop students' critical thinking skills, enhance their problem-solving abilities, and prepare them for advanced engineering studies.

| Student Workload (SWL) الحمل الدراسي للطالب | | | | | | | | |
|---|-----------|-----|--|-----|--|--|--|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | | | | | | | | |
| In class lectures | 55 | 78 | Structured SWL (h/w) | 5 | | | | |
| In class tests | 5 | 70 | الحمل الدراسي المنتظم للطالب أسبوعيا | 3 | | | | |
| Tutorial | 15 | | | | | | | |
| Final Exam | 3 | | | | | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Assignment Preparation for tests | 20 20 | 72 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 5.1 | | | | |
| Homework | 32 | | | | | | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 150 | | | | | | |

| Module Evaluation تقييم المادة الدراسية | | | | | | |
|--|--------------|---------------------|------------------|-----------------|------------------------------|--|
| | | Time(hr)/ Number | Weight (Marks) | Week Due | Relevant Learning Outcome | |
| Formative assessment | Quizzes | 3 | 20% (20) | 3,5, 10, 12, 14 | LO #1, 2, 3, 4,5 and 7 | |
| | Assignments | 6 | 10% (10) | 4, 8, 12 | LO # 1, 2, 3, 4, 5 and 6 | |
| | Home Work | 6 | 10% (10) | 2,5,7,9,11,13 | LO # 1, 2, 3, 4, 5,6 and 7 | |
| Summative | Midterm Exam | 2 hr | 10% (10) | 7 | LO # 1,4 | |
| assessment | Final Exam | 3 hr | 50% (50) | 16 | All | |
| Total assessment | | | 100% (100 Marks) | | | |

| | Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | |
|--------|--|--|--|
| | Material Covered | | |
| Week 1 | Cartesian coordinates, slope of lines, angle of inclination, functions, types of functions, graph of the functions, domain and range ,identifying functions, Circles and parabolas | | |
| Week 2 | Introduction to vectors | | |
| Week 3 | •Preliminaries Sum, differences, products and quotients of Composite functions, shifting a graph of a function, scaling and reflecting a graph of a function, Absolute value | | |

| Week 4 | •Review of trigonometric function graph of trigonometric function, range and domain, identities | | |
|---------|--|--|--|
| Week 5 | •Limits and Continuity Properties, limits involving infinity, continuity | | |
| Week 6 | •Transcendental functions Inverse function, graph of inverse function, Logarithmic and exponential functions, trigonometric functions , inverse trigonometric functions, hyperbolic functions hyperbolic functions | | |
| Week 7 | •Derivatives Definition, rules of derivative, slopes, tangent lines, chain rule, derivative of trigonometric functions, Implicit differentiation, L hospital's rule | | |
| Week 8 | derivative of inverse trigonometric functions, derivative of exponential and logarithmic functions | | |
| Week 9 | •Applications of derivatives Speed and acceleration, Relative maximum and relative minimum | | |
| Week 10 | Curve sketching with 1st and 2nd derivative | | |
| Week 11 | Linearization | | |
| Week 12 | rate of change problems | | |
| Week 13 | Mean value theorem -Initial value problem | | |
| Week 14 | Complex numbers: Basic definitions. The geometric representations of the complex numbers, argand diagram | | |
| Week 15 | Basic operations with complex numbers, Euler's Formula | | |
| Week 16 | Final Exam | | |

| Learning and Teaching Resources | | | | | |
|---------------------------------|---|---------------------------|--|--|--|
| | مصادر التعلم والتدريس | | | | |
| | Text | Available in the Library? | | | |
| Required Texts | George B. Thomas and Ross L. Finney, "Calculus and Analytic Geometry, Addison- Wesley | Yes | | | |
| Tiequireu Teiles | That yet Geometry, Teamson Wesley | | | | |
| | Thomas Calculus, by George B.Thomas,Jr,Elevnth Edition Media Upgrade 2008 | | | | |
| Recommended Texts | Calculus Early Transcendental (Sixth Edition) James Stewart | Yes | | | |
| Websites | | | | | |



Ministry of Higher Education and Scientific Research - Iraq University of Diyala College of Engineering Department of Chemical Engineering



MODULE DESCRIPTION FORM

نموذج وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | |
|---|-------------|----------------------|------------|------------------------|----------------------------|--|
| Module Title | Organic Che | mistry | | Module Delivery | | |
| Module Type | Core | | | Theory | | |
| Module Code | CHE 101 | | | Lecture | | |
| ECTS Credits | 6 | | | Lab Tutorial | | |
| SWL (hr/sem) | 79 | | | Seminar | | |
| Module Level | | 1 | Semester o | f Delivery | 1 | |
| Administering Department | | Chemical engineering | College | College of Engineering | | |
| Module Leader | Mohanad Ali | Sultan | e-mail | maalazzawi@uodiyala | naalazzawi@uodiyala.edu.iq | |
| Module Leader's Acad. Title | | Lecture | Module Le | ader's Qualification | MSc | |
| Module Tutor | | | e-mail | | | |
| Peer Reviewer Name | | | e-mail | | | |
| Scientific Committee Approval Date | | | Version Nu | umber 1.0 | | |

| Relation with other Modules | | | | | |
|-----------------------------|-----------------------------------|----------|--|--|--|
| | العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | Prerequisite module None Semester | | | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | |
|--|---|--|--|--|
| Module Objectives أهداف المادة الدر اسية | To develop problem solving skills and understanding of basic principles of organic chemistry. To understand sstructure, properties and synthesis, apply fundamental reactions of organic compounds. This course deals with the basic concept of organic chemistry. Recognize Structure Relationships Between Chemicals. To understand Stability-Reactivity Principles. Recognize and Apply Functional Groups. | | | |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | Discuss polarization of a bond with electronegativity. understand nuclephile and electrophile groups and their properties. Describe different bond types of carbon and its hybriorbitals .Express the differences between valence bond and molecular orbital approaches Evaluate effects of atomic properties on acidity and basicity. enlighten relationship between acidity constant pKa and an acid-base reaction. Relate structure of molecule with strength of acidity and basicity write side chain reactions of aromatic compounds. Explain the reactions and properties of halogen compounds. Identify alkane, alkene and alkyn. prepare alkane, alkene and alkynes using different methods. Interpret the reactions and properties of alchols and phenols. Recognize the main differences between the acidities of alcohols and phenols. Explain the reactions and properties of ethers and epoxides. Recognize the main differences between open chain ethers and epoxides. Recognize the concept of aromaticity and the main properties of aromatic compounds. Explain aromaticity concept. | | | |
| | Explain reactions of ketones. Explain preparation methods of aldehydes Describe preparation methods of ketones Evaluates the importance of carbonyl function in organic chemistry. Explain Carboxylic Acid Nomenclature. Describe Structure and Bonding Physical Properties . Identify Acidity of Carboxylic Acids | | | |
| Indicative Contents المحتويات الإرشادية | Introduction (3 hrs) Alkanes and Cycloalkanes (3 hrs) Alkyl halides (3 hrs) Alkenes and Cycloalkenes (3 hrs) Addition Reactions of Alkenes (3 hrs) Alkynes (5 hrs) Arenes and Aromaticity (5 hrs) Alcohols (5 hrs) | | | |

| | Aldehydes and Ketones (5 hrs) |
|---------------------|--|
| | Carboxylic Acids (5 hrs) |
| | • Carboxylic Acid Derivatives (2 hrs) |
| | • Amine (3 hrs) |
| | Hydrocarbons, aliphatic, ring, olefins, acytelines, aromatics, Orbitals, Thermal |
| | cracking process, hydrogenation, reduction, Special reactions: oxidation and |
| Course Description | combustion, addition, substitution, nitration, Sulphonation, halogenation, |
| وصف المادة الدراسية | polymerization, Halides, Alcohols and phenols, Ethers, Aldehydes and ketones, |
| | Carboxylic acids, esters, amino-acids, amides, Amines and some. |
| | |

| | Learning and Teaching Strategies |
|------------|--|
| | استر اتيجيات التعلم والتعليم |
| Strategies | Begin to establish a strong conceptual understanding of the principles of organic chemistry. Use real-life examples and measurements to help students relate abstract concepts to everyday experiences. Encourage discussions and questions to clear up any misconceptions. In addition, provide students with real-world problem scenarios that require the application of organic chemistry principles. Encourage active participation and group discussions to enhance critical thinking and problem-solving skills. Guide students through the |
| | problem-solving process and provide constructive feedback. |

| Student Workload (SWL) | | | |
|---|------------|--|-----|
| ١ اسبوعا | محسوب له ٥ | الحمل الدر اسي للطالب | |
| Structured SWL (h/sem) | | | |
| الحمل الدراسي المنتظم للطالب خلال الفصل | | | |
| In class lectures 28 | | Structured SWL (h/w) | |
| In class tests 2 | 79 | الحمل الدراسي المنتظم للطالب أسبوعيا | 5 |
| Tutorial 15 | | العمل الدراسي المستعم للتعالب المبورعيا | |
| Lab 30 | | | |
| Final Exam 4 | | | |
| Unstructured SWL (h/sem) | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | Unstructured SWL (h/w) الحمل الدر اسى غير المنتظم للطالب أسبو عيا | 4.7 |
| Libraray, dorm, home, memorize 49 | 71 | | |
| Preparation for tests 10 | | الحامل الدراسي غير المسطع لتصالب المبلوعيا | |
| Homework 12 | | | |
| Total SWL (h/sem) | 150 | | |
| الحمل الدراسي الكلي للطالب خلال الفصل | | 130 | |

| | Module Evaluation | | | | | | |
|------------------|--------------------------|------------|----------------|------------|----------------------|--|--|
| | تقييم المادة الدراسية | | | | | | |
| | | Time/numbe | Weight (Marks) | Week Due | Relevant Learning | | |
| | | r | weight (warks) | | Outcome | | |
| | Quizzes | 2 | 10% (10) | 2,4,8, | LO #1, #4 and #5, | | |
| | Quizzes | 2 | 10% (10) | 12 | #10 | | |
| Formative | Aggianmenta | 4 | 10% (10) | 5, 10, 12, | LO #5, #9 , #10, #11 | | |
| | Assignments | | | 14 | LO #3, #9 , #10, #11 | | |
| assessment | Projects / Lab. | 1 | 100/ (10) | Continuou | All | | |
| | | | 10% (10) | S | All | | |
| | Report | 1 | 10% (10) | 13 | LO #7, #9 , #11 | | |
| Summative | Midterm | 2hr | 100/ (10) | 7 | LO #1 - #5 | | |
| assessment | Exam | 2111 | 10% (10) | / | LO #1 - #3 | | |
| | Final Exam | 3hr | 50% (50) | 16 | All | | |
| Total assessment | | 100% (100 | | | | | |
| | | Marks) | | | | | |

| Delivery Plan (Weekly Syllabus) | | | | |
|---------------------------------|---|--|--|--|
| المنهاج الاسبوعي النظري | | | | |
| | Material Covered | | | |
| Week 1 | Introduction - Atoms, molecules, bonding, polar and nonpolar molecules, intermolecular forces, solubilities, Lewis structures, acids and bases. Writing Organic Structures, Bond strength, Intro. to electrophiles & nucleophiles | | | |
| Week 2 | Alkanes and Cycloalkanes: Introduction to Hydrocarbons Molecular Formulas, nomenclature, alkyl group, Rings (cycloalkanes, polycyclic compounds) | | | |
| Week 3 | Alkyl halides IUPAC Nomenclature of Alkyl Halides substitution reactions of alkyl halides. Elimination reactions Physical Properties of Alcohols and Alkyl Halides: Intermolecular Forces | | | |
| Week 4 | Alkenes and Cycloalkenes structure and bonding, nomenclature, Physical Properties of Alkenes, Preparation of Alkenes: | | | |
| Week 5 | Addition Reactions of Alkenes | | | |

| | Addition of Sulfuric Acid to Alkenes |
|----------|---|
| | Acid-Catalyzed Hydration of Alkenes |
| | Hydroboration—Oxidation of Alkenes |
| | Addition of Halogens to Alkenes |
| | Epoxidation of Alkenes |
| | Ozonolysis of Alkenes |
| | Reactions of Alkenes with Alkenes |
| | Alkynes |
| | Sources of Alkynes |
| | Nomenclature |
| Week 6 | Physical Properties of Alkynes |
| | Structure and Bonding in Alkynes: sp Hybridization |
| | Acidity of Acetylene and Terminal Alkynes |
| | Preparation of Alkynes by Alkylation of Acetylene |
| Week 7 | Mid-term Exam |
| | Alkynes |
| | Reactions of Alkynes |
| Week 8 | |
| WCCK 0 | Hydrogenation, Hydration of Alkynes Metal, Ammonia Raduction of Alkynes |
| | Metal—Ammonia Reduction of Alkynes |
| | Addition of Hydrogen Halides to Alkynes |
| | Arenes and Aromaticity |
| Week 9 | Benzene The Structure of Bonzone The Stability of Bonzone |
| VV CCK 9 | The Structure of Benzene, The Stability of Benzene |
| | Substituted Derivatives of Benzene and Their Nomenclature |
| | Physical Properties of Arenes Reactions of Arenes: |
| | |
| Week 10 | Oxidation of Alkylbenzenes, Reactions of Benzylic Halides |
| VVCCN 10 | Reactions of Benzylic Halides |
| | Addition Reactions of Alkenylbenzenes Hückel's Rule |
| | Alcohols |
| | Sources of Alcohols |
| | Preparation of Alcohols by Reduction of Aldehydes and Ketones |
| | |
| Week 11 | Preparation of Alcohols by Reduction of Carboxylic Acids Preparation of Alcohols from Enovides |
| | Preparation of Alcohols from Epoxides Preparation of Diele |
| | Preparation of Diols Reactions of Alcohols: |
| | Esterification, Oxidation of Alcohols |
| | Aldehydes and Ketones |
| | Nomenclature |
| Week 12 | |
| | Structure and Bonding: The Carbonyl Group Physical Properties |
| | Physical Properties Sources of Aldehydes and Veteros |
| | Sources of Aldehydes and Ketones |
| | Reactions of Aldehydes and Ketones |

| | Acetal Formation |
|-----------|--|
| | Carboxylic Acids |
| | Carboxylic Acid Nomenclature |
| Week 13 | Structure and Bonding |
| | Physical Properties |
| | Acidity of Carboxylic Acids |
| | Substituents and Acid Strength |
| | Salts of Carboxylic Acids |
| Week 14 | Sources of Carboxylic Acids |
| | Synthesis of Carboxylic Acids |
| | Reactions of Carboxylic Acids |
| | Amine |
| | Amine Nomenclature |
| | Structure and Bonding |
| Week 15 | Physical Properties |
| VVCCIX 10 | Basicity of Amines |
| | mines as Natural Products |
| | Preparation of Amines by Alkylation of Ammonia |
| | Reactions of Amines: |
| Week 16 | Final Exam |

| | Delivery Plan (Weekly Lab. Syllabus) المنهاج الاسبوعي للمختبر | | |
|---------|---|--|--|
| | Material Covered | | |
| Week 1 | Lab 1: Melting point determination | | |
| Week 2 | Lab 2: Boiling point determination | | |
| Week 3 | Lab 3: Simple Distillation | | |
| Week 4 | Lab 4: saturation and combustion tests for organic test | | |
| Week 5 | Lab5: Distinguish between aliphatic and aromatic hydrocarbons | | |
| Week 6 | Lab 6: Preparation of aspirin | | |
| Week 7 | Lab 7: Identification of functional groups | | |
| Week 8 | Lab8: Saponfication reaction | | |
| Week 9 | Lab 9: Preparation of ester | | |
| Week 10 | Lab 10: Identification of carboxylic acid and phenols | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | | | |
|--|--|------------------------------|--|--|--|--|
| | Text | Available in the Library? | | | | |
| Required Texts | Any textbooks for organic chemistry | Yes | | | | |
| Recommended Texts | Organic Chemistry 8th edition by Francis A. Carey University of Virginia and Robert M. Giuliano Villanova University Organic Chemistry (fifth edition)by Francis A. Carey Tata Mc Graw Hill publishing company Limited, New Delhi. | No | | | | |
| Websites | | | | | | |

APPENDIX:

| GRADING SCHEME مخطط الدرجات | | | | | | |
|--------------------------------|-------------------------|-------------|-----------|---------------------------------------|--|--|
| Group | Grade | التقدير | Marks (%) | Definition | | |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | | |
| | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors | | |
| Success Group (50 - 100) | C - Good | ختر | 70 - 79 | Sound work with notable errors | | |
| (30 - 100) | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | | |
| | | | | | | |
| Note: | | | | · | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.



Ministry of Higher Education and Scientific Research - Iraq University of Diyala College of Engineering Department of Chemical Engineering



MODULE DESCRIPTOR وصنف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | | | |
|---|------------|-------------------------|-------------------------|----------------|---------------------|-------|--|--|
| Module Title | Principles | s of Chemical Engineeri | ing | Мо | dule Deliver | y | | |
| Module Type | CORE | | | | Theory | | | |
| Module Code | CHE 102 | | | | Lecture Tutorial | | | |
| ECTS Credits | 7 | | | | Tutoriai | | | |
| SWL (hr/sem) | 78 | | | | | | | |
| Module Level | | 1 | Semester | er (s) offered | | 1 | | |
| Administering Department | | Chemical Engineering | College | Engine | ering | | | |
| Module Leader | Dr. Moham | nmed H. Msaed | e-mail Mhmmsaed1@uodiy | | yala.edu.iq | | | |
| Module Leader's Acad. Title | | Lecturer | Module Lo Qualificat | | | Ph.D. | | |
| Module Tutor None | | | e-mail | | | | | |
| Peer Reviewer Name | | None | e-mail | | | | | |
| Review Committee Approval | | | Version N | umber | | | | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | | |
|--|------|----------|--|--|--|--|
| Prerequisite module | None | Semester | | | | |
| Co-requisites module | None | Semester | | | | |

| المنوالة Aims into large scale industrial processes. This course serves as introduction to the principles and calculation techniques used in chemical engineering by developing knowledge and expertise in the basic principles chemical engineering. It also comprehensively introduces problem-solv methods for material balances without chemical reactions. Knowledge and Understanding Having successfully completed this module, the student will be able demonstrate knowledge and understanding of: Definition of the dimensions, units and their conversions. Describing the Dimensional Consistency (Homogeneity). Explanation the concentrations, mole and density. Discussion the Temperature and pressure. Showing the General strategy for solving material balance problem Solving material balance problems for single units without reaction of the dimensions of the strategy for solving material balance problems for single units without reaction of the dimensions of the strategy for solving material balance problems for single units without reaction of the dimension of the principles of chemical dimension of the dimension of the principles of the dimension o | Module Aims, Learning Outcomes, Indicative Contents and Brief Description | | | | | | | |
|--|---|--|--|--|--|--|--|--|
| المنوالة Aims into large scale industrial processes. This course serves as introduction to the principles and calculation techniques used in chemical engineering by developing knowledge and expertise in the basic principles chemical engineering. It also comprehensively introduces problem-solv methods for material balances without chemical reactions. Knowledge and Understanding Having successfully completed this module, the student will be able demonstrate knowledge and understanding of: Definition of the dimensions, units and their conversions. Describing the Dimensional Consistency (Homogeneity). Explanation the concentrations, mole and density. Discussion the Temperature and pressure. Showing the General strategy for solving material balance problem Solving material balance problems for single units without reaction of the dimensions of the strategy for solving material balance problems for single units without reaction of the dimensions of the strategy for solving material balance problems for single units without reaction of the dimension of the principles of chemical dimension of the dimension of the principles of the dimension o | ىختصر | أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | | | |
| Having successfully completed this module, the student will be able demonstrate knowledge and understanding of: Definition of the dimensions, units and their conversions. Describing the Dimensional Consistency (Homogeneity). Explanation the concentrations, mole and density. Discussion the Temperature and pressure. Showing the General strategy for solving material balance problem solving material balance problems for single units without reaction Indicative content includes the following. Understanding Definition the dimensions, units and the conversions(6 hrs). Understanding Dimensional Consistency (6 hrs). Express of concentrations, mole and density (8 hrs). Temperature and pressure (16 hrs). General strategy for solving material balance problems (6 hrs). Atterial Balances for Batch and Semi-Batch Processes (16 hrs). Solving material balance problems for single units without reaction (16 hrs). This module is designed to provide first-year Chemical Engineering, including the dimensions, units, and their conversional consistency (homogeneity), moles, density, concentration choosing a basis, temperature, pressure, introduction to material balance problems solving material balance problems solving material balance problems solving material balance problems, solving material balance problems single units without reaction. Learning and Teaching Strategies Learning to establish a strong understanding of the principles of chemical policy. Beginning to establish a strong understanding of the principles of chemical policy. | | Fundamentally, chemical engineering is the discipline that transforms scientific breakthroughs into large scale industrial processes. This course serves as an introduction to the principles and calculation techniques used in chemical engineering by developing knowledge and expertise in the basic principles of chemical engineering. It also comprehensively introduces problem-solving methods for material balances without chemical reactions. | | | | | | |
| demonstrate knowledge and understanding of: Definition of the dimensions, units and their conversions. Describing the Dimensional Consistency (Homogeneity). Explanation the concentrations, mole and density. Discussion the Temperature and pressure. Showing the General strategy for solving material balance problem. Solving the Material Balances for Batch and Semi-Batch Processes. Solving material balance problems for single units without reaction Indicative content includes the following. Understanding Definition the dimensions, units and the conversions(6 hrs). Understanding Dimensional Consistency (6 hrs). Express of concentrations, mole and density (8 hrs). Temperature and pressure (16 hrs). General strategy for solving material balance problems (6 hrs). Material Balances for Batch and Semi-Batch Processes (16 hrs). Solving material balance problems for single units without reaction (16 hrs). This module is designed to provide first-year Chemical Engineering, including the dimensions, units, and their conversions of the students with the foundational concepts of principles of chemical balances for batch and semi-batch processes, general strategy solving material balances for batch and semi-batch processes, general strategy solving material balance problems, solving material balance problems single units without reaction. Learning and Teaching Strategies Learning to establish a strong understanding of the principles of chemical balances of the principles of the p | | | | | | | | |
| Indicative Contents Understanding Review (4 hrs). Understanding Definition the dimensions, units and the conversions(6 hrs). Understanding Dimensional Consistency (6 hrs). Express of concentrations, mole and density (8 hrs). Temperature and pressure (16 hrs). General strategy for solving material balance problems (6 hrs). Material Balances for Batch and Semi-Batch Processes (16 hrs). Solving material balance problems for single units without reactic (16 hrs). This module is designed to provide first-year Chemical Engineeristudents with the foundational concepts of principles of chemical engineering, including the dimensions, units, and their conversional consistency (homogeneity), moles, density, concentrative choosing a basis, temperature, pressure, introduction to material balance material balances for batch and semi-batch processes, general strategy solving material balance problems, solving material balance problems single units without reaction. Learning and Teaching Strategies Hitalian | Outcomes | Definition of the dimensions, units and their conversions. Describing the Dimensional Consistency (Homogeneity). Explanation the concentrations, mole and density. Discussion the Temperature and pressure. Showing the General strategy for solving material balance problems. Solving the Material Balances for Batch and Semi-Batch Processes. | | | | | | |
| • Understanding Review (4 hrs). • Understanding Definition the dimensions, units and the conversions (6 hrs). • Understanding Dimensional Consistency (6 hrs). • Express of concentrations, mole and density (8 hrs). • Temperature and pressure (16 hrs). • General strategy for solving material balance problems (6 hrs). • Material Balances for Batch and Semi-Batch Processes (16 hrs). • Solving material balance problems for single units without reaction (16 hrs). This module is designed to provide first-year Chemical Engineering students with the foundational concepts of principles of chemical engineering, including the dimensions, units, and their conversed dimensional consistency (homogeneity), moles, density, concentration choosing a basis, temperature, pressure, introduction to material balance problems and semi-batch processes, general strategy solving material balance for batch and semi-batch processes, general strategy solving material balance problems, solving material balance problems single units without reaction. Learning and Teaching Strategies Punderstanding Dimensional Consistency (6 hrs). | | Solving material balance problems for single units without reaction. | | | | | | |
| Students with the foundational concepts of principles of chemic engineering, including the dimensions, units, and their conversion dimensional consistency (homogeneity), moles, density, concentration choosing a basis, temperature, pressure, introduction to material balance material balances for batch and semi-batch processes, general strategy solving material balance problems, solving material balance problems single units without reaction. Learning and Teaching Strategies استراتیجیات التعلم والتعلیم و Beginning to establish a strong understanding of the principles of chemical balance problems. | | Understanding Review (4 hrs). Understanding Definition the dimensions, units and their conversions(6 hrs). Understanding Dimensional Consistency (6 hrs). Express of concentrations, mole and density (8 hrs). Temperature and pressure (16 hrs). General strategy for solving material balance problems (6 hrs). Material Balances for Batch and Semi-Batch Processes (16 hrs). Solving material balance problems for single units without reaction | | | | | | |
| استر اتیجیات التعلم و التعلیم Beginning to establish a strong understanding of the principles of chemi | Course Description | This module is designed to provide first-year Chemical Engineering students with the foundational concepts of principles of chemical engineering, including the dimensions, units, and their conversion, dimensional consistency (homogeneity), moles, density, concentration, choosing a basis, temperature, pressure, introduction to material balance, material balances for batch and semi-batch processes, general strategy for solving material balance problems, solving material balance problems for single units without reaction. | | | | | | |
| Beginning to establish a strong understanding of the principles of chemi | | | | | | | | |
| | | <u> </u> | | | | | | |
| Strategies to help students relate the basic concepts of material balance to r industrial processes. Encouraging the discussions and questions to clear any misconceptions. In addition, encouraging active participation a | Strategies | Beginning to establish a strong understanding of the principles of chemical engineering, in particular, material balance. Using real industrial examples to help students relate the basic concepts of material balance to real industrial processes. Encouraging the discussions and questions to clear up any misconceptions. In addition, encouraging active participation and group discussions to enhance critical thinking and problem-solving skills. | | | | | | |

Guiding the students through the problem-solving process and providing constructive feedback.

| Student Workload (SWL) الحمل الدراسي للطالب | | | | |
|--|--------------------|-----|---|-----|
| Structured SWL (h/sem) Use In class lectures In class tests Tutorial Final exam | 57 3 15 3 | 78 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 5 |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Library, dorm, home memorizing Preparation for tests Homework | 40 40 17 | 97 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 6.5 |
| Total SWL (h/sem) الحمل الدر اسي الكلي للطالب خلال الفصل | | 175 | | |

| | Module Evaluation تقبيم المادة الدر اسية | | | | | | | |
|----------------------|---|---|-----------------|-------------------|--------------------|--|--|--|
| | Time/ Number Weight (Marks) Week Due Relevant Learning Outcome | | | | | | | |
| | Quizzes | 2 | 20% (20) | 2,4,6,9,11 and 13 | LO #1, 2, 6, and 7 | | | |
| Formative assessment | Assignments | 2 | 10% (10) | 3,5,7,9,11 and 14 | LO # 1,2,3,5 and 6 | | | |
| assessment | Home work | 1 | 10% (10) | 14 | LO # 4-7 | | | |
| Summative | Midterm Exam | 1 | 10 % (10) | 7 | LO # 1-4 | | | |
| assessment | Final Exam | 1 | 50% (50) | 16 | All | | | |
| Total assessment | | | 100% (100Marks) | | | | | |

| | Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري |
|---------|---|
| | Material Covered |
| Week 1 | Definition of dimensions and units |
| Week 2 | Conversion of units and conversion factors |
| Week 3 | Dimensional Consistency (Homogeneity) |
| Week 4 | Express of concentrations, mole and density |
| Week 5 | Express of concentrations and choosing a basis |
| Week 6 | Temperature |
| Week 7 | Pressure |
| Week 8 | Introduction to Material Balances and Open, Closed, Steady-State Unsteady-State Systems |
| Week 9 | The material balance for a single component process |
| Week 10 | Multiple Component Systems |

| Week 11 | Accounting for Chemical Reactions in Material Balances | |
|---------|---|--|
| Week 12 | Material Balances for Batch and Semi-Batch Processes | |
| Week 13 | General Strategy for Solving Material Balance Problems | |
| Week 14 | olving material balance problems for single units without reaction | |
| Week 15 | Solving material balance problems for single units without reaction | |
| Week 16 | Final Exam | |

| Learning and Teaching Resources | | | | |
|---------------------------------|--|------------------------------|--|--|
| مصادر التعلم والتدريس | | | | |
| | Text | Available in the Library? | | |
| Required Texts | David M Himmelblau, Basic principles and calculations in chemical engineering, Prentice Hall. | Yes | | |
| Recommended Texts | Richard M Felder & Ronald W. Rousseau Elementary Principles of Chemical Processes, Wiley India. | No | | |
| Websites | Nil | | | |

APPENDIX:

| GRADING SCHEME مخطط الدر جات | | | | |
|---------------------------------|-------------------------|-------------|-----------|---------------------------------------|
| Group | Grade | التقدير | Marks (%) | Definition |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance |
| Success Group (50 - 100) | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors |
| | C - Good | جيد | 70 - 79 | Sound work with notable errors |
| | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required |
| | | | | |
| Note: | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.



Ministry of Higher Education and Scientific Research - Iraq University of Diyala College of Engineering Department of Chemical Engineering



MODULE DESCRIPTOR وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | | |
|---|--------------|----------------------|----------------------------------|------------|---------------------------|--|--|
| Module Title | Engineerin | g Mechanics | | Mod | Module Delivery | | |
| Module Type | BASIC | | | | Theory | | |
| Module Code | CHE 103 | | | | Lecture - Tutorial | | |
| ECTS Credits | 5 | | | | | | |
| SWL (hr/sem) | 125 | | | | | | |
| Module Level | Module Level | | Semester | (s) offere | s) offered 1 | | |
| Administering Department | | Chemical Engineering | College | Engineer | ing | | |
| Module Leader | Khalid Ahmed | | e-mail | khaliden | khalideng@uodiyala.edu.iq | | |
| Module Leader's Acad. Title | | Lecturer | Module Leader's Qualification | | M. Sc. | | |
| Module Tutor None | | | e-mail | | | | |
| Peer Reviewer Name | | None | e-mail | | | | |
| Review Committee Approval | | | Version N | umber | 1.0 | | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | |
|---|--|----------|--|--|--|
| Prerequisite module | None | Semester | | | |
| Co-requisites module | None | Semester | | | |
| Module Aims, Learning Outcomes, Indicative Contents and Brief Description أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | | |
| Module Aims أهداف المادة الدر اسية | This course aims to teach the students to learn and recognize the Engineering Mechanics. The student learns the followings: 1. Understand and apply Newton's laws to analyze forces and predict force | | | | |

reactions.

- 2. Understand the concept of vectors to use them in analyzing forces, velocity, and related quantities.
- 3. Understand the concept of forces to apply them to determine their effects on mechanical systems.
- 4. To use free-body diagrams to represent and simulate mechanical systems in order to analyze forces and identify points of force application.
- 5. Learn how to Analyze forces in 2D mechanical systems using the concept of force components in two dimensions.
- 6. Learn how to analyze forces in 3D mechanical systems using the concept of force components in three dimensions.
- 7. Learn how to apply the principles of the moment of force to analyze and calculate moments and their effects on mechanical systems.
- 8. Understand the concept of coupling and use it to analyze and calculate its effects on mechanical systems.
- 9. Learn how to apply the principles of equilibrium for a rigid body to stabilize mechanical systems.
- 10. Understand the concepts of dry friction and friction forces to use them in analyzing and calculate friction in mechanical systems.
- 11. Apply the principles of equilibrium in bodies involving dry friction to analyze and calculate friction forces required for equilibrium in mechanical systems.
- 12. Learn how to determine the centroid and center of gravity of bodies to achieve stable system designs.
- 13. Calculate the center of volume and area to enhance the system design.
- 14. Learn how to apply the principles of the moment of inertia to improve systems design.
- 15. Learn how to use the Parallel Axis Theorem to determine the moment of inertia in optimizing the system design.

In general, students will be able to analyze and predict weights and forces, ensure equilibrium and stability in mechanical systems, and determine the effects and locations of forces to achieve better system designs.

Module Learning Outcomes

مخرجات التعلم للمادة الدراسية

On successful completion of this course, students will be able to:

- 1. Understand Newton's laws and apply them to analyze forces and predict force reactions.
- 2. Understand the concept of vectors and use them to analyze forces, velocity, and related quantities.
- 3. Understand the concept of forces and apply them to determine their effects on mechanical systems.
- 4. Understand the concept of free-body diagrams and use them to represent and simulate mechanical systems in order to analyze forces and determine their points of effect.
- 5. Understand the concept of forces in 2D components and apply them to analyze forces in mechanical systems.
- 6. Understand the concept of forces in 3D components and apply them to analyze forces in mechanical systems.
- 7. Understand the principles of the moment of force and use them to analyze and calculate moments and their effects on mechanical systems.
- 3. Understand the concept of coupling and apply it to analyze and calculate

| | coupling effects on mechanical systems. 9. Understand the principles of equilibrium for a rigid body and use them to stabilize mechanical systems. 10. Understand the concept of dry friction and friction forces and apply them to analyze and calculate friction forces in mechanical systems. 11. Understand the principles of equilibrium for bodies involving dry friction and apply them to analyze and calculate friction forces to achieve equilibrium in mechanical systems. 12. Understand the concept of the centroid and use it to determine the center of gravity and centroids of bodies to enhance system stability. 13. Understand the concepts of finding the center of volumes and areas and apply them to achieve stable system designs. 14. Understand the principles of the moment of inertia and apply them to improve system design. 15. Understand the concept of the Parallel Axis Theorem and use it to determine the moment of inertia for better system design. 16. Understand the concept of the Parallel Axis Theorem and use it to determine the moment of inertia for better system design. 17. Understand the concept of the Parallel Axis Theorem and use it to determine the moment of inertia for better system design. 18. Understand the concept of the Parallel Axis Theorem and use it to determine the moment of inertia for better system design. 19. Understand the concept of the Parallel Axis Theorem and use it to determine the moment of inertia for better system design. |
|--|--|
| Indicative Contents المحتويات الإرشادية | Introduction to statics science, Newton's fundamental laws [3 hrs] Vectors [3 hrs] Forces [3 hrs] Free body diagrams [3 hrs] Force analysis in 2D components [3 hrs] Force analysis in 3D components [3 hrs] Moment of force, Midterm Exam [3 hrs] Coupling [3 hrs] Equilibrium for a Rigid Body [3 hrs] Friction [3 hrs] Center of gravity [3 hrs] Centroid of areas [3 hrs] Moment of inertia [3 hrs] Parallel axis theorem [3 hrs] |
| Course Description | This course familiarizes students with the principles of applying Newton's laws of motion to solve engineering problems. Emphasis is placed on drawing free body diagrams and self-checking strategies. Topics include introduction to forces; 2D equilibrium of particles and rigid bodies; center of gravity and centroids; friction; force vectors analysis, equilibrium of particles, force systems resultant, moments and couples, equilibrium of rigid bodies; friction, center of gravity and centroids, moment of inertia. |
| | Learning and Teaching Strategies استراتیجیات التعلم والتعلیم |
| Strategies | The main strategy that will be adopted in delivering this module through classes and in involving the student in solving example problems, home works, class works, and reports/seminars to refine and expanding their critical thinking skills in this field. This will helps students to understand the course material. |

| Student Workload (SWL) الحمل الدراسي للطالب | | | |
|--|-----|---|-----|
| Structured SWL (h/sem) الحمل الدراسي المنتظم الطالب خلال الفصل In class lectures In class tests Final Exam | 63 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 4 |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Library, dorm, home memorizing Preparation for tests Homework | 62 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 4.1 |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | 125 | | |

| Module Evaluation تقييم المادة الدراسية | | | | | |
|--|--------------|-------------|---------------------|----------|---------------------------|
| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome |
| | Quizzes | 2 | 10% (10) | 5 and 11 | LO #1-11 |
| Formative assessment | Assignments | 2 | 20% (10) | 6 and 12 | LO # 1-12 |
| | Home work | 2 | 5% (10) | 4 and 9 | LO # 1-9 |
| | Report | 1 | 5%(10) | 15 | LO # 1-15 |
| Summative | Midterm Exam | 2 | 10% (10) | 7 | LO # 1-6 |
| assessment | Final Exam | 3 | 50% (50) | 16 | All |
| Total assessment | | | 100% (100 Marks) | | |

| Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | | |
|--|--|--|--|
| | Material Covered | | |
| Week 1 | Introduction to statics science, Newton's fundamental laws | | |
| Week 2 | Vectors | | |
| Week 3 | Forces | | |
| Week 4 | Free body diagrams | | |
| Week 5 | Force analysis in 2D components | | |
| Week 6 | Force analysis in 3D components | | |

| Week 7 | Moment of force, Midterm Exam |
|---------|--|
| Week 8 | Coupling |
| Week 9 | Equilibrium for a Rigid Body |
| Week 10 | Friction |
| Week 11 | Equilibrium of bodies involving dry friction |
| Week 12 | Center of gravity |
| Week 13 | Centroid of areas |
| Week 14 | Moment of inertia |
| Week 15 | Parallel axis theorem |
| Week 16 | Final Exam |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | | |
|--|---|---------------------------|--|--|--|
| | Text | Available in the Library? | | | |
| Required Texts | Engineering Mechanics SI Version, J. L. Meriam, and L.G. Kraige Sixth Edition Publisher: John Wiley and Sons Inc. | Yes | | | |
| Recommended Texts | Applied Statics and Strength of Materials Leonard Spiegel, George F. Limbrunner Publisher: Prentice Hall | Yes | | | |
| Websites | https://www.adelaide.edu.au/course-outlines/1098' | 75/1/sem-1/ | | | |

APPENDIX:

| GRADING SCHEME مخطط الدرجات | | | | | | |
|--------------------------------|-------------------------|-------------|-----------|---------------------------------------|--|--|
| Group | Grade | التقدير | Marks (%) | Definition | | |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | | |
| | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors | | |
| Success Group (50 - 100) | C - Good | جيد | 70 - 79 | Sound work with notable errors | | |
| (30 - 100) | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | | |
| | | | | | | |
| Note: | | | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.





MODULE DESCRIPTION FORM

نموذج وصف المادة الدراسية

| Module Information | | | | | | | | |
|------------------------------------|---------------|-------------------------|-------------------------------------|---------------|-------------------|-------|--|--|
| معلومات المادة الدراسية | | | | | | | | |
| Module Title | | حقوق الانسان و | Modu | ıle Delivery | | | | |
| Module Type | | | Supor | t | | | | |
| Module Code | | | U 101 | - | Theory Lecture | | | |
| ECTS Credits | | | 2 | | Seminar | | | |
| SWL (hr/sem) | | | | | | | | |
| Module Level | | 1 | Semester o | of Delivery 1 | | 1 | | |
| Administering Dep | partment | Chemical Engineering | College | Engineering | | | | |
| Module Leader | Mohamed Ali H | ameed | e-mail amohamed_902@uodiyala.edu.iq | | ala.edu.iq | | | |
| Module Leader's Acad. Title | | Assist. Proff. | Module Leader's Qualification | | alification | Ph.D. | | |
| Module Tutor | | | e-mail | | | | | |
| Peer Reviewer Name | | | e-mail | | | | | |
| Scientific Committee Approval Date | | | Version Nu | mber | 1.0 | | | |

| Relation with other Modules | | | | | | | | |
|-----------------------------|-----------------------------------|----------|--|--|--|--|--|--|
| | العلاقة مع المواد الدراسية الأخرى | | | | | | | |
| Prerequisite module | None | Semester | | | | | | |
| Co-requisites module | None | Semester | | | | | | |

Module Aims, Learning Outcomes and Indicative Contents

| | أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية |
|--|--|
| Module Objectives أهداف المادة الدراسية | 1- يتعلم الطالب خلال السنه الدراسية اساسيات حقوق الانسان و الديمقر اطية ما حقوقه كيف يد افع عنها بالطرق القانونية وما هي ضماناتها الداخلية و الدولية. 2-استحصال المعرفة في مجال الديمقر اطية و أنواع أنظمتها و اثر ها على حقوق الانسان . 3-تنمية شخصية الطالب وتعزيز وعيهم في الأنظمة السياسية الديمقر اطية وتفاصيلها وكيفية تطبيقها على ارض الواقع و اهمية ان يكون فعال في المجتمع من خلال احترامه لحقوق الاخرين ومعرفه ان الحقوق و الحريات تنتهي عند بداية حقوقهم وحرياتهم ويؤدي و اجباته بدلا من اكتساب الحقوق فقط. |
| Module Learning | 4- تعزيز ثقافة السلام القائمة على العدل والمساواة. 1 -تمكين الطالب من معرفة اساسيات الدفاع عن حقوقه وحقوق الاخرين بعد معرفتها ومعرفة أهميتها له وللمجتمع بصورة عامة وأيضا معرفه كل شخص حدود حقوقه وحريته . ٢- تمكين الطالب في المشاركة السياسية وذلك من خلال معرفته بأ همية مشاركته في الانتخابات وتأثير هذه المشاركة على سير الانتخابات وتشكيل السلطة فيما بعد . ٣- معرفه الطالب ضمانات حقوقه وحرياته وما هي مصادرها . |
| Outcomes مخرجات التعلم للمادة الدراسية | ٤ - معرفة الفرق بين الحقوق و الحريات. ٥ - تمكين الطالب من معرفة ما هي المفهوم العلمي للديمقر اطية وما هي جذور ها و انو اعها و اشكالها . ٢ - يتعلم الطالب كيف يؤثر النظام الديمقر اطي على حقوق الانسان وما هي العلاقة بينها . ٧ - ادر اك الطالب ضرورة ان يكون مو اطن فعال في المجتمع |
| | ايضا معرفه شروط الناخب وشروط المرشح للانتخابات. ٨- معرفه أنظمة الانتخابات وايهما افضل. ٩ - فهم الطالب للقانون الدولي لحقوق الانسان وايضا معرفة مختصرة عن المنظمات الدولية والية عملها كالأمم المتحدة ومنظمة الصليب الأحمر وغيرها. 1 الجزء الأول -تعريف حقوق الانسان وحقوق الانسان في |
| | الحضارات القديمة. (تعريف الحق وتعريف الانسان ومعرفة أهمية حقوق الانسان بالنسبة للإنسان و المجتمع أيضا دراسة حقوق الانسان في الحضارات كالحضارة المصرية و العراقية و اليونانية و الرومانية) (عساعات) الجزء الثاني معرف حقوق الانسان في الأديان السماوية و اهمها الإسلام (٢ساعة) |
| Indicative Contents المحتويات الإرشادية | مصادر حقوق الانسان تتضمن (مصادر دولية كالإعلان العالمي لحقوق الانسان و العهد ان الدوليان و المصادر الإقليمية التي تشمل الاتفاقيات الإقليمية كالاتفاقية الاوربية و الأمريكية و الدستور) (٢ساعة) ضمانات حقوق الانسان (كالضمانات الدستورية و القانونية |
| |) (الساعة) الدولية والإقليمية لحقوق الانسان (الساعة) الاتفاقيات الدولية والإقليمية لحقوق الانسان (الساعة) الحريات العامة وانواعها والمقارنة فيما بينها (الساعة) مستقبل حقوق الانسان والعولمة وحقوق الانسان (الساعة) تعريف وتاريخ وأنواع الديمقراطية (دراسة تعريف ونشأة وتطور الديمقراطية مبادئها وانواعها كالديمقراطية |

| | المباشرة وغير المباشرة والنظام الرئاسي والبرلماني) (٦سا عات) تعريف الانتخاب وشروطه وأنواع النظم الانتخابية وتعريف المجلس النيابي (٦سا عات) العلاقة بين الديمقراطية وحقوق الانسان (٢ساعة) | | | | | | |
|---|--|-----------------|---|-----|--|--|--|
| Course description وصف المادة | إن حقوق الإنسان والديمقراطية توصف بأنها موضوع مهم وحيوي اذ تسهم في تمكين الطالب من معرفة حقوقه وحرياته وأهمية التمسك بها والدفاع عنها . حيث يهدف إلى تعريف الطالب ما هي حقوق الإنسان ، وما تأثيرها وأهميتها على الإنسان والمجتمع بشكل عام ، وما هي العلاقة بينها وبين الديمقراطية . و دراسة الديمقراطية كدراسة علمية . | | | | | | |
| | Learni | ng and Tead | ching Strategies | | | | |
| | | التعلم والتعليم | استراتيجيات | | | | |
| Strategies | 1-زيادة وعي الطالب بأهمية معرفه حقوقه وواجباته اتجاه المجتمع وعلاقة حقوق الانسان بالنظام الديمقراطي 2-ثقافة عامة في مجموعة من المجالات ومنها المجال القانوني و السياسي والاجتماعي ورفع ثقة الطالب بنفسه من خلال ربط المادة النظرية بالواقع العملي | | | | | | |
| Student Workload (SWL) | | | | | | | |
| الحمل الدر اسي للطالب محسوب لـ ١٥ أسبو عا | | | | | | | |
| Structured SWL (h/sem) سي المنتظم للطالب خلال الفصل | الحمل الدر ا | 30 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 2 | | | |
| Unstructured SWL (h/ser غير المنتظم للطالب خلال الفصل | | 17 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 1.1 | | | |
| Total SWL (h/sem) الحمل الدر إسى الكلى للطالب خلال الفصل | | 50 | | | | | |

Module Evaluation

تقييم المادة الدراسية

| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome |
|----------------------|-------------|-------------|----------------|----------|------------------------------|
| Formative assessment | Quizzes | 2 | 10% (10) | 5 and 10 | LO #1, #2 #,3,and #6 #7#8 |
| | Assignments | 2 | 10% (10) | 2 and 12 | LO #3, #4 and #6, #7 |

| | Projects / Lab. | | | | |
|------------------|-----------------|-----|-----------|----|------------------|
| | Report | 1 | 10% (10) | 13 | LO #5, #8 and #9 |
| Summative | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - #7 |
| assessment | Final Exam | 3hr | 50% (50) | 16 | All |
| Total assessment | | | 100% (100 | | |
| Total assessment | | | Marks) | | |

Delivery Plan (Weekly Syllabus)

المنهاج الاسبوعي النظري

| | Material Covered |
|---------|---|
| Week 1 | محاضرة تعريفية عن المادة واهميتها |
| Week 2 | تعريف الحق و الانسان وحقوق الانسان و اهمية حقوق الانسان ,حقوق الانسان في الدين الإسلامي و الحضارات القديمة. |
| Week 3 | مصادر حقوق الانسان الدولية والإقليمية والمحلية. |
| Week 4 | ضمانات حقوق الانسان الدستورية والقانونية وضمانات حقوق الانسان على الصعيد الدولي. |
| Week5 | ضمانات حقوق الانسان في الإسلام |
| Week 6 | دور المنضمات الإقليمية في حماية حقوق الانسان. |
| Week 7 | خصائص حقوق الانسان وتعريف الحريات العامة وانواعه والمقارنة بينها وبين الحقوق القانون الدولي لحقوق الانسان والقانون الدولي الإنساني ومنظمة الصليب الأحمر. |
| Week 8 | مستقبل حقوق الانسان وسبل تطويرها . |
| Week 9 | العولمة وحقوق الانسان . |
| Week 10 | تعريف الديمقر اطية وتطورها التاريخي ومبادئها . الديمقر اطية بين العالمية والخصوصية . اشكال الديمقر اطية / الديمقر اطية المباشرة. |
| Week 11 | الديمقر اطية شبه المباشرة والديمقر اطية التمثيلية / اركان النظام التمثيلي. |
| Week 12 | المجلس النيابي وانواعه / الانتخاب وشروطه / هيئة الناخبين. |
| Week 13 | تنظيم عملية الانتخاب / تحديد الدوائر الانتخابية / القوائم الانتخابية / المرشحون/ الحملة الانتخابية / التصويت . |
| Week 14 | نظم الانتخابات. |
| Week 15 | علاقة الديمقر اطية بحقوق الانسان وكيفية التأثير والتأثر فيما بينها . |
| Week 16 | الامتحان النهائي |

Learning and Teaching Resources

مصادر التعلم والتدريس

| | Text | Available in the Library? |
|----------------------|--|---------------------------|
| Required Texts | حقوق الانسان والطفل والديمقراطية /تأليف ما هر صالح علاوي ورياض عزيز هادي وعلي عبد الرزاق محمد واخرون / العاتك / بيروت / ۲۰۰۹ | نعم |
| Recommended Texts | عباس الدليمي / حقوق الانسان الفكر و الممارسة فخري رشيد ،صلاح ياسين /المنظمات الدولية / العاتك لصناعة الكتاب / بغداد عصام العطية / القانون الدولي العام / المكتبة القانونية /بغداد /2012 | У |
| Websites | | |





MODULE DESCRIPTION FORM

نموذج وصف المادة الدراسية

| Module Information معلومات المادة الدر اسية | | | | | | | |
|--|---------------|-------------------------|------------------------|----------------------------|-----|--|--|
| Module Title | Computer Ski | lls | Module Delivery | | | | |
| Module Type | Basic | | | | | | |
| Module Code | U 103 | | | Theory Lecture | | | |
| ECTS Credits | 4 | | | Lab | | | |
| SWL (hr/sem) | 100 | | | | | | |
| Module Level | | 1 | Semester | of Delivery 1 | | | |
| Administering Do | epartment | Chemical Engineering | College | College of Engineering | | | |
| Module Leader | Sura F. Yousi | f | e-mail | sura.fahmy@uodiyala.edu.iq | | | |
| Module Leader's | Acad. Title | Lecturer | Module L | eader's Qualification | MSc | | |
| Module Tutor - | | | e-mail | - | | | |
| Peer Reviewer Name - | | - | e-mail | - | | | |
| Scientific Commi Date | ttee Approval | | Version N | umber 1.0 | | | |

| Relation with other Modules العلاقة مع المواد الدراسية الأخرى | | | | | | |
|--|--|----------|--|--|--|--|
| Prerequisite module | None | Semester | | | | |
| Co-requisites module | Co-requisites module None Semester | | | | | |
| Module Aims, Learning Outcomes and Indicative Contents أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | | | |
| Module Objectives أهداف المادة الدر اسية | 1. Training students on the basics of using the comput with the necessary skills to deal with the computer | 1 0 | | | | |

| | 2. Assisting the student in distinguishing and developing his scientific and |
|--|---|
| | artistic abilities. |
| | 3. Enriching the student's skills to be able to deal with the computer with high |
| | efficiency. |
| | 4. Providing students with a way to use other modern technologies related to the |
| | educational process. |
| | 1. Enabling the student to know the concepts of information technology by |
| | learning the basics of the computer. |
| | 2. Enabling the student to know about the use of GUI operating systems. |
| Module Learning | 3. Enabling the student to deal with the skills of using the operating system |
| Outcomes | (Windows operating system) through exploring, customizing, and controlling |
| 2 1 - 11 - 1- 21 (21 - 12 - 12 - 12 - 12 | its settings. |
| مخرجات التعلم للمادة الدر اسية | 4. Enabling the student to work on the word processing program (Microsoft |
| الدراسية | Word). |
| | 5. Enabling the student to work on the spreadsheet program (Microsoft Excel). |
| | 6. Enabling the student to work on the presentation program (Microsoft |
| | PowerPoint). Indicative content includes the following |
| | Indicative content includes the following. |
| | • Course introduction (4 hrs) |
| Indicative Contents | • Working with GUI operating systems with a focus on Microsoft Windows OS (8 hrs) |
| المحتويات الإرشادية | Microsoft Office Word (MS Word) (16 hrs) |
| | Microsoft Office Excel (MS Excel) (16 hrs) |
| | Microsoft Office PowerPoint (MS PowerPoint) (16 hrs) |
| Description | Overview of computers: basic components, applications. GUI operating systems: Microsoft Windows operating system. Microsoft Office Word: getting started with Word, editing a document and formatting text and paragraphs, adding tables and inserting graphic objects, controlling page appearance and proofing a document. Microsoft Office Excel: getting started with Excel, sorting, selecting and subtotaling data, formulas and functions, worksheet formatting and presentation. Microsoft Office PowerPoint: getting started with PowerPoint, developing a PowerPoint presentation, adding graphical elements to your presentation and modifying objects in your presentation, adding graphical elements, tables and charts to your presentation and modifying objects in your presentation, prepare to deliver your presentation. |
| | in Jour presentation, prepare to deliver Jour presentation. |

| Learning and Teaching Strategies | | | | |
|----------------------------------|--|--|--|--|
| استراتيجيات التعلم والتعليم | | | | |
| | In this course, students are guided by: | | | |
| | Using different examples. | | | |
| | • Using different styles of discussion that aim to connect the theoretical and | | | |
| Strategies | practical sides. | | | |
| Strategies | • Asking questions and giving exercises that require analysis and conclusions | | | |
| | related to lectures. | | | |
| | • Encourage students to participate in discussions and do the practical work. | | | |
| | • Encourage students to work in groups. | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | |
|--|----|---|-----|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures 13 In class tests 4 Lab 43 Final Exam 4 | 64 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 4 | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Library, dorm, home memorizing 14 Preparation for tests 14 Homework 8 | 36 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2.4 | |
| Total SWL (h/sem) الحمل الدر اسي الكلي للطالب خلال الفصل | | | | |

| | Module Evaluation | | | | | | |
|-----------------------|---|------|----------------|------------|--------------------------|--|--|
| تقييم المادة الدراسية | | | | | | | |
| | Time/Nu Weight (Marks) Week Due Relevant Learning | | | | | | |
| | | mber | Weight (Marks) | Week Buc | Outcome | | |
| | Quizzes | 2 | 10% (10) | 6 and 12 | LO #1 to #3 and #4 to #6 | | |
| Formative | Assignments | 2 | 10% (10) | 2 and 13 | LO #3 to #6 | | |
| assessment | Projects / Lab. | 1 | 10% (10) | Continuous | All | | |
| | Report | 1 | 10% (10) | 13 | LO #3, #4 and #6 | | |
| Summative | Midterm Exam | 2hr | 10% (10) | 9 | LO #1 - #5 | | |
| assessment Final Exam | | 3hr | 50% (50) | 16 | All | | |
| Total assessm | Total assessment 100% (100) | | | | | | |

| | Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | | |
|---------|--|--|--|--|
| | Material Covered | | | |
| Week 1 | Overview of computers and their basic components and applications | | | |
| Week 2 | Operating computer using GUI operating systems | | | |
| Week 3 | The basic use of Microsoft Windows operating system | | | |
| Week 4 | Microsoft Office Word: Getting Started with Word | | | |
| Week 5 | Microsoft Office Word: Editing a Document and Formatting Text and Paragraphs | | | |
| Week 6 | Microsoft Office Word: Adding Tables and Inserting Graphic Objects | | | |
| Week 7 | Microsoft Office Word: Controlling Page Appearance and Proofing a Document | | | |
| Week 8 | Microsoft Office Excel: Getting Started with Excel | | | |
| Week 9 | Microsoft Office Excel: Sorting, Selecting and Subtotaling data | | | |
| Week 10 | Microsoft Office Excel: Formulas and Functions | | | |
| Week 11 | Microsoft Office Excel: Worksheet Formatting and Presentation | | | |
| Week 12 | Microsoft Office PowerPoint: Getting Started with PowerPoint | | | |

| Week 13 | Microsoft Office PowerPoint: Developing a PowerPoint Presentation, Adding Graphical | |
|--|---|--|
| week 13 | Elements to Your Presentation and Modifying Objects in Your Presentation | |
| Week 14 Microsoft Office PowerPoint: Adding Graphical Elements, tables and charts to Y | | |
| WEEK 14 | Presentation and Modifying Objects in Your Presentation | |
| Week 15 | Microsoft Office PowerPoint: Prepare to deliver your presentation | |
| Week 16 | Final exam | |

| | Delivery Plan (Weekly Lab. Syllabus) المنهاج الاسبوعي للمختبر | | | |
|--|---|--|--|--|
| | Material Covered | | | |
| Week 1 | Introduction to the lab and get started with use of computer | | | |
| Week 2 | Basic use of Windows operating system | | | |
| Week 3 | General view of Windows OS tools with a focus on Microsoft Office tools | | | |
| Week 4 | Microsoft Office Word: Getting Started with Word | | | |
| Week 5 | Microsoft Office Word: Editing a Document and Formatting Text and Paragraphs | | | |
| Week 6 | Week 6 Microsoft Office Word: Adding Tables and Inserting Graphic Objects | | | |
| Week 7 | Microsoft Office Word: Controlling Page Appearance and Proofing a Document | | | |
| Week 8 | Veek 8 Microsoft Office Excel: Getting Started with Excel | | | |
| Week 9 | Microsoft Office Excel: Sorting, Selecting and Subtotaling data | | | |
| Week 10 | Microsoft Office Excel: Formulas and Functions | | | |
| Week 11 | Microsoft Office Excel: Worksheet Formatting and Presentation | | | |
| Week 12 | Microsoft Office PowerPoint: Getting Started with PowerPoint | | | |
| Week 13 | Microsoft Office PowerPoint: Developing a PowerPoint Presentation, Adding Graphical | | | |
| Elements to Your Presentation and Modifying Objects in Your Presentation | | | | |
| Week 14 | Microsoft Office PowerPoint: Adding Graphical Elements, tables and charts to Your | | | |
| | Presentation and Modifying Objects in Your Presentation | | | |
| Week 15 | Microsoft Office PowerPoint: Prepare to deliver your presentation | | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | | | |
|--|--|---------------------------|--|--|--|--|
| | Text | Available in the Library? | | | | |
| Required Texts | Joan Lambert and Steve Lambert, Windows 10 step by step, 1st Edition 2015. Joan Lambert and Curtis Frye, Microsoft Office 2016 step by step, 1st Edition 2015. | Yes | | | | |
| Recommended Texts | Michael Miller, ABSOLUTE BEGINNER'S GUIDE TO COMPUTER BASICS, 5th EDITION, QUE Indianapolis, Indiana 46240, 2010. Paul McFedries, TEACH YOURSELF VISUALLY MICROSOFT WINDOWS 10, ANNIVERSARY | No | | | | |

FIRST YEAR (SEMESTER 2)





MODULE DESCRIPTOR وصف المادة الدراسية

| Module Information معلومات المادة الدر اسية | | | | | | | |
|--|-------------|----------------------|--------------------------|-----------|----------------|-------------------|------------|
| Module Title | Mathematic | Mathematics II | | | | ıle Delivery | |
| Module Type | Basic | | | | | | |
| Module Code | E 102 | | | | | Theory Lecture | |
| ECTS Credits | 6 | | | | | Tutorial | |
| SWL (hr/sem) | 150 | | | | | | |
| Module Level 1 | | Semester (| s) of | offered 2 | | 2 | |
| Administering De | epartment | Chemical Engineering | College | En | gineeri | ng | |
| Module Leader | Ali I. Abda | lla | e-mail | alia | <u>alnuain</u> | nmy@uodiy | ala.edu.iq |
| Module Leader's Acad. Title | | lecturer | Module Lo Qualificati | | er's | | MSc |
| Module Tutor None | | e-mail | | | | | |
| Peer Reviewer Name None | | None | e-mail | | | | |
| Review Committee Approval | | | Version N | umb | er | 1.0 | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | |
|---|--|----------|---|--|--|
| Prerequisite module | E 101 | Semester | 1 | | |
| Co-requisites module | None | Semester | - | | |
| Module Aims, L | Module Aims, Learning Outcomes, Indicative Contents and Brief Description | | | | |
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | | |
| | This module aims to provide students with an understanding of, and competence in the | | | | |
| Module Aims | use of, mathematical techniques that are relevant to the solution of engineering | | | | |
| أهداف المادة الدراسية | problems. It will also give students a firm foundation from which to develop solutions | | | | |
| | to a wider and deeper range of engineering problems that they will encounter | | | | |
| | throughout their undergraduate engineering program of study. | | | | |

| | 8. Integration: Demonstrate an understanding of the fundamental concept of |
|---|---|
| | integration and antiderivative including types of integrations |
| | 9. Integration and transcendental functions: Extend the concept of integration to |
| | cover the integration of different types of transcendental functions |
| | 10. Numerical integration: Explain the fundamentals of numerical integration |
| | focusing on trapezoidal rule and Simpson's rule. |
| Module Learning | 11. Methods of integration: Apply the techniques of integration to evaluate the |
| Outcomes | integrals that cannot be solved directly. |
| Outcomes | 12. Application of definite integrals: Extend the concept of integration to solve |
| مخرجات التعلم للمادة الدراسية | several problems involving area, volume, length of curve, surface area by |
| محربت مصم عدده محرمت | revolution, center of mass and moment of inertia. |
| | 13. Area with polar coordinates: Demonstrate an understanding of polar coordinate |
| | system and its difference with Cartesian coordinate system, graphing and |
| | problems solution of such system. |
| | 14. Matrix: Explain the concept of matrix in mathematics, matrix algebra and |
| | solution of system of linear equations. |
| | The topics listed under the indicative content below are the underpinning areas of |
| | knowledge and understanding that will be obtained from successful completion of |
| | the module. The mathematical topics are illustrated in the context of relevant |
| | engineering scenarios. |
| | • Integration: Definition, antiderivative, definite and indefinite integral. |
| | Integration and transcendental functions: integration of trigonometric and |
| | inverse trigonometric functions, integration of exponential and logarithmic |
| Indicative Contents | functions, Integration of hyperbolic and inverse hyperbolic functions. |
| litarcative Contents المحتويات الإرشادية | Numerical integration: Introduction, trapezoidal rule and Simpson's rule. |
| المحتويات الإرساديات | • Methods of integration: Substitution method, integration by parts, |
| | Trigonometric substitution method, integration by partial fraction. • Application of definite integrals: Area, Volume, Lengths of curves in the |
| | plane, Areas of surfaces of revolution, Center of mass, moment of inertia. |
| | • Area of polar coordinates: Definition, polar equation, relating polar and |
| | Cartesian coordinates, Graph in polar coordinates, applications using polar |
| | coordinate system |
| | Matrix: definition, matrix algebra, Determinant of matrix, Grammar's rule, |
| | Inverse of matrix, Gauss Elimination Method This course discuss the foundation for a robust understanding of mathematical concepts |
| | that underpin the various disciplines within engineering. It covers the integration and |
| | its types followed by methods of integration. The concept of numerical integration is |
| | also highlighted. Students will be able to utilize integration to solve several problems |
| | such as area between curves and volume by revolution. A focus is also given to the |
| Course Description | understanding of polar coordinate system and how to graph the curves and solve |
| | difficult integral in an easy way using such system. Matrix topic is also covered in this course so the students will be able to solve system of linear equations using matrix in |
| | different approaches. By the end of the course, students will have a sound |
| | understanding of these principles, preparing them for more advanced engineering |
| | courses in their respective fields |
| | Learning and Teaching Strategies |
| | استر اتيجيات التعليم |
| | The main strategies include the following: |
| | • Beginning In Mathematics II, then employ a range of teaching strategies to ensure |
| | first-year engineering students fully grasp the various mathematical concepts. |
| Strategies | • Instructional methods include interactive lectures, where core mathematical |
| | principles are explained in detail, and practical problem-solving sessions to provide |
| | hands-on learning experiences. • Collaborative group work encourages peer-to-peer learning and reinforces |
| | • Collaborative group work encourages peer-to-peer learning and reinforces |

- understanding through shared insights.
- Regular formative assessments will be conducted to monitor students' understanding of the material, and feedback will be promptly given to guide their learning process.
- Instructors will maintain office hours for personalized support, and online resources will be available to supplement classroom instruction.
- Emphasis will be placed on relating mathematical concepts to real-world engineering applications to make the learning experience more relevant and engaging.

These strategies aim to develop students' critical thinking skills, enhance their problemsolving abilities, and prepare them for advanced engineering studies.

| Student Workload (SWL) الحمل الدراسي للطالب | | | | | |
|---|-----|---|-----|--|--|
| Structured SWL (h/sem) | | | | | |
| الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures 55 | | Structured SWL (h/w) | | | |
| In class tests 5 | 78 | الحمل الدراسي المنتظم للطالب أسبوعيا | 5 | | |
| Tutorial 15 | | | | | |
| Final Exam 3 | | | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Assignment 20 Preparation for tests 30 | 72 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 5.1 | | |
| Homework 22 | | | | | |
| Total SWL (h/sem) | 150 | | | | |

Module Evaluation تقییم المادة الدر اسیة

| | | Time/ Number | Weight (Marks) | Week Due | Relevant Learning Outcome |
|----------------------|--------------|------------------|----------------|-----------------|------------------------------|
| | Quizzes | 3 | 20% (20) | 3,5, 10, 12, 14 | LO #1, 2, 3, 4,5 and 7 |
| Formative assessment | Assignments | 6 | 10% (10) | 4, 8, 12 | LO # 1, 2, 3, 4, 5 and 6 |
| assessment | Home Work | 6 | 10% (10) | 2,5,7,9,11,13 | L0 # 1, 2, 3, 4, 5,6 and 7 |
| Summative | Midterm Exam | 2 hr | 10% (20) | 7 | LO # 1,4 |
| assessment | Final Exam | 3 hr | 50% (50) | 16 | All |
| Total assessment | | 100% (100 Marks) | | | |

| Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | | | |
|--|--|--|--|--|
| | Material Covered | | | |
| Week 1 | Integration: Definition, antiderivative, definite and indefinite integral | | | |
| Week 2 | Integration and transcendental functions: (trigonometric and inverse trigonometric functions, exponential and logarithmic functions) | | | |

| Week 3 | Integration and transcendental functions: |
|---------|--|
| WCCK 3 | Integration and transcendental functions (hyperbolic and inverse hyperbolic functions) |
| Week 4 | Numerical integration Introduction, trapezoidal rule and Simpson's rule |
| Week 5 | Methods of integration Substitution method, integration by parts |
| Week 6 | Methods of integration Trigonometric substitution method |
| Week 7 | Methods of integration Integration by partial fraction method. |
| Week 8 | • Application of definite integrals Areas under the curve, area between curves, |
| Week 9 | • Application of definite integrals Volume by revolution |
| Week 10 | • Application of definite integrals Length of curve in the plane, Area of surface of revolution |
| Week 11 | • Application of definite integrals Center of mass, moment of inertia |
| Week 12 | Application of definite integrals Area by polar coordinates |
| Week 13 | Matrix Definition, matrix algebra |
| Week 14 | Matrix Determinant of matrix, Grammar's rule |
| Week 15 | Matrix Inverse of matrix, Gauss Elimination Method |
| Week 16 | Final Exam |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|---|---------------------------|--|--|
| | Text | Available in the Library? | | |
| Required Texts | George B. Thomas and Ross L. Finney, "Calculus and Analytic Geometry, Addison- Wesley | Yes | | |
| Recommended Texts | Thomas Calculus, by George B.Thomas,Jr,Elevnth Edition Media Upgrade 2008 Calculus Early Transcendental (Sixth Edition) James Stewart | Yes | | |
| Websites | | | | |





MODULE DESCRIPTION FORM

نموذج وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | |
|---|--------------|--------------------|--------------|---------------------------------|-----------------|--|
| Module Title | Analytical C | hemistry | | Module Delivery | | |
| Module Type | Core | | | Theory | | |
| Module Code | CHE 104 | | | Lecture Lab | | |
| ECTS Credits | 6 | | | Tutorial | | |
| SWL (hr/sem) | 150 | | | Seminar | | |
| Module Level 1 | | Semester of | f Delivery 2 | | | |
| Administering Department Chemical Engineering | | | College | Engineering | | |
| Module Leader | Nabaa Burha | n Ali | e-mail | nabaa_burhan_eng@ | uodiyala.edu.iq | |
| Module Leader's | Acad. Title | Assistant Lecturer | Module Le | dule Leader's Qualification MSc | | |
| Module Tutor None | | e-mail | | | | |
| Peer Reviewer Name None | | e-mail | _ | | | |
| Scientific Committee Approval Date | | Version Nu | umber 1.0 | | | |

| Relation with other Modules | | | | | | |
|-----------------------------------|------|----------|--|--|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | | | |
| Prerequisite module | None | Semester | | | | |
| Co-requisites module | None | Semester | | | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | |
|--|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | |
| Module Objectives | 1.During the semester, the student learns an idea about analytical chemistry and | | |
| the main principles of analysis methods. | | | |
| | 2.Learn and understand the methods of instrumental analysis | | |

| | 3.Learn and understand ways to express the concentrations of solutions. 4.Learn and understand the methods of leaching in analytical chemistry theoretically and practically. 5.Identify the types of solutions and solve the required issues. 6.Learn and understand gravimetric methods of analysis. 7.Learn about complex acid-base systems. 8.Learn and understand the Nernst equation and the measurement of concentration by cell potential |
|---|--|
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | 1.Understanding and teaching students the concepts of analytical chemistry, which is one of the branches of chemistry. 2.Enabling students to obtain knowledge and understanding of the fields of analytical chemistry. 3.To make the student understand the basics and types of analytical chemistry, as well as enable students to obtain knowledge and understanding of the practical framework in the subject of analytical chemistry. 4.Enabling students to obtain knowledge and understanding of the different methods of analysis. 5.Enabling students to obtain knowledge and understanding to determine the appropriate methods of analysis for the required samples. 6.To make the student understand the solution of problems related to acid-base dissolutions. 7.Enable the student to obtain knowledge and understanding in the subject of chemical equilibrium (Equivalent calculations, balance in precipitation, partial precipitation) 8.Understanding and teaching students the effect of electrolytes on chemical balance 9.Understanding the student and teaching the student the methods of gravimetric analysis 10.Enable the student to obtain knowledge and understanding in the topic of titration in analytical chemistry (pH, titration charts, indicators of bases and acids, principles of titration equation) 11.Understanding and teaching students the Nernst equation and measuring concentration according to cell potential 12.Enable the student to obtain knowledge and understanding in the subject matter of inorganic materials and interactions |
| Indicative Contents المحتويات الإرشادية | Indicative content includes the following: The Nature of Analytical Chemistry (The Role of Analytical Chemistry , Quantitative Analytical Methods , A Typical Quantitative Analysis , An Integral Role for Chemical Analysis) [4 hrs] Calculations Used in Analytical Chemistry (Some Important Units of Measurement , Solutions and Their Concentrations, Chemical Stoichiometry) [6 hrs] Sampling, Standardization, and calibration (Analytical Samples and Methods, Sampling, Automated Sample Handling , Standardization and Calibration.[4 hrs] Aqueous Solutions and Chemical Equilibria (The Chemical Composition of Aqueous Solutions, Chemical Equilibrium, The Method of Successive Approximations, Buffer Solutions) [20 hrs] Effect of Electrolytes on Chemical Equilibria (The Effect of Electrolytes on |

| | Chemical Equilibria, Activity Coefficients) [8 hrs] |
|--------------------|---|
| | Solving Equilibrium Problems for Complex Systems(Solving Multiple- |
| | Equilibrium Problems Using a Systematic Method, Calculating Solubilities by |
| | the Systematic Method) [8 hrs] |
| | Gravimetric Methods of Analysis (Specific Surface Area of |
| | Colloids, Calculation of Results from Gravimetric Data, Applications of |
| | Gravimetric Methods) [6 hrs] |
| | Titrations in Analytical Chemistry(Some Terms Used in Volumetric Titrations, |
| | Standard Solutions ,Volumetric Calculations, Gravimetric Titrations |
| | ,Titration Curves [14 hrs] |
| | Complex Acid/Base Systems Complex Acid/Base Systems, Mixtures of |
| | Strong and Weak Acids or Strong and Weak Bases ,Polyfunctional Acids and |
| | Bases ,Buffer Solutions Involving Polyprotic Acids) [9 hrs] |
| | |
| | This course provides students an idea of analytical chemistry and its methods, |
| | perform the required analytical operations on samples, determining the |
| | appropriate methods for each sample, as well as identifying ways to express the |
| | concentrations of solutions and units used, solving the required problems, |
| | Equilibrium problems for complex systems are solved, Nernst equation and |
| Course Description | measurement of concentrations by potentials of cells, applying the necessary |
| | experiments in the analytical chemistry laboratory, Preparing primary and |
| | |
| | secondary standard solutions by applying theoretical calculations and then |
| | preparing solutions practically and learning about titration methods in |
| | chemistry. |

| Learning and Teaching Strategies | | | | | | |
|----------------------------------|---|--|--|--|--|--|
| | استراتيجيات التعلم والتعليم | | | | | |
| | 1. Explanation of analytical chemistry topics by subject specialists. | | | | | |
| Strategies | 2. provide them with practical problem-solving skills related to analysis. | | | | | |
| | 3. Topics related to chemical analysis methods are presented. | | | | | |
| | 4. Emphasis is placed on the issues of titration, its types, and performing the | | | | | |
| | titration in practice. | | | | | |
| | 5. Topics related to the calculation of equilibrium constants are presented. | | | | | |
| | 6. Explanation and clarification of some inorganic materials and their | | | | | |
| | interactions. | | | | | |

| Student Workload (SWL) lead lead lead lead lead lead lead lead | | | | | |
|---|--------------------------|----|--|---|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | | | | | |
| In class lectures In class tests Tutorial Lab Final Exam | 28 2 15 30 4 | 79 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبو عيا | 5 | |

| Unstructured SWL (h/sem) | | | |
|---|-----|--|-----|
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | 71 | Unstructured SWL (h/w) | 4.7 |
| Libraray, dorm, home, memorize 45 | /1 | الحمل الدراسي غير المنتظم للطالب أسبوعيا | 4.7 |
| Preparation for tests 14 | | | |
| Homework 12 | | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | 150 | | |

| Module Eval | uation | | | | تقييم المادة الدر اسية |
|-----------------------------|-----------------|-----------------|-------------------|----------------|---------------------------------|
| | | Time/N umber | Weight (Marks) | Week Due | Relevant Learning Outcome |
| Formative | Quizzes | 4 | 10% (10) | 3, 6,10 and 13 | LO #1, 2, 4, 5, 8, 9, 11 and 12 |
| assessment | Assignments | 4 | 10% (10) | 2, 5, 9 and 12 | LO #1, 4, 10 and 11 |
| assessment | Projects / Lab. | 1 | 10% (10) | Continuous | All |
| | Seminars | 1 | 10% (10) | 13 | LO # 6 – 11 |
| Summative assessment | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - #7 |
| | Final Exam | 3hr | 50% (50) | 16 | All |
| Total assessment 100% (100) | | | | | |

| Delivery Plan (Weekly Syllabus) | | | |
|---------------------------------|---|--|--|
| | المنهاج الاسبوعي النظري | | |
| | Material Covered | | |
| Week 1 | The Nature of Analytical Chemistry | | |
| Week 2 | Calculations Used in Analytical Chemistry | | |
| Week 3 | Sampling, Standardization, and calibration | | |
| Week 4 | Aqueous Solutions and Chemical Equilibria | | |
| Week 5 | Effect of Electrolytes on Chemical Equilibria | | |
| Week 6 | Solving Equilibrium Problems for Complex Systems | | |
| Week 7 | Mid-term Exam - Gravimetric Methods of Analysis | | |
| Week 8 | Titrations in Analytical Chemistry | | |
| Week 9 | Principles of Neutralization Titrations, Solutions and Indicators for Acid/Base titrations | | |
| Week 10 | Titration Curves for Weak Acids , Determining Dissociation Constants of Weak Acids and Bases. | | |
| Week 11 | Complex Acid/Base Systems | | |
| Week 12 | Calculation of the pH of Solutions of NaHA , Titration Curves for Polyfunctional Acids | | |

| Week 13 | Principles of Neutralization Titrations |
|---------|--|
| Week 14 | Nernest equation and measurement of concentration by potential of the cell |
| Week 15 | Selected Inorganic materials and reactions |
| Week 16 | Final Exam |

| Delivery Pl | an (Weekly Lab. Syllabus) |
|-------------|--|
| | المنهاج الاسبوعي للمختبر |
| | Material Covered |
| Week 1 | Lab 1: Preparation of primary standard solution (preparation of pure sodium carbonate solution) |
| Week 2 | Lab 2: Preparation of secondary standard solution of hydrochloric acid |
| Week 3 | Lab 3: Titrate a dilute hydrochloric acid solution and find the normality of sodium hydroxide |
| Week 4 | Lab 4: Quantitative determination of the components of a solution composed of a mixture of sodium carbonate and sodium hydroxide |
| Week 5 | Lab 5: Titrations curves |
| Week 6 | Lab 6: Refracting titrant |
| Week 7 | Lab 7: Quantitative determination of chlorine ions in drinking water |
| Week 8 | Lab 8: Titration of potassium permanganate solution and finding the normality of ferrous sulfate solution |
| Week 9 | Lab 9: Water hardness measurement |
| Week10 | Lab 10: The acidity of vinegar |
| Week11 | Lab 11: Purification of table salt |
| Week12 | Lab 12: Determination of the amount of ammonia in ammonium salts |
| Week13 | Lab 13: Paper chromatography |
| Week14 | Lab 14: detection of ions |
| Week15 | Lab 15: Detection of elements by organic compounds |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | | |
|--|---|------------------------------|--|--|--|
| | Text | Available in the Library? | | | |
| Required Texts | Skoog, D.A., West, D.M., Holler, F.J. and Crouch, S.R., 2013. Fundamentals of analytical chemistry. Cengage learning. | Yes | | | |
| Recommended Texts | David H., 2000. Modern Analytical Chemistry. 1st ed. | Yes | | | |





MODULE DESCRIPTOR وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | | | | | |
|---|----------------------|----------------------------------|-------------------------|-------|------------------------|-------------|--|--|--|--|
| Module Title | Material B | Material Balance Module Delivery | | | | | | | | |
| Module Type | Core | Core | | | | | | | | |
| Module Code | CHE 105 | | | | Lecture Tutorial | | | | | |
| ECTS Credits | 7 | | | | Tutoriai | | | | | |
| SWL (hr/sem) | 78 | 78 | | | | | | | | |
| Module Level | ıle Level 1 | | | | Semester (s) offered 2 | | | | | |
| Administering Department | Chemical Engineering | | College | Engin | eering | | | | | |
| Module Leader | Dr. Mohan | nmed H. Msaed | e-mail | Mhm | msaed1@uodiy | yala.edu.iq | | | | |
| Module Leader's Acad. Title | | Lecturer | Module Lo Qualificat | | i | Ph.D. | | | | |
| Module Tutor | None | | e-mail | | | | | | | |
| Peer Reviewer Name None | | None | e-mail | | | | | | | |
| Review Commit Approval | ttee | | Version N | umber | | | | | | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | |
|--|--|-------------------|------------|--|--|
| Prerequisite module | Principles of Chemical Engineering | Semester | 1 | | |
| Co-requisites module | None | Semester | | | |
| Module Aims, Lea | arning Outcomes, Indicative Contents and | d Brief Descr | iption | | |
| ختصر | ادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف ه | أهداف الم | | | |
| Module Aims | Learn the basic concepts of material balances in cher | nical engineering | processes. | | |
| أهداف المادة الدراسية | It also comprehensively introduces problem-solving | methods with a | nd without | | |
| | chemical reactions. | | | | |
| Module Learning | Knowledge and Understanding | | | | |
| Outcomes | Having successfully completed this module, the | student will b | e able to | | |
| | demonstrate knowledge and understanding of: | | | | |
| مخرجات التعلم للمادة الدراسية | Definition of Chemical Reaction Equation and | d Stoichiometry. | | | |

| | Determination the Limiting and Excess Reactants, Conversion and Selectivity. |
|----------------------------|---|
| | Solving the Material Balances for Processes Involving Chemical Reaction. |
| | Describing the Processes Involving Multiple Reactions. |
| | Explanation the Element Material Balances. |
| | Solving the Material Balances Involving Combustion. |
| | Showing the Material Balance Problems Involving Multiple Units. |
| | Describing the Recycle with and without Chemical Reaction. |
| | Describing the Bypass, Purge, and the Industrial Application of Material Balances. |
| | Indicative content includes the following. |
| | The Chemical Reaction Equation and Stoichiometry (6 hrs). |
| | Limiting and Excess Reactants, Conversion, Selectivity and Yield (12 hrs). |
| | Material Balances for Processes Involving Chemical Reaction (12 hrs). |
| Indicative Contents | Processes Involving Multiple Reactions (3 hrs). |
| المحتويات الإرشادية | Element Material Balances (6 hrs). |
| | Material Balances Involving Combustion (6 hrs). |
| | Material Balance Problems Involving Multiple Unit (12 hrs)s. |
| | Recycle with and without Chemical Reaction (12 hrs). |
| | Bypass, Purge, and the Industrial Application of Material Balances (9 hrs). |
| | This This module is designed to provide first-year Chemical Engineering students |
| | with the foundational concepts of material balances, including the chemical |
| | reaction equation and stoichiometry, limiting and excess reactants, conversion, |
| Course Description | selectivity and yield, material balances for processes involving chemical reaction, processes involving multiple reactions, element material balances, material |
| | balances involving combustion, material balance problems involving multiple |
| | units, recycle with and without chemical reaction, bypass, purge, and the |
| | industrial application of material balances. |
| | Learning and Teaching Strategies |
| | استراتيجيات التعلم والتعليم |
| | Beginning to establish a strong understanding of the basics of material balance |
| | processes. Using real industrial examples to help students relate the basic |
| | concepts of material balance to real industrial processes. Encouraging the |
| Strategies | discussions and questions to clear up any misconceptions. In addition, |
| | encouraging active participation and group discussions to enhance critical |
| | thinking and problem-solving skills. Guiding students through the problem- |
| | solving process and providing constructive feedback. |
| | |

| Student Workload (SWL) الحمل الدر اسي للطالب | | | | |
|---|--------------------|-----|--|-----|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures In class tests Tutorial Final Exam | 57 3 15 3 | 78 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 5 |
| | 10 10 | 97 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 6.5 |
| Total SWL (h/sem) | | 175 | | |

Module Evaluation

تقييم المادة الدراسية

| | | Time/ | Weight (Marks) | Week Due | Relevant Learning | |
|----------------------|--------------|-----------------|----------------|-------------------|--------------------|--|
| | | Number | weight (Marks) | week Due | Outcome | |
| F | Quizzes | 6 | 20% (20) | 2,4,6,9,11 and 13 | LO #1, 2, 6, and 7 | |
| Formative assessment | Assignments | 6 | 10% (10) | 3,5,7,9,11 and 14 | LO # 1,2,3,5 and 6 | |
| assessment | Home work | 1 | 10% (10) | 14 | LO # 4-9 | |
| Summative | Midterm Exam | 1 | 10% (10) | 7 | LO # 1-5 | |
| assessment | Final Exam | 1 | 50% (50) | 16 | All | |
| Total assessment | | 100% (100Marks) | | | | |

| | Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | |
|---------|--|--|--|
| | Material Covered | | |
| Week 1 | The Chemical Reaction Equation and Stoichiometry | | |
| Week 2 | Terminology for Applications of Stoichiometry | | |
| Week 3 | Limiting and Excess Reactants | | |
| Week 4 | Conversion and degree of completion | | |
| Week 5 | Selectivity and Yield | | |
| Week 6 | Material Balances for Processes Involving Chemical Reaction | | |
| Week 7 | Material Balances for Processes Involving Chemical Reaction | | |
| Week 8 | Processes Involving Multiple Reactions | | |
| Week 9 | Element Material Balances | | |
| Week 10 | Material Balances Involving Combustion | | |
| Week 11 | Material Balance Problems Involving Multiple Units | | |
| Week 12 | Material Balance Problems Involving Multiple Units | | |
| Week 13 | Recycle with and without Chemical Reaction | | |
| Week 14 | Bypass, Purge, and the Industrial Application of Material Balances | | |
| Week 15 | Bypass, Purge, and the Industrial Application of Material Balances | | |
| Week 16 | Final Exam | | |

| Learning and Teaching Resources | | | | | |
|---------------------------------|--|-----|--|--|--|
| | مصادر التعلم والتدريس | | | | |
| | Text Available in the Library? | | | | |
| Required Texts | David M Himmelblau, Basic principles and calculations in chemical engineering, Prentice Hall. | Yes | | | |
| Recommended Texts | Richard M Felder & Ronald W. Rousseau Elementary Principles of Chemical Processes, Wiley India. | No | | | |





MODULE DESCRIPTOR وصف المادة الدر اسية

| Module Information معلومات المادة الدراسية | | | | | | |
|---|---------------|----------------------|-------------------------|--------------|-------------------|--------|
| Module Title | Engineeri | ng Drawing | | Mo | Module Delivery | |
| Module Type | Basic | | | | Theory | |
| Module Code | CHE 106 | | | | Theory Lecture | |
| ECTS Credits | 5 | | | | Lab Practica | 1 |
| SWL (hr/sem) | 100 | 100 | | | | |
| Module Level | | 1 | Semester | (s) offer |) offered 2 | |
| Administering Department | | Chemical Engineering | College | Engine | ering | |
| Module Leader | Yaser I. Jase | em | e-mail | <u>Yaser</u> | ij@uodiyala.e | edu.iq |
| Module Leader's Acad. Title | | Assist. Proff. | Module Lo Qualificat | | | MS.c. |
| Module Tutor | | | e-mail | | | |
| Peer Reviewer Name | | | e-mail | | | |
| Review Committee Approval | | | Version N | umber | 1.0 | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | |
|---|--|----------|---|--|
| Prerequisite module | None | Semester | - | |
| Co-requisites module | None | Semester | - | |
| Module Aims, Learning Outcomes, Indicative Contents and Brief Description | | | | |
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | |
| Module Aims أهداف المادة الدر اسية | Engineering drawing is the principal method of communication for engineers, the objective is to introduce the students, to the techniques of constructing the various types of polygons, curves and scales. In addition to engineering drawing, students become familiar with the AutoCAD user interface. Understand the fundamental concepts and features of AutoCAD. Use the precision drafting tools in AutoCAD to develop accurate technical drawings—present | | | |

| | drawings in a detailed and visually impressive manner. |
|-------------------------------|--|
| | At the end of this course the students will be able to: |
| | 1- Get information about the important tools for engineering drawing. This will give |
| | |
| | student basic knowledge of technical drawings professions and means of communications to others. |
| | |
| | 2- Learning how to draw the shapes, angels and lines and others which is essential for |
| | engineer |
| | 3- Develop student's imagination and ability to represent the shape size and |
| | specifications of physical objects. |
| Module Learning | 4- Understand the main idea of using dimension for engineering drawing |
| Outcomes | 5- Familiarize with different drawing equipment, technical standards and procedures |
| | for construction of geometric figures. This will give students ability to draw three- |
| مخرجات التعلم للمادة الدراسية | dimension objects on the paper and to draw the pectoral drawings. |
| | 6- Explain the principle of projection and sectioning |
| | 7- Utilize the power and precision of AutoCAD as a drafting and design tool used in |
| | the mechanical design and manufacturing industries. |
| | 8- Apply basic CAD concepts to develop and construct accurate 2D geometry through |
| | creation of basic geometric constructions. |
| | 9- Create, manipulate and edit 2D drawings and figures. |
| | 10- Apply elements of mechanical drafting such as layers, dimensions, drawing |
| | formats, and 2D figures in projects. |
| | Indicative content includes the following. |
| | Paper size, Lettering & title blocks |
| | Orthographic projection |
| | Isometric and oblique projection |
| | Perspective drawing |
| | Basic geometrical solids |
| Indicative Contents | Development of surfaces Creating Basic Drawings |
| المحتويات الإرشادية | Creating Basic Drawings Manipulating Objects |
| | Implementing Drawing Organization and Inquiry Commands |
| | Altering Objects |
| | Annotate a Drawing |
| | Dimension Drawings |
| | Hatching Objects |
| | Creating Additional Drawing Objects and working on Projects |
| | Plotting the Drawing Output |
| | This course introduces students to the introduction to the drawing tools and how to use |
| | them, lines drawing, Basic engineering processes, Composition of the engineering |
| | drawings, Letters, numbers, dimensions, Projection, Sectioning, Isometric drawing. |
| Course Description | Understand the fundamental concepts and features of AutoCAD. Use the precision |
| Course Description | drafting tools in AutoCAD to develop accurate technical drawings. Present drawings in |
| | a detailed and visually impressive manner. Develop a level of comfort and confidence |
| | with AutoCAD through hands-on experience. |
| | |
| | Learning and Teaching Stratogics |
| | Learning and Teaching Strategies |
| | استراتيجيات التعلم والتعليم |
| | Begin to establish a strong conceptual understanding of the principles of engineering |
| Strategies | drawing. Use the reality of work examples and measurements to help students relate |
| - | abstract concepts to the planning. Encourage discussions and questions to clear up any |
| | misconceptions. In addition, provide students with the tool to deal with schematic |

problems. Encourage active participation and group discussions to enhance critical thinking and problem-solving skills. Guide students through the problem-solving process and provide constructive feedback.

| Stu | Student Workload (SWL) الحمل الدراسي للطالب | | | |
|---|--|---|-----|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures 26 Lab 45 Practical 15 In class tests 4 Final Exam 3 | 93 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبو عيا | 6 | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Library, dorm, home memorizing 6 Preparation for tests 8 Homework 18 | 32 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2,3 | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | 125 | | | |

| | Module Evaluation تقييم المادة الدر اسية | | | | | |
|----------------------|---|--------------|------------------|------------|---|--|
| | | Time (hr) | Weight (Marks) | Week Due | Relevant Learning Outcome | |
| Formative | Quizzes | 2 | 15% (15) | 5, 12 | LO #1, 2, 3,7,8 and LO# 4,5,6,9,10 | |
| assessment | Assignments | 6 | 5% (5) | 7, 12 | LO # 7, 8 and LO# 9,10 | |
| | Homework | 2 | 20% (20) | Continuous | | |
| Summative assessment | Midterm Exam | 2 | 10% (10) | 7 | LO #1, 2, 3,4,7,8 and LO# 4,5,6,9,10 | |
| assessificit | Final Exam | 3 | 50% (50) | 16 | All | |
| Total assessment | | | 100% (100 Marks) | | | |

| Delivery Plan (Weekly Syllabus) المنهاج الاسبو عي النظري | | | |
|---|--|--|--|
| | Material Covered | | |
| Week 1 | Definition of tools and how to use them, Introduction Status Bar and Command Prompt AutoCAD Commands Dynamic Input Menus, Ribbons, and Toolbars Cursor and Colors Undo and Redo | | |

| | - Initial principles of drawing, |
|---|--|
| | - Drawing Aids |
| Week 2 | 1. Open Existing Drawings |
| | 2. Creating a New Drawing 3. Saving Drawings |
| | 4. Exiting AutoCAD |
| | - Letters and numbers, |
| | 5. SNAP Command |
| Week 3 | 6. Grid Command |
| week 3 | 7. Running Object Snaps |
| | 8. Osnap Settings |
| | 9. UNITS Command |
| | - Dimensions, |
| | - Draw Commands |
| Week 4 | 1. Line Command |
| ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,, | 2. Cartesian Coordinate System |
| | 3. Orthogonal Lines |
| | 4. Polar Tracking |
| | - Line drawing, |
| *** | - 5. Circles |
| Week 5 | 6. Arc Command |
| | 7. Polyline Command |
| | 8. Explode Command |
| Week 6 | - Line drawing, |
| week o | 9. Rectangle |
| | 10. Ellipse |
| | - Engineering operations, |
| | - Edit Commands |
| Week 7 | 1. The Move Command |
| VVCCII 7 | 2. The Copy Command 3. The Offset Command |
| | 4. The Extend Command |
| | 5. Trim Command |
| | - Engineering operations, |
| | 6. The Erase Command |
| | 7. The Zoom Command |
| Week 8 | 8. The Pan Command |
| | 9. The Mirror Command |
| | 10. The Rotate Command |
| | 11. The Scale Command |
| | - Projection drawing, |
| Week 9 | 12. The Break Command |
| ,, con | 13. The Stretch Command |
| | 14. The Explode Command |
| | - Projection drawing, |
| | 15. The Fillet Command |
| Week 10 | 16. The Chamfer Command |
| | 17. The Array Command |
| | 18. The Lengthen Command |
| XX71- 11 | - Projection drawing, |
| Week 11 | - Dimensions |
| | 1. Linear Dimensions |
| Week 12 | - Drawing of sectional Views, |
| VVCCK 12 | 2. Aligned Dimensions |
| | 3. Radial Dimensions |
| Week 13 | - Drawing of sectional Views, |
| | 4. Angular Dimensions |

| | 5. Continued and Baseline Dimensions |
|---|--------------------------------------|
| Week 14 | - Isometric, |
| ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,, | 6. Modifying Dimensions |
| | - Isometric, |
| Week 15 7. Dimension Styles | |
| | * Creating |
| Week 16 | Final Exam |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|--|---------------------------|--|--|
| | Text | Available in the Library? | | |
| Required Texts | Parkinson, A.C., 1961. A First Year Engineering Drawing J Luzadder, W., 1965. Fundamentals of Engineering Drawing, by Warren J. Luzadder. Prentice-hall. Text book 1: James A. Leach, "AutoCad 2002 companion", 2003. Text book 2: AutoCAD 2D Tutorials, AutoCAD 2013, By Kristen S. Kurland, 2012. | Yes | | |
| Recommended Texts | 3. Text book 3: 2D_AutoCAD. | Yes | | |
| Websites | | | | |





MODULE DESCRIPTOR

وصف المادة الدراسية

Module Information معلومات المادة الدر اسبة **English language Module Title Module Delivery Module Type Suplement Module Code** U 104 **Theory** Lecture **ECTS Credits** 2 Seminar **50** SWL (hr/sem) 2 **Module Level** 1 Semester (s) offered **Administering Department Chemical Engineering College** Engineering **Module Leader** Essa9781@uodiyala.edu.iq Ali Ismaael e-mail Module Leader's Module Leader's Acad. Title Assistant Lecturer Master Qualification **Module Tutor** e-mail **Peer Reviewer Name** e-mail **Review Committee Version Number** 1.0 **Approval**

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | |
|--|--|----------|---|--|--|
| Prerequisite module Semester - | | | - | | |
| Co-requisites module | None | Semester | - | | |
| Module Aims, Lea | Module Aims, Learning Outcomes, Indicative Contents and Brief Description | | | | |
| ، مختصر | المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف | أهداف ا | | | |
| Module Aims أهداف المادة الدراسية | The module aims to develop the students' English skills in reading, writing, listening and speaking. | | | | |
| Module Learning | Module Learning 15. Read and understand simple texts in English. | | | | |
| Outcomes | 16. Answer simple comprehension questions and match sentences about | | | | |
| | texts. | | | | |
| مخرجات التعلم للمادة الدراسية | 17. Reconstruct texts by reordering sentences. | | | | |

| | 18. Understand the main idea of a text. | | | | |
|---------------------|--|--|--|--|--|
| | | | | | |
| | 19. Identify specific information in a text. | | | | |
| | 20. Writing and paraphrasing paragraphs. | | | | |
| | Indicative content includes the following. | | | | |
| | i) Grammar has a core place in language teaching and learning. | | | | |
| Indicative Contents | ii) A wide variety of practice tasks in all the four skills are essential to | | | | |
| المحتويات الإرشادية | language learning. | | | | |
| | iii) Everyday expressions, particularly of spoken English, also need a place in | | | | |
| | the syllabus. These can be functional, social, situational or idiomatic | | | | |
| | Each unit is organized to enhance students' basic knowledge of vocabulary and | | | | |
| | grammar through reading texts. The students will learn how to form simple | | | | |
| Course Description | sentences and use them in real life situations as well as in writing different | | | | |
| | assignments. By the end of the course, students will be able to produce basic | | | | |
| | sentences and communicate in simple real-life situations. | | | | |
| | Learning and Teaching Strategies | | | | |
| | استراتيجيات التعلم والتعليم | | | | |
| | Headway's trusted methodology combines solid grammar and practice, vocabulary | | | | |
| | development, and integrated skills with communicative role-plays and | | | | |
| | personalization. | | | | |
| | Authentic material from a variety of sources enables students to see new language | | | | |
| Strategies | in context, and a range of comprehension tasks, language and vocabulary | | | | |
| Strategies | exercises, and extension activities practice the four skills. 'Everyday English' and | | | | |
| | 'Spoken grammar' sections practice real-world speaking skills, and a writing | | | | |
| | | | | | |
| | section for each unit at the back of the book provides models for students to | | | | |
| | analyze and imitate. | | | | |

| Student Workload (SWL) | | | | |
|--|------------|------------|--|-----|
| | (| اسي للطالب | الحمل الدر | |
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | | | | |
| In class lectures | 2 6 | 33 | Structured SWL (h/w) | 2 |
| In class tests | 2 | | الحمل الدراسي المنتظم للطالب أسبوعيا | 2 |
| Seminars | 2 | | | |
| Final Exam | 3 | | | |
| Unstructured SWL (h/sem) | | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | | Unstructured SWL (h/w) | |
| Library, dorm, home memorizing | 5 | 17 | الحمل الدراسي غير المنتظم للطالب أسبوعيا | 1.1 |
| Preparation for tests | 8 | | العمل الدراسي غير المستعم سعاب اللبوعيا | |
| Homework | 4 | | | |
| Total SWL (h/sem) | | | | |
| الحمل الدراسي الكلي للطالب خلال الفصل | | | | |

| Module Evaluation | | | | | | | |
|----------------------|-----------------------|------|----------------|--------------------|--------------------------|--|--|
| | تقييم المادة الدراسية | | | | | | |
| | | Time | Weight (Marks) | Week Due | Relevant Learning | | |
| | | (hr) | , , | | Outcome | | |
| T 4* | Quizzes | 2 | 10% (10) | 5, 10, 12, 14 | LO #1, 2, 3, and 4 | | |
| Formative assessment | Assignments | 6 | 20% (20) | 2, 4, 6, 8, 10, 12 | LO # 1, 2, 3, 4, 5 and 6 | | |
| assessment | Seminars | 2 | 10% (10) | Continuous | | | |

| Summative | Midterm Exam | 2 hr | 10% (10) | 7 | LO # 1-3 |
|---------------|--------------|------|------------------|----|----------|
| assessment | Final Exam | 3 hr | 50% (50) | 16 | All |
| Total assessm | ent | | 100% (100 Marks) | | |

| Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | | |
|--|--|--|--|
| | | | |
| | Material Covered | | |
| Week 1 | GRAMMAR, READING, MAIN COURSE SPEAKING, LISTENING, VOCABULARY am/is/are my/your This is Introduction dialogues, Everyday English dialogues Introductions, Good morning! Practicing introduction dialogues. People meet each other and introduce someone else. How are you? What's this in English? Numbers 1-10 and plurals. | | |
| Week 2 | He/she/they His/her. Questions Where are they from?, Two people are on holiday in New York. Students ask and answer questions about where people are from. Countries, Numbers 10-20 A set of cities and countries: Brazil, Spain Adjectives: awful, really good, fantastic, beautiful Nouns: centre, hospital, building, park | | |
| Week 3 | Verb to be is recycled and extended to include negative and question forms. We're in Las Vegas! Roleplay: in a band. An interview with the band Metro 5. Jobs: a nurse, a doctor Personal information: surname, first name, address, married Social expressions: I'm sorry, thanks, please | | |
| Week 4 | Possessive adjectives. Possessive 's. Has/ have Adjective + noun Irregular Plurals Paddy McNab and his family, My best friend. The alphabet, On the phone, Saying email addresses. Who are they? Listen and identify the people. The family: mother, son Describing a friend: very beautiful, really funny | | |
| Week 5 | Present Simple: I/you/we/they a/an Adjective + noun Colin Brodie from Dundee. Role play: At a party. Where is Colin? Who is he with? At a party: Flavia and Terry are at a party in London. The lexical set of sports/food/drinks. Languages and nationalities. | | |
| Week 6 | Present Simple: He/she Question and negatives Adverbs of frequency Prepositions of time Lois Maddox Talking about daily routines, Asking and answering questions about daily routines, Lifestyle questionnaire Listening a phone conversation between Lois and Elliot. Days of the week. The time. Words that go together: watch TV, get up early | | |
| Week 7 | Question words Subject pronouns Object pronouns Possessive pronouns This and that A postcard from San Francisco, A holiday postcard. Describing lifestyles, preferences and places, Roleplay: conversations in town. Listening the requests with Can I? Adjectives: lovely, terrible, comfortable, friendly Opposite adjectives: new/old, big/small Places: chemist, post office | | |
| Week 8 | There is /are Prepositions: in, on, under, next to Vancouver-the best city in the world, What to do and where to go. Talking and asking about rooms and furniture, Giving directions. My home town, Steve talks about living in Vancouver. Rooms and furniture: living room, bedroom In and out of town: beach, mountain, sailing, | | |
| Week 9 | Was/were born Past simple: irregular verbs It's a Jackson Pollock. Telling a story from pictures, Saying the dates in English. Magalie Dromand, Magalie dromand talks about her family. Saying years People and jobs Irregular verbs Have, do, go: have lunch, do homework, go shopping | | |
| Week 10 | Past simple: regular and irregular Questions Negatives Ago Dialogues with simple past. Did you have a good weekend? Asking about holidays, A questionnaire, My last holiday, Roleplay: asking and giving directions. Angie and Rick are at work, Jack and Millie's holiday. Weekend activities: go to the cinema, have a meal Time expressions: on Monday, last night Sports and leisure: tennis, skiing, windsurfing Play or go: play tennis, go skiing Seasons: winter, summer | | |
| Week 11 | Can / can't, Adverbs, Adjective + noun Requests and offers The Internet, What can you do on the internet? Talking about what you can do, Talking about everyday problems, Five people talk about what they do on the internet. Verbs: draw, run, drive Verb+noun: Listen to the radio, chat to friends Adjective+noun: fast car, busy city, dangerous sport Opposite adjectives: dangerous/ safe, old/modern, old/young. | | |
| Week 12 | I'd like, You are what you eat, Discussion-what is a good diet? Conversation with Adam, Shopping: bread, milk, fruit, Please and thank you Some /any, Like and would like People from different parts of | | |

| | | the world describe what they eat. Roleplay: Ordering a meal. Birthday wishes, What people want on |
|-----|--------|---|
| | | their birthday. stamps, cheese, ham Food: cereal, salad, pasta, fish In a restaurant: menu, starter, |
| | | desert, soup, salmon |
| | | Present continuous, Present simple and present continuous. This week is different, Colin, a |
| W | eek 13 | millionaire, gives money to homeless teenagers What's the matter? Why don't you? What is Nigel |
| • • | | wearing? Nigel is on holiday, What's the matter. Colours: blue, red, green Clothes: jacket, trousers, |
| | | shoes and socks Opposite verbs: buy/sell, love/hate, open/close |
| | | Future plans, Revision: question words, tenses. Seven countries in seven days, Life's big events: three |
| w | eek 14 | people talk about their family, education, work and ambitions. A mini autobiography. Eddie is talking |
| | | to a friend about his holiday plans, social expressions Transport: travel by bus, coach, motorbike, |
| | | plane Revision |
| W | eek 15 | Irregular verbs, phonetic symbols, consonants and vowels. |
| | | |
| | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | | |
|--|---|-----|--|--|--|
| Text Available in t | | | | | |
| Required Texts | New Headway Beginner, by lizand john soars | Yes | | | |
| | https://www.learnenglish.de/ https://www.englishgrammar.org/ https://www.phrasebank.manchester.ac.uk/ | | | | |
| Websites | | | | | |





MODULE DESCRIPTION FORM

نموذج وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | | | |
|---|--------------|-------------------------|---|---------|---------------------|--------------|--|
| Module Title | Engineering | Engineering Workshops | | | Module Delivery | | |
| Module Type | BASIC | | | | | | |
| Module Code | CHE 107 | | | | Theory Practical | | |
| ECTS Credits | 4 | | | | Practical | | |
| SWL (hr/sem) | 100 | | | | | | |
| Module Level | | 1 | Semester of Delivery 2 | | 2 | | |
| Administering De | epartment | Chemical Engineering | College | Engine | eering | | |
| Module Leader | Yasseen Alwa | n Jaddoa | e-mail | yasseen | alwan_eng@uod | iyala.edu.iq | |
| Module Leader's | Acad. Title | Assistant Lecturer | Module Leader's Qualification MSc. | | MSc. | | |
| Module Tutor | | e-mail | | | | | |
| Peer Reviewer Name | | e-mail | | | | | |
| Scientific Committee Approval Date | | | Version Nu | mber | 1.0 | | |

| Relation with other Modules | | | | | |
|-----------------------------|-----------------------------------|----------|--|--|--|
| | العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | None | Semester | | | |
| Co-requisites module | None | Semester | | | |

| Module | Module Aims, Learning Outcomes and Indicative Contents | | | | | |
|---|---|--|--|--|--|--|
| | أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | | |
| Module Objectives أهداف المادة الدر اسية | 1- Y الآهد اف المعرفية -1 الفهام وتعليم الطالب مفاهيم ومبادى مادة الورش الهندسية العامة . -2 تمكين الطلبة من الحصول على المعرفة والفهم | | | | | |

| | الكامل لكافة مهارات الورش الهندسية . 5- افهام الطالب وتعريفه بكافة الممهارات نظريا وعمليا وتعريفه بكافة أجزاء الماكنة التي يتم تطبيق المهارة با لإضافة الى التطبيق العملي على الماكنة لكل طالب ولكافة المهارات. 4- تمكين الطلبة من الحصول على المعرفة والفهم لكل أجزاء الماكنة وفائدة كل جزء . 5- تمكين الطلبة من الحصول على المعرفة والفهم على تشخيص انواع الاعمال التي تنجزها كل ماكنة وطريقة العمل عليها . |
|--|---|
| | ب - الأهداف المهاراتية الخاصة بالبرنامج 1 - شرح المهارات بالتفصيل وتطبيقها على الماكنة عمليا و التاكيد على الطلبة بضرورة الالتزام بقواعد السلامة المهنية. 2 - تزود هم بمعلومات وطرق حل المشاكل العملية |
| | المتعلقة بجميع المهارات. 3 - يتم عرض مواضيع كافة المهارات نظريا. 4 - يتم التركيز على العملي في المهارات وضرورة مشاركة الطالب في العملي. مخرجات التعلم للمادة الدراسية |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | 1- بناء الطالب علميا وعمليا وتأهيله للعمل في مجال تقنيات الهندسة. 2- بناء وإعداد الطالب نفسيا ليقوم بدوره كمهندس يعتمد عليه في هذا المجال. 3- بناء طلبة قادرين على التنافس مع مهندسين اخرين لفرس العمل و الحصول على المقاعد المطلوبة في اكمال دراسات عليا. 4- قابلية التقديم لاختبارات خارجية من قبل هيئات محلية أو أقليمية أو عالمية لغرض اكمال الدراسة او التعيين. 5- حث الطالب على الإبداع والتفكير في مشاريع التخصص ومو اكبة التطور الحاصل في هذا المجال. 6- تزويد الطلبة بمهارات علمية وعملية ومهارات ذاتية تمكنه من حل المشاكل العملية والتعامل معها بمفاهيم علمية. |
| Indicative Contents المحتويات الإرشادية | المحتوبات الإرشادية اولا-: السلامة المهنية. المحتويف الطالب بقواعد وإجراءات السلامة المهنية لجميع الممهارات وجميع الأقسام لاجل سلامة المستخدم من مخاطر المتعامل مع هذه الأدوات والمكائن (3 ساعات). التعامل مع هذه الأدوات والمكائن (3 ساعات). شرح نظري لمهارة القياسات. والأدوات المستخدمة في عملية القياس بالتفصيل ومن شم يتم التطبيق العملي لمهارة القياسات حيث يقوم كل طالب باجراء عملية القياس لمختلف الأدوات إضافة الى تسليم باجراء عملية اللهامة بالمهارة (6ساعات) شرح نظري لمهارة اللحام وتعريف الطالب بجميع العدد شرح نظري لمهارة اللحام وتعريف الطالب بجميع العدد والأدوات المستخدمة في عملية اللحام وشرع طرق اللحام بالمتفصيل إضافة الى التطبيق اللحام وشعريف الطالب بجميع العدد |

بعملية اللحام وذلك لاكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6ساعات).

رابعا:- مهارة البرادة .

شرح نظري لمهارة البرادة وتعريف الطالب بجميع العدد والأدوات المستخدمة في عملية البرادة وشرع طرق البرادة بالتفصيل إضافة الى التطبيق العملى حيث يقوم كل طالب بعملية البرادة يدويا وذلك لاكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6سا عات).

خامسا:- مهارة السباكة

شرح نظري لمهارة السباكة وتعريف الطالب بجميع العدد و الأدوات المستخدمة في عملية السباكة وشرع طرق السباكة بالتفصيل إضافة الى التطبيق العملى حيث يقوم كل طالب بعملية السباكة يدويا وذلك لاكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6سا عات)

سادسا:- مهارة التفريز .

شرح نظري لمهارة التفريز وتعريف الطالب بجميع أجزاء ماكنة التفريزاضافة الى العدد والأدوات المستخدمة في عملية التفريز وشرع طرق التفريز بالتفصيل إضافة الى التطبيق العملى حيث يقوم كل طالب بالعمل على ماكنة التفريز وتنفيذ التمارين المختلفة على الماكنة وذلك من اجل اكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6ساعات).

سابعا:- مهارة التجليخ .

شرح نظري لمهارة التجليخ وتعريف الطالب بجميع أجزاء ماكنة التجليخ اضافة الى العدد والأدوات المستخدمة في عملية التجليخ وشرع طرق التفريز بالتفصيل إضافة الى التطبيق العملي حيث يقوم كل طالب بالعمل على ماكنة التفريز وتنفيذ التمارين المختلفة على الماكنة وذلك من اجل اكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6ساعات).

ثامنا:- مهارة الخراطة .

شرح نظري لمهارة الخراطة وتعريف الطالب بجميع أجزاء ماكنة الخراطة اضافة الى العدد والأدوات المستخدمة في عملية الخراطة وشرع طرق الخراطة بالتفصيل إضافة الى التطبيق العملي حيث يقوم كل طالب بالعمل على ماكنة الخراطة وتنفيذ التمارين المختلفة على الماكنة وذلك من اجل اكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6ساعات).

تاسعا:- مهارة التأسيسات الكهربانية .

شرح نظري لمهارة التأسيسات الكهربائية وتعريف الطالب بجميع العدد والأدوات المستخدمة في عمليات التأسيسات الكهربائية وشرع طرق التأسيسات الكهربائي إضافة الى التطبيق العملى حيث يقوم كل طالب بالعمل على ربط الدوائر الكهربائية المختلفة وذلك من اجل اكتساب الخبرة العملية إضافة الى تسليم ومناقشة التقارير الخاصة بالمهارة (6ساعات).

Course Description

The industrial safety, Tools of instrument (such as vernier caliper), Micrometer as tool of measurement. The general idea about abrasion and its type. The tools and materials that used in abrasion. General idea about some materials that can be used. In making files and their kinds. Milling. General concept and explaining the use of lathe. The ways of operating lathe. Turning tools and minerals used in making them. Drilling and gearing. The angles of tools that used in turnings. Maintenance of lathe. General idea about carpentry. Tools of measuring used in carpentry.

| Learning and Teaching Strategies | | | | |
|----------------------------------|---|--|--|--|
| | استر اتيجيات التعلم والتعليم | | | |
| Strategies | *امتحانات يومية باسئلة عملية وعلمية. *درجات مشاركة لاسئلة المنافسة الصعبة بين الطلاب . *وضع درجات للواجبات البيتية والتقارير المكلفة بهم . امتحانات فصلية للمنهج الدراسي اضافة الى امتحان نصف *السنة والامتحان النهائي. | | | |

| Student Workload (SWL) | | | | |
|---|------------|---|-----|--|
| ١ اسبوعا | محسوب له ٥ | الحمل الدر اسي للطالب | | |
| Structured SWL (h/sem) | | | | |
| الحمل الدراسي المنتظم للطالب خلال الفصل | | Structured SWL (h/w) | | |
| In class lectures 43 | 48 | الحمل الدر اسى المنتظم للطالب أسبوعيا | 3 | |
| In class tests 2 | | , y , g y | | |
| Final Exam 3 | | | | |
| Unstructured SWL (h/sem) | | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 3.5 | |
| Libraray, dorm, home, memorize 34 | 52 | | | |
| Preparation for tests 10 | | , , , , , , , , , , , , , , , , , , , | | |
| Homework 8 | | | | |
| Total SWL (h/sem) | | | | |
| الحمل الدراسي الكلي للطالب خلال الفصل | | 100 | | |

| Module Evaluation | | | | | |
|-----------------------|-----------------|-------------|-----------|-----------|----------------------|
| تقييم المادة الدراسية | | | | | |
| | | Time/Number | Weight | Week Due | Relevant Learning |
| | | | (Marks) | | Outcome |
| | Quizzes | 2 | 10% (10) | 5 and 10 | LO #1, #2 and #10, |
| Formative | Quizzes | 2 | 1070 (10) | 2 unu 10 | #11 |
| | Assignments | 2 | 10% (10) | 2 and 12 | LO #3, #4 and #6, #7 |
| assessment | Projects / Lab. | 1 | 10% (10) | Continuou | All |
| | Trojects / Lab. | 1 | 1070 (10) | S | 1111 |
| | Report | 1 | 10% (10) | 13 | LO #5, #8 and #10 |
| Summative | Midterm | 2hr | 10% (10) | 7 | LO #1 - #7 |

| assessment | Exam | | | | |
|------------------|------------|-----------|----------|----|-----|
| | Final Exam | 3hr | 50% (50) | 16 | All |
| Total assessment | | 100% (100 | | | |
| | | Marks) | | | |

| | Delivery Plan (Weekly Syllabus) |
|---------|--|
| | المنهاج الاسبوعي النظري |
| | Material Covered |
| Week 1 | شرح نظري وعملي لمبادئ السلامة المهنية وبيان خطورة عدم الالتزام |
| | بها على حياة الطالب. |
| Week 2 | شرح نظري لمهارة القياسات وتعريف جميع العدد والأدوات المستخدمة في القياس. |
| Week 3 | تطبيق عملي لمهارة القياسات. |
| Week 4 | شرح نظري لمهارة اللحام وتعريف جميع العدد والأدوات المستخدمة في عملية اللحام. |
| Week 5 | تطبيق عملي لمهارة اللحام. |
| Week 6 | شرح نظري لمهارة البرادة وتعريف جميع العدد والأدوات المستخدمة في عملية البرادة. |
| Week 7 | تطبيق عملي لمهارة البرادة. |
| Week 8 | شرح نظري لمهارة السباكة وتعريف جميع أجزاء فرن السباكة. |
| Week 9 | تطبيق عملي لمهارة السباكة . |
| Week 10 | شرح نظري لمهارة التفريز وتعريف الطالب بجميع أجزاء ماكنة |
| | التفريز. |
| Week 11 | تطبيق عملي لمهارة التفريز. |
| Week 12 | شرح نظري لمهارة التجليخ وتعريف الطالب بجميع أجزاء ماكنة التجليخ. |
| XX 1.42 | التجليح. |
| Week 13 | <u> </u> |
| Week 14 | شرح نظري لمهارة التاسيسات الكهربائية مع التطبيق العملي. |
| Week 15 | شرح نظري لمهارة الخراطة وتعريف الطالب بجميع أجزاء ماكنة الخراطة. |
| Week 16 | تطبيق عملي لمهارة الخراطة. |

| | Delivery Plan (Weekly Lab. Syllabus) المنهاج الاسبو عي للمختبر |
|--------|---|
| | Material Covered |
| Week 1 | التعريف بمبادئ السلامة المهنية. |
| Week 2 | شرح مهارة القياسات. |
| Week 3 | شرح مهارة اللحام. |
| Week 4 | شرح مهارة البرادة. |

| Week 5 | شرح مهارة السباكة |
|--------|-----------------------------|
| Week 6 | شرح مهارة التجليخ. |
| Week 7 | شرح مهارة الخراطة والتفريز. |

| | Learning and Teaching Resources مصادر التعلم والتدريس | | |
|----------------------|---|---------------------------|--|
| | Text | Available in the Library? | |
| Required Texts | MECH6014 - Mechanical Workshop Practice Tarafdar, J.C. and Raliya, R., "The Nanotechnology", Published by Scientific Publisher (SP), India, (2012). | Yes | |
| Recommended Texts | MECH6028 - Mechanical Workshop Practice 2 - CIT Modules | No | |
| Websites | https://www.coursera.org/browse/workshop-and-engineering/workshop - | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | | |
|---|-----------------|----------------------|------------|---------------------------|-------------------------|-------|--|
| Module Title | اللغة العربية 1 | | Modu | ule Delivery | | | |
| Module Type | Basic learning | g activities | | ⊠Theory | | | |
| Module Code | UD 12 | | | | □Lecture □Lab | | |
| ECTS Credits | 2 | | | | □Tutorial □Practical | | |
| SWL (hr/sem) | 50 | | | □Seminar | | | |
| Module Level | | UGI | Semester o | f Deliver | | | |
| Administering De | epartment | Chemical Engineering | College | | ng | | |
| Module Leader | Ali Esmaeel | | e-mail | | | | |
| Module Leader's | Acad. Title | Assistant Lecturer | Module Le | eader's Qualification Ph. | | Ph.D. | |
| Module Tutor | | | e-mail | E-mail | | | |
| Peer Reviewer Name | | Name | e-mail | E-mail | | | |
| Scientific Committee Approval Date | | 3/11/2024 | Version Nu | mber | 1.0 | | |

| | Relation with other Modules | | | |
|----------------------|-----------------------------------|----------|--|--|
| | العلاقة مع المواد الدراسية الأخرى | | | |
| Prerequisite module | None | Semester | | |
| Co-requisites module | None | Semester | | |

| M | odule Aims, Learning Outcomes and Indicative Contents | | | | | |
|--|---|--|--|--|--|--|
| *** | أهداف المادة الدر اسية ونتاتج التعلم والمحتويات الإرشادية | | | | | |
| Module Objectives أهداف المادة الدر اسية | الهدف الأساس في تدريس اللغة العربية للأقسام العلمية: اخذ الطالب الى روعة بيان القران الكريم ، وادراكه يقينا ان التعبير القراني تعبير فني مقصود كل لفظة ، وكل حرف وضع وضعا فنياً مقصوداً تقويم اللسان العربي . واعتماد العربية الفصحى في الحديث والكتابة ، ولزيادة رصيد الطالب من ادب وتراث ، على تناول النصوص المختارة من العصور الأدبية المختلفة ، لزيادة رصيد ادب لتراث والادب المعاصر . التأكيد على دور الطالب في المتابعة واثراء المعرفة باللغة العربية وفنونها بجهد خاص ، ذا ما وضعنا مفاتيح المنهاج الدراسي لتقع على الطالب بعد ذلك مهمة فتح الأبواب والنوافذ الى مصادر المعرفة الواسعة . في جعل العربية الفصيحة تحتل موضع الصدارة وتجاوز العامية ، خدمة الى لغتنا العربية المقدسة. وحفاظا على قوتها وجمالها . ومن نافلة القول في اهداف تدريس اللغة العربية : هي الجانب المحقق للوحدة وتلزمنا دوافع الوفاء بالحرص عليها والمحافظة على جوهرها . | | | | | |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | يمكن ايجاز مخرجات العلم لمادة اللغة العربية العامة لاقسام غير الاختصاص بالنسبة لمرحلة الدراسة الأولى بالاتي :- 1 - سيتمكن الطالب من التعرف على خصائص اللغة العربية كلغة سامية ، وفهم الخصائص المشتركة للغات السامية ، ومكانة اللغة العربية ضمن هذه العائلة اللغوية مما يعزز من ادراكه لاصولها وتطورها عبر التاريخ . 2 - تحليل أصوات اللغة العربية من حيث أماكن وطرق النطق ، والتمييز بين مختلف الأصوات العربية . 3 - معالجة القضايا الصرفية وتحليل بنية الكلمات العربية ، وفهم كيفية تكوينها وتغييرها لأداء معان مختلفة مما يطور مهاراته في تكوين وصياغة الكلمات بشكل صحيح . 4 - تحديد التراكيب النحوية في اللغة العربية واستخدامها بشكل سليم مما يعزز قدرته على بناء جمل صحيحة نحويا ومعبرة بوضوح . 5 - فهم العلاقات الدلالية مثل الترادف والتضاد والتضمين بين الكلمات مما يوسع من ادراكه لمعاني الكلمات ، وتوظيفها في سياقات مختلفة . 6 - اتباع قواعد الاملاء الصحيحة وتطبيقها مما يسهم في تحسين كتابته ويضمن وضوح المعنى ، ودقته . 7 - تحليل الأنواع الأدبية وتوظيف البلاغة في التعبير مما يعزز من فهمها للنصوص الأدبية ويطور مهارته في الكتابة بأسلوب مؤثر وبلاغي. | | | | | |
| Indicative Contents المحتويات الإرشادية | مقدمة حول اللغة العربية كجزء من اللغات السامية التعريف باللغة العربية كجزء من اللغات السامية التعريف باللغة العربية واصولها واهم خصائصها . مكانة اللغة العربية ضمن عائلة اللغات السامية وتاريخ تطورها . دراسة خصائص اللغة العربية المتأصلة في جذورها السامية . | | | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | |
|---|----|---|-----|
| Structured SWL (h/sem) 33 Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا الحمل الدراسي المنتظم للطالب خلال الفصل | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل | 17 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 1.1 |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | | |

| Module Evaluation | | | | | | |
|-----------------------|-----------------|-------------|------------------|------------|-------------------------------|--|
| تقييم المادة الدراسية | | | | | | |
| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome | |
| Formative assessment | Quizzes | 4 | 20% (5) | 3 , 6,8,11 | LO #1#3, #4#6, #7, #9 #11 | |
| | Assignments | 2 | 10% (5) | 2 and 12 | LO #3, #4 and #6, #7 | |
| assessificit | Projects / Lab. | | | | | |
| | Report | 1 | 10% (10) | 13 | LO #5, #8 and #10 | |
| Summative | Midterm Exam | 1hr | 10% (10) | 7 | LO #1 - #7 | |
| assessment | Final Exam | 2hr | 50% (50) | 15 | All | |
| Total assessm | ent | 1 | 100% (100 Marks) | | | |

| | Delivery Plan (Weekly Syllabus) |
|---------|--|
| | المنهاج الاسبوعي النظري |
| | Material Covered |
| Week 1 | (الفصل الأول) التعبير القرآني): نص قراني محدد من سورة الكهف (قصة موسى والخضر عليهم السلام) من الاية رقم "60" الى الاية رقم "82" . |
| Week 2 | المهارات اللغوية: (1) الحروف الشمسية و القمرية. |
| Week 3 | (3) كتابة حرفي الضاد و الظاء . |
| Week 4 | (4) كتابة التاء المربوطة و الطويلة . |
| Week 5 | (5) علامات الترقيم . |
| Week 6 | (الفصل الثاني) الأصوات و المعجم العربي: (6) الأصوات العربية: الأصوات الصامتة و الصانتة. |
| Week 7 | (7) أنواع المعاجم (معجمات الألفاظ و معجمات المعاني) و طريقة استخراج الألفاظ من المعجم. |
| Week 8 | (الفصل الثالث) القواعد النحوية : (8) أنواع الكلم (الاسم ، و الفعل ، و الحرف) . |
| Week 9 | (9) المفرد ، و المثنى ، و الجمع ، و علامات إعرابها . |
| Week 10 | (10) الجملة الفعلية و الجملة الاسمية. |
| Week 11 | (11) من الأدوات النحوية: أدوات الجر، أدوات نصب الفعل المضارع، أدوات جزم الفعل المضارع. |
| Week 12 | (الفصل الرابع) البلاغة و الأدب : (12) من الفنون البلاغية : التشبيه . |

| Week 13 | (13) نصوص من (الشعر الجاهلي = امرئ القيس) ، و (الإسلامي = حسان بن ثابت) . |
|---------|---|
| Week 14 | (14) نصوص من (الشعر العباسي = المتنبي)، و (الأندلسي = الموشحات). |
| Week 15 | (15) نصوص نثرية عربية قديمة (الخطب ، و الوصايا ، و المقامات) . |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | |
|--|---|---------------------------|--|
| | Text | Available in the Library? | |
| | 1 – التعبير القرآني - تأليف الدكتور فاضل السامرائي | | |
| | 2اللغة العربية العامة لأقسام غير الاختصاص / تأليف مجموعة من | | |
| Required Texts | أساتذة اللغة العربية. | | |
| | 3- شذا العرف في فن الصرف – تاليف الدكتور احمد الحملاوي | | |
| | 4- البلاغة الواضحة ١ – تاليف الدكتور احمد مطلوب | | |
| Recommended | | | |
| Texts | | | |
| Websites | | | |

SECOND YEAR (SEMESTER 3)





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|----------------|----------------------|------------|---------------------------|--------------------|---------|
| Module Title | Engineering | Mathematics I | | Modu | ıle Delivery | |
| Module Type | Core | | | | ■ Theory | |
| Module Code | CHE 201 | | | | ⊠ Lecture □ Lab | |
| ECTS Credits | 4 | | | | ☐ Practical | |
| SWL (hr/sem) | 100 | | | | ☐ Seminar | |
| Module Level | | UGII | Semester o | of Delivery Three | | Three |
| Administering De | partment | Chemical engineering | College | College of Engineering | | |
| Module Leader | Ali I. Abdalla | ı | e-mail | alialnua | immy@uodiyala | .edu.iq |
| Module Leader's Acad. Title | | Lecturer | Module Le | le Leader's Qualification | | MSc. |
| Module Tutor | | | e-mail | | | |
| Peer Reviewer Na | ıme | | e-mail | | | |
| Scientific Committee Approval Date | | | Version Nu | ımber | 1.0 | |

| Relation with other Modules | | | | |
|-----------------------------------|-------|----------|-----|--|
| العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | E 102 | Semester | Two | |
| Co-requisites module | None | Semester | | |

| Module Aims, Learning Outcomes and Indicative Contents أهداف المادة الدر اسية و نتائج التعلم و المحتويات الإرشادية | | | |
|---|---|--|--|
| Module Objectives أهداف المادة الدراسية | 1-Provide a Strong Foundation in Mathematical Concepts Introduce fundamentals of calculus, linear algebra, and other core mathematical tools Develop logical understanding and mathematical skills to solve engineering | | |

| | problems |
|---|---|
| | 2- Develop Problem-Solving Skills Equip students with techniques to interpret objectives, mathematize, solve, and communicate results Cultivate critical thinking and creativity in applying mathematics to engineering problems |
| | 3- Facilitate Active Learning Use a mix of lectures, tutorials, discussions, and hands-on exercises for active learning Provide guided practice with relevant engineering problems |
| | 4- Teach Applications to Engineering Fields Relate mathematical concepts to real-world engineering scenarios and applications Integrate engineering examples and case studies throughout the curriculum Demonstrate the importance and symbiosis between mathematics and |
| | engineering The overall goal is to develop students' mathematical maturity and ability to apply advanced concepts to solve complex engineering problems. This requires a combination of solid foundational knowledge, practice with relevant applications, and explicit connections between theory and practice. |
| Module Learning Outcomes مخرجات التعلم للمادة الدر اسية | Module Learning Outcomes Demonstrate an understanding the concept of translation of axes and apply this concept on conic sections. Understand functions of several variables, partial derivatives, total derivatives, and the chain rule Relative minima and maxima of functions of two variables, linearization, small change problems Polar coordinates system (Definition, polar equation, relating polar and Cartesian coordinates). Graph of functions and equations in polar coordinates system and find the area using such system. Understand vectors, vector fields, vector operations like dot and cross product, and find the equations of line and plane in the space Solve optimization problems using Lagrange multipliers method. |
| Course Description | Course Description Engineering Mathematics I is a foundational course that introduces students to the key mathematical concepts and techniques essential for success in various engineering disciplines. The course typically covers the following topics: 1- Differential Calculus • Basic differential calculus for functions of one and several variables • Maxima and minima for functions of several variables 2- Polar Coordinate System • Polar equations • Relating Polar and Cartesian coordinates • Graph of Polar equations • Area calculation using polar coordinate system 2-Linear Algebra • Vectors in 3-space |

- Vector equations of lines and planes
- Gradient vector and vector field
- Lagrange Multipliers

The course is typically delivered through a combination of lectures, tutorials, classwork, and homework. Assessment includes a final exam worth 50%, mid-term exam worth 10% and other coursework such as homework assignments, classwork and tests worth 40%.

Learning and Teaching Strategies

استراتيجيات التعلم والتعليم

- 1- Integrate Lectures and Tutorials for Active Learning
 - Mix direct instruction in lectures with guided practice in tutorials for active learning
 - Include a variety of learning activities like problem-solving, discussions, and hands-on exercises
- 2- Use Relevant Examples and Applications
 - Use examples and problems with engineering contexts embedded throughout the course
 - Include some engineering guest lectures to explain real-world applications of the mathematical concepts
 - Relate the mathematics to specific engineering disciplines like fluid mechanics, structures, etc
- 3- Establish High Expectations and Develop Critical Thinking
 - Establish high expectations for students' mathematical understanding
 - Develop critical thinking and problem-solving skills
 - Cultivate creativity in applying mathematics to engineering problems

The overall goal is to help students develop a strong conceptual understanding of the mathematics that is directly applicable to their engineering studies. This requires a combination of solid foundational knowledge, practice with relevant problems, and explicit connections to engineering applications. Both students and instructors play an active role in employing effective learning and teaching strategies to achieve this goal.

Strategies

| Student Workload (SWL) | | | |
|---|--------------|--|------|
| ۱ اسبوعا | ، محسوب لـ ٥ | الحمل الدراسي للطالب | |
| Structured SWL (h/sem) | | | |
| الحمل الدراسي المنتظم للطالب خلال الفصل | | | |
| In class lectures 41 | 63 | Structured SWL (h/w) | 4.2 |
| In class tests 4 | | الحمل الدراسي المنتظم للطالب أسبوعيا | |
| Tutorial 15 | | | |
| Final Exam 3 | | | |
| Unstructured SWL (h/sem) | 27 | Unstructured SWL (h/w) | 2.47 |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | 37 | الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2.47 |

| Library, dorm, home memorizing | 12 | | | |
|---------------------------------------|----|--|-----|--|
| Preparation for tests | 13 | | | |
| Homework | 12 | | | |
| Total SWL (h/sem) | | | | |
| الحمل الدراسي الكلي للطالب خلال الفصل | | | 100 | |
| | | | | |
| | | | | |

| | Module Evaluation | | | | | |
|------------------|------------------------|------|------------|----------|---------------------------|--|
| | تقييم المادة الدر اسية | | | | | |
| | Time/Numbe | | | Week Due | Relevant Learning Outcome | |
| Formative | Quizzes | 2 | 20% (20) | 3 and 10 | LO #1, 2, 3, 4,5 and 7 | |
| assessment | Assignments | 4 | 10% (10) | 5 and 11 | LO #1, 2, 3, 4,5 and 7 | |
| assessment | homework | 4 | 10%(10) | 4 and 12 | LO #1, 2, 3, 4,5 and 7 | |
| Summative | Midterm Exam | 2 hr | 10% (10) | 8 | LO #1 - #4 | |
| assessment | Final Exam | 3 hr | 50 % (40) | 16 | All | |
| Total assessment | | | 100% (100) | | | |

| Delivery Plan (Weekly Syllabus) | | | | | |
|---------------------------------|---|--|--|--|--|
| | المنهاج الاسبوعي النظري | | | | |
| | Material Covered | | | | |
| Week 1 | Analytical geometry. Translation of axes equations. | | | | |
| Week 2 | Conic sections (Circles – parabola-ellipse – hyperbola) | | | | |
| Week 3 | Functions of two or more variables, limit, continuity | | | | |
| Week 4 | Partial derivatives, Total differentials | | | | |
| Week 5 | Geometrical meaning of partial derivatives | | | | |
| Week 6 | Higher order partial derivatives, implicit differentiation | | | | |
| Week 7 | Relative minima and maxima of functions of two variables, linearization, small change problems | | | | |
| Week 8 | Midterm one | | | | |
| Week 9 | Polar coordinates system (Definition, polar equation, relating polar and Cartesian coordinates) | | | | |
| Week 10 | Graph in polar coordinates, | | | | |
| Week 11 | Area with polar coordinates | | | | |
| Week 12 | Vectors (Definition, properties, vectors in surface and in space) | | | | |
| Week 13 | Scalar and cross product of vectors, product of three vectors | | | | |
| Week 14 | Equations of lines. Equations of planes | | | | |
| Week 15 | Lagrange multipliers. | | | | |
| Week 16 | Final Exam | | | | |

| Learning and Teaching Resources | | |
|---------------------------------|---|---------------------------|
| | مصادر التعلم والتدريس | |
| | Text | Available in the Library? |
| Required Texts | George B. Thomas and Ross L. Finney, "Calculus and Analytic Geometry, Addison- Wesley, ISBN:0201531747. Erwin Kreyszig, Advanced Engineering Mathematics, John Wiley | Yes |
| Recommended Texts | George F. Simmons, "Calculus with Analytic Geometry", McGraw-Hill, ISBN: 0070576424 | yes |
| Websites | https://www.geogebra.org/calculator https://www.desmos.com/calculator https://www.symbolab.com/solver/integral-calculator | |





MODULE DESCRIPTOR وصف المادة الدراسية

Module Information معلومات المادة الدراسية **Module Title Energy Balance Module Delivery Module Type** Core Theory Lecture **Module Code CHE 202 Tutorial** 6 **ECTS Credits** SWL (hr/sem) 78 3 **Module Level** 2 Semester (s) offered **Administering** Chemical Engineering **College** Engineering **Department Module Leader** Yussur Dh. Abdulwahhab e-mail esar@uodiyala.edu.iq Module Leader's Acad. Module Leader's Assistant Lecturer MSc. Title Qualification **Module Tutor** None e-mail **Peer Reviewer Name** None e-mail **Review Committee Version Number Approval**

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | |
|---|--|----------------|-----------|
| Prerequisite module | Material Balance | Semester | 2 |
| Co-requisites module | None | Semester | |
| Module Aims, Lea | arning Outcomes, Indicative Contents and | d Brief Descr | iption |
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | |
| Module Aims | Learn the basic concepts of gases and energy balances in chemical engineering | | |
| أهداف المادة الدراسية | processes. It also comprehensively introduces problem-solving methods with and | | |
| | without chemical reactions. | | |
| Module Learning | Knowledge and Understanding | | |
| Outcomes | Having successfully completed this module, the | student will b | e able to |
| | demonstrate knowledge and understanding of: | | |

| parameters, reduced parameters, normal boiling point. Determination the compressibility Factor and describing the gaseous mixtures and vapor pressure. Solving the Antoine equation and material balances for involving ideal gases. Describing the liquid properties and Vapor-Liquid equilibrium for multicomponent systems. Definition of Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). The energy balance in the industry. (12 hrs). The merial and energy balance, heat of solution calculation and heat capacity (6 hrs). The merial and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational Concepts of gases energy balance, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for pressure, Jenry Shance without chemica | | _ |
|---|-------------------------------|---|
| Determination the compressibility Factor and describing the gaseous mixtures and vapor pressure. Solving the Antoine equation and material balances for involving ideal gases. Describing the liquid properties and Vapor-Liquid equilibrium for multicomponent systems. Definition of Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). The menchanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gase law, ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Pressure, Liquids, Normal boiling point, Antoine equation, Material balan | مخرجات التعلم للمادة الدراسية | |
| mixtures and vapor pressure. Solving the Antoine equation and material balances for involving ideal gases. Describing the liquid properties and Vapor-Liquid equilibrium for multicomponent systems. Definition of Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal bo | | |
| Describing the liquid properties and Vapor-Liquid equilibrium for multicomponent systems. Definition of Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). The Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gib's phase rule (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The material and energy balance for complete projects (9 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balance, involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature and | | |
| multicomponent systems. Definition of Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's quality of reaction and application of energy balance in the industry. (12 hrs). The mechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundation o | | Solving the Antoine equation and material balances for involving ideal gases. |
| Definition of Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundation of complete prosets. Vapor-Liquid Equilibrium for Multicompone | | • Describing the liquid properties and Vapor-Liquid equilibrium for |
| temperature and Gibb's phase rule. Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antioine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antonie equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemic | | multicomponent systems. |
| Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. Solving the energy balance with chemical reaction and application of energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry, (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature on Fernergy balance with chemical reaction, Energy balance with chemica | | Definition of Henry's law, Raoult's law, bubble point temperature, dew point |
| • Solving the energy balance with chemical reaction and application of energy balance in the industry. • Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. • Solving the material and energy balance for complete projects. • Solving the unsteady state for material and energy balance. Indicative content includes the following. • The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). • The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). • Antoine equation and material balances for involving ideal gases (12 hrs). • The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). • Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). • Latent heat of vaporization and Enthalpy of reaction (6 hrs). • The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). • The material and energy balance, heat of solution calculation and heat capacity (6 hrs). • The material and energy balance, heat of solution calculation and heat capacity (6 hrs). • The module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, E | | temperature and Gibb's phase rule. |
| energy balance in the industry. Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom | | Definition of Energy, Latent heat of vaporization and Enthalpy of reaction. |
| Explanation the mmechanical energy balance, heat of solution calculation and heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| nd heat capacity. Solving the material and energy balance for complete projects. Solving the unsteady state for material and energy balance. Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The merchanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Baoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Indicative Contents The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy belance with out chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, Degree of freedom of systems, Material and energy balance for complete projects. | | |
| Indicative content includes the following. The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy balance with chemical reaction, Energy balance with chemica | | Solving the material and energy balance for complete projects. |
| The ideal gas law, ideal gas mixtures, partial pressure, critical parameters, reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | Solving the unsteady state for material and energy balance. |
| reduced parameters, normal boiling point (6 hrs). The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance with chemical reaction, Energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | Indicative content includes the following. |
| * The compressibility Factor and describing the gaseous mixtures and vapor pressure (6 hrs). * Antoine equation and material balances for involving ideal gases (12 hrs). * The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). * Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). * Latent heat of vaporization and Enthalpy of reaction (6 hrs). * The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). * The material and energy balance, heat of solution calculation and heat capacity (6 hrs). * The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| pressure (6 hrs). Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Antoine equation and material balances for involving ideal gases (12 hrs). The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The material and energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| The liquid properties and Vapor-Liquid equilibrium for multicomponent systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| systems (6 hrs). Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| ## Henry's law, Raoult's law, bubble point temperature, dew point temperature and Gibb's phase rule (6 hrs). ### Latent heat of vaporization and Enthalpy of reaction (6 hrs). ### Latent heat of vaporization and Enthalpy of reaction and application of energy balance in the industry. (12 hrs). ### The meregy balance with chemical reaction and application and heat capacity (6 hrs). ### The material and energy balance, heat of solution calculation and heat capacity (6 hrs). ### The unsteady state for material and energy balance (9 hrs). ### This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Henry's law, Raount's law, bubble point temperature and Gibb's phase rule (6 hrs). Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | Indicative Contents | |
| Latent heat of vaporization and Enthalpy of reaction (6 hrs). The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| The energy balance with chemical reaction and application of energy balance in the industry. (12 hrs). The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | . 3, .3 | |
| balance in the industry. (12 hrs). • The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). • The material and energy balance for complete projects (9 hrs). • The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| The mmechanical energy balance, heat of solution calculation and heat capacity (6 hrs). The material and energy balance for complete projects (9 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| capacity (6 hrs). • The material and energy balance for complete projects (9 hrs). • The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| The material and energy balance for complete projects (9 hrs). The unsteady state for material and energy balance (9 hrs). This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | 3 |
| This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| This module is designed to provide second-year Chemical Engineering students with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| with the foundational concepts of gases energy balances, including the ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures, Vapor Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Pressure, Liquids, Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced |
| involving ideal gases, Liquid Properties, Vapor-Liquid Equilibrium for Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Multicomponent Systems, Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Dew Point Temperature and Gibb's phase rule. The energy balance part including, Energy definition, Energy forms, Latent heat of vaporization, Enthalpy of reaction, Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| Energy balance without chemical reaction, Energy balance with chemical reaction, Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | Course Description | |
| Application of energy balance in the industry, Mechanical energy balance, Heat of solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| solution calculation, Definitions of different kinds of humidity, Humidity charts and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | 0. |
| and their uses, Heat capacity, Enthalpy-concentration charts and their uses, Degree of freedom of systems, Material and energy balance for complete projects, | | |
| | | |
| Unsteady state material halance | | Degree of freedom of systems, Material and energy balance for complete projects, |
| · | | Unsteady state material balance. |
| Learning and Teaching Strategies | | |
| t wto t wto a c | | , , |
| استراتيجيات التعلم والتعليم | Strategies | Beginning to establish a strong understanding of the basics of gases and energy |
| ا مداعد و الأخد و | | , , |
| · · · | Strategies | beginning to establish a strong understanding of the basics of gases and energy |

balance processes. Using real industrial examples to help students relate the basic concepts of gases and energy balance to real industrial processes. Encouraging the discussions and questions to clear up any misconceptions. In addition, encouraging active participation and group discussions to enhance critical thinking and problem-solving skills. Guiding students through the problem-solving process and providing constructive feedback.

| Student Workload (SWL) الحمل الدر اسي للطالب | | | | | |
|--|-----------|-----|---|-----|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | | | | | |
| In class lectures | 57 | 78 | Structured SWL (h/w) | 5 | |
| In class tests | 3 | 70 | الحمل الدراسي المنتظم للطالب أسبوعيا | J | |
| Tutorial | 15 | | | | |
| Final Exam | 3 | | | | |
| Unstructured SWL (h/sem) | | | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | | He atmost and CMI (b/m) | | |
| Library, dorm, home memorizing 3 | 0 | 72 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 4.8 | |
| Preparation for tests 3 | 0 | | الحمل الدر اللي عير المنتظم للطالب اللبوعيا | | |
| Homework 1 | 2 | | | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 150 | | | |

| Module Evaluation تقييم المادة الدر اسية | | | | | | | |
|---|--|---|-----------------|----------|--------------------|--|--|
| | Time/ Number Weight (Marks) Week Due Outcome | | | | | | |
| _ | Quizzes | 2 | 20% (20) | 3 and 6 | LO #1, 2, 6, and 7 | | |
| Formative | Assignments | 2 | 10% (10) | 2 and 14 | LO # 1,2,3,5 and 6 | | |
| assessment | Home work | 2 | 10% (10) | 3 and 10 | LO # 4-9 | | |
| Summative | Midterm Exam | 1 | 10% (10) | 7 | LO # 1-5 | | |
| assessment | Final Exam | 1 | 50% (50) | 16 | All | | |
| Total assessr | nent | | 100% (100Marks) | | | | |

| Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | | | |
|--|---|--|--|--|
| | Material Covered | | | |
| Week 1 | The ideal gas law, ideal gas mixtures, partial pressure, Critical Parameters, Reduced Parameters, Compressibility Factor, Equations of State, Gaseous Mixtures 2. Vapor Pressure and Liquids. | | | |
| Week 2 Normal boiling point, Antoine equation, Material balances involving ideal gases, Liquid Properties and Vapor-Liquid Equilibrium for Multicomponent Systems. | | | | |
| Week 3 | Henry's law, Raoult's law, Bubble Point Temperature, Dew Point Temperature and Gibb's phase rule. | | | |

| Week 4 | Energy definition, Energy forms, Latent heat of vaporization and Enthalpy of reaction. |
|---------|--|
| Week 5 | Energy balance with and without chemical reaction. |
| Week 6 | Application of energy balance in the industry. |
| Week 7 | Mechanical energy balance and Heat of solution calculation |
| Week 8 | Definitions of different kinds of humidity and Humidity charts and their uses. |
| Week 9 | Heat capacity, Enthalpy-concentration charts and their uses and Degree of freedom of systems |
| Week 10 | Material and energy balance for complete projects. |
| Week 11 | Material and energy balance for complete projects. |
| Week 12 | Material and energy balance for complete projects. |
| Week 13 | Unsteady state material balance |
| Week 14 | Unsteady State energy balance |
| Week 15 | Unsteady State energy balance |
| Week 16 | Final Exam |

| Learning and Teaching Resources | | | | | |
|--|---|----|--|--|--|
| مصادر التعلم والتدريس | | | | | |
| Text Available in the Library? | | | | | |
| Required Texts | Required Texts David M Himmelblau, Basic principles and calculations in chemical engineering, Prentice Hall. | | | | |
| Recommended TextsRichard M Felder & Ronald W. Rousseau Elementary Principles of Chemical Processes, Wiley India. | | No | | | |
| Websites | Nil | | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | | |
|---|---------------|----------------------|----------------------|----------------------------------|-----------------------|--------------|--|
| Module Title | Fluid Flow | Fluid Flow I | | | ale Delivery | | |
| Module Type | Core | | | | ☑ Theory | | |
| Module Code | CHE 203 | | | | ☑ Lecture | | |
| ECTS Credits | 4 | | | | □ Lab □ Practical | | |
| SWL (hr/sem) | 100 | 100 | | | ☐ Seminar | | |
| Module Level | | 2 | Semester of Delivery | | 2 | | |
| Administering De | epartment | Chemical engineering | College | College of Engineering | | | |
| Module Leader | Walaa abid m | nahmood | e-mail | whalaa_alkhaisi76@uodiyala.edu.i | | iyala.edu.iq | |
| Module Leader's | Acad. Title | Assistant Lecturer | Module Le | ader's Q | ualification | MSc. | |
| Module Tutor | | | e-mail | | | | |
| Peer Reviewer Name | | | e-mail | | | | |
| Scientific Commi Date | ttee Approval | | Version Nu | ımber | 1.0 | | |

| Relation with other Modules | | | | | |
|-----------------------------------|------|----------|--|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | | |
| Prerequisite module None Semester | | | | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | | |
|--|--|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | |
| | The primary purpose of the study of fluid flow is to develop the ability to predict the | | | |
| Module Objectives | application of fluid in industries. In physics, physical chemistry, and engineering, fluid | | | |
| أهداف المادة الدراسية | dynamics is a sub discipline of fluid mechanics that describes the flow of fluids— | | | |
| | liquids and gases. It has several sub-disciplines, including aerodynamics (the study of | | | |
| | air and other gases in motion) and hydrodynamics (the study of liquids in motion). | | | |

| | Fluid dynamics has a wide range of applications, including calculating forces and | | | | | |
|-------------------------------|---|--|--|--|--|--|
| | | | | | | |
| | moments on aircraft, determining the mass flow rate of petroleum through pipelines, | | | | | |
| | predicting weather patterns, understanding nebulae in interstellar space, and | | | | | |
| | modeling fission weapon detonation. | | | | | |
| | Students who successfully finish this course will be eligible to: | | | | | |
| | 1.Explain the fluid, fluid dynamics, and explain the physical characteristics of fluids: | | | | | |
| | pressure, bulk modulus of elasticity, density, specific gravity, viscosity, kinematic | | | | | |
| | viscosity, Newton's law of viscosity, surface tension and capillarity, varieties of fluids, | | | | | |
| | Newtonian, non-Newtonian, ideal, and real fluids | | | | | |
| Module Learning | 2.Unit-free aggregate definition and fluid flow applications of dimensional analysis | | | | | |
| Outcomes | 3. The concept of pressure measurement instruments, the static fluid and its uses, and | | | | | |
| مخرجات التعلم للمادة الدراسية | the fluid's pressure difference between its two states of rest and motion | | | | | |
| | 4. Definition of the equation of continuity and momentum with applications | | | | | |
| | defining the different kinds of energy lost during fluid flow as a result of pressure | | | | | |
| | drops | | | | | |
| | 5. Considering the energy computations required for fluid transport through the use | | | | | |
| | of the | | | | | |

| Learning and Teaching Strategies | | | | | |
|----------------------------------|---|--|--|--|--|
| استر اتيجيات التعلم والتعليم | | | | | |
| Strategies | The main strategy that will be adopted in delivering this module is to encourage students' participation in the exercises, while at the same time refining and expanding their critical thinking skills. This will be achieved through classes, homework's and examples. Practical examples helps students to understand the course material. | | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | |
|--|-------------------|---|---|--|
| Structured SWL (h/sem) Structured SWL (h/w) 4 الحمل الدر اسي المنتظم للطالب أسبوعيا الحمل الدر اسي المنتظم للطالب خلال الفصل | | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل | 37 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 4 | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | Total SWL (h/sem) | | | |

| Module Evaluation تقييم المادة الدر اسية | | | | | | |
|---|--------------|------|----------|--------------|--------------|--|
| Time/Number Weight (Marks) Week Relevant Learning Outcome | | | | | | |
| | Quizzes | 2 | 10% (10) | 7 and 14 | LO #2 and #5 | |
| Formative assessment | Assignments | 4 | 10% (10) | 11 and 13 | LO #4 and #5 | |
| assessment | homework | 1 | 10%(10) | 14 | | |
| | Report | 2 | 10%(10) | 14 | | |
| Summative | Midterm Exam | 2 hr | 10% (10) | 15 | LO #1 - #3 | |

| assessment | Final Exam | 3 hr | 50 % (40) | 16 | All |
|------------------|------------|--------|-----------|----|-----|
| Total aggaggment | | | 100% (100 | | |
| Total assessment | | Marks) | | | |

| Delivery Plan (Weekly Syllabus) | | | | |
|---------------------------------|---|--|--|--|
| | المنهاج الاسبوعي النظري | | | |
| | Material Covered | | | |
| Week 1 | Introduction to fluid flow, Physical properties of fluids | | | |
| Week 2 | Kinematic viscosity, Newton's, law of viscosity | | | |
| Week 3 | surface tension and capillarity, bulk modulus of Elasticity, Pressure | | | |
| Week 4 | Tutorial | | | |
| Week 5 | Dimensional Analysis Fundamental dimensions, dimensions, Dimensional homogeneity, methods of | | | |
| WEEK 3 | dimensional analysis, Rayleigh's method (power series) | | | |
| Week 6 | Buckingham's П method / Theorem Dimensionless numbers | | | |
| Week 7 | Tutorial | | | |
| Week 8 | Midterm one | | | |
| Week 9 | Fluid Statics and its applications | | | |
| Week 10 | Buoyancy Measurement of pressure: (Piezometer, Manometers, types) | | | |
| Week 11 | Tutorial | | | |
| Week 12 | Fluid kinematics Types of fluid flow (steady and unsteady flows, uniform and non-uniform flows, | | | |
| WCCK 12 | one, two, and three dimensional flows, , | | | |
| Week 13 | Continuity equation | | | |
| Week 14 | Newtonian's Fluid (Incompressible flow in Pipe and Channels) Reynolds experiment | | | |
| Week 15 | Velocity distribution for laminar and turbulent flow | | | |
| Week 16 | Final Exam | | | |

| Learning and Teaching Resources | | | | | | | |
|---------------------------------|--|-----|--|--|--|--|--|
| | مصادر التعلم والتدريس | | | | | | |
| Text Available in the Library | | | | | | | |
| Required Texts | 1 R. W. Fox and A. T. McDonald, "Introduction to fluid mechanics", 5th. Ed., John Wiley & Sons, 1998. 2.R. M. Holland, "Fluid Flow", John Wiley & Sons, 1982. | Yes | | | | | |

| Recommended Texts | W. L. McCabe, J. Smith and P. Harriot, "Unit Operations of Chemical Engineering", 6th Ed., McGraw – Hill, International Edition, 2001 | yes |
|----------------------|---|-----|
| Websites | http://www.efluids.com/ | |

APPENDIX:

| GRADING SCHEME مخطط الدرجات | | | | | | |
|--------------------------------|-------------------------|-------------|-----------|---------------------------------------|--|--|
| Group | Grade | التقدير | Marks (%) | Definition | | |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | | |
| g | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors | | |
| Success Group (50 - 100) | C - Good | جيد | 70 - 79 | Sound work with notable errors | | |
| (30 - 100) | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | | |
| | | | | | | |
| Note: | | | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|--------------|-------------------------|-------------|----------------------------------|----------------|---------------|
| Module Title | Physical Che | emistry | | Modu | le Delivery | |
| Module Type | Core | | | | Theory | |
| Module Code | CHE 203 | | | | Lecture Lab | |
| ECTS Credits | 5 | | | | Tutorial | |
| SWL (hr/sem) | 125 | | | | Seminar | |
| Module Level | | 1 | Semester of | er of Delivery | | 1 |
| Administering De | epartment | Chemical Engineering | College | Engine | ering | |
| Module Leader | Nabaa Burhan | Ali | e-mail | nabaa_burhan_eng@uodiyala.edu.iq | | liyala.edu.iq |
| Module Leader's | Acad. Title | Assistant Lecturer | Module Le | Leader's Qualification Msc | | Msc |
| Module Tutor | None | | e-mail | | | |
| Peer Reviewer Name | | None | e-mail | | | |
| Scientific Committee Approval Date | | | Version Nu | mber | | |

| Relation with other Modules | | | | | |
|-----------------------------|-----------------------------------|----------|--|--|--|
| | العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | None | Semester | | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | | | |
|--|--|--|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | | |
| Module Objectives | 1.During the semester, the student learns an idea about the branches of physical | | | | |
| أهداف المادة الدراسية | chemistry and necessary facts, concepts, principles and theories of physical | | | | |
| | chemistry. | | | | |

| | Ta |
|-------------------------------|--|
| | 2.Learn and understand the principles of thermodynamics |
| | 3.Learn and understand types of thermodynamic processes |
| | 4.Learn and understand Gibbs free energy and Helmholtz energy |
| | 5.Identify the types of Phase Equilibrium |
| | 6. Learn about Henry's law and Raoult's Law |
| | 7. Learn and understand Electromotive force (EMF) of a cell |
| | 8.Learn and understand the application of chemical kinetics, and Effect of |
| | temperature on reaction rate. |
| | Understanding and teaching students the branches of physical chemistry. Enabling students to obtain knowledge and understanding Fundamentals of |
| Module Learning | thermodynamics. |
| Outcomes | 3.Enabling students to obtain knowledge and understanding of |
| مخرجات التعلم للمادة الدراسية | Thermochemistry. 4.Enable the student to obtain knowledge and understanding of Kinetic |
| | chemistry. 5. Enabling students to obtain knowledge and understanding of Electrophenistry. |
| | 5. Enabling students to obtain knowledge and understanding of Electrochemistry. Indicative content includes the following: |
| | Principles of thermodynamics (heat, work, types of systems, Zeroth, first law of |
| | thermodynamics, Thermochemistry, Effect of temperature on the reaction |
| | enthalpy) (4h) |
| | Types of thermodynamic processes(isothermal process, adiabatic process)(4h) |
| | Second and Third laws of thermodynamics (Heat engines) (4h) |
| | Entropy (Calculation of entropy, Gibbs free energy and Helmholtz energy, |
| | effect of pressure on the Gibbs energy) (4h) |
| | Chemical Equilibrium (Relation between Kc and Kp, Relation between Kp and Kx, |
| | Effect of inert gas on equilibrium) (4h) |
| | Phase Equilibrium (Phase rule, phase diagram of one component system, Phase |
| | diagram of two components system, Equilibrium between two phases, |
| | Clausius-Clapeyron equation) (6h) |
| Indicative Contents | Henry's law, Raoult's Law (Applications of osmosis, calculating Molar Mass |
| المحتويات الإرشادية | from Colligative Properties) (6h) |
| | Applications of phase rule(Types of solutions, ideal and non- ideal solutions, |
| | intermolecular forces, Solubility of gases in liquids) (6h) |
| | Electrochemical cells Electromotive force (EMF) of a cell (The polarity of |
| | electrodes, Electrode potential, The cell reaction and reversible cells) (6h) |
| | Types of half EMF, Standard electrode potentials (Thermodynamic data from |
| | cell E.M.F.S, Oxidation reduction cells, Determination of pH, Concentration |
| | cells, the temperature dependence of the EMF) (6h) |
| | |
| | Kinetic reaction (Application of chemical kinetics, The rate of chemical |
| | reactions, order of reaction, theories of chemical reactions) (4h) Rate constant (zero first, second and third order rate equations, Determination |
| | <u>-</u> |
| | of the reactions order, Effect of temperature on reaction rate, Activation energy |
| | and Arrhenius equation) (6h) This course provides students an idea the brenches of physical chemistry and |
| | This course provides students an idea the branches of physical chemistry and practical applications in physical chemistry The program has been arranged in a |
| | way that covers all areas of physical chemistry, where the kinetic theory of |
| Course Description | gases, the foundations of thermodynamics, thermochemistry, chemical |
| | 1 - |
| | equilibrium, phase equilibrium, the movement of chemical reactions, and |
| | electrochemistry were identified. |

| Learning and Teaching Strategies | | | | | | | |
|----------------------------------|--|--|--|--|--|--|--|
| | استراتيجيات التعلم والتعليم | | | | | | |
| | 1.Use different sources to obtain information. | | | | | | |
| | 2. Work within a team and a group for the purpose of completing some reports | | | | | | |
| | and solutions to duties. | | | | | | |
| Stratagies | 3.Gain experience in finding solutions to some industrial problems. | | | | | | |
| Strategies | 4.Calculating the rank of chemical reactions and finding the chemical reaction | | | | | | |
| | constant. | | | | | | |
| | 5.Calculation of the temperature of chemical reactions. Calculation of | | | | | | |
| | electrochemical cell potential | | | | | | |

| Student Workload (SWL) | | | | | | |
|---|---|--|-----|--|--|--|
| ۱ اسبوعا | الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | | |
| Structured SWL (h/sem) | | | | | | |
| الحمل الدراسي المنتظم للطالب خلال الفصل | | | | | | |
| In class lectures 43 | | | | | | |
| In class tests 2 | 94 | Structured SWL (h/w) | 6.3 | | | |
| Tutorial 15 | 74 | الحمل الدراسي المنتظم للطالب أسبوعيا | 0.3 | | | |
| Lab | | | | | | |
| 30 | | | | | | |
| Final Exam 4 | | | | | | |
| Unstructured SWL (h/sem) | | | | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | | | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | 31 | Unstructured SWL (h/w) | 2 | | | |
| Libraray, dorm, home, memorize 18 | 31 | الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2 | | | |
| Preparation for tests 8 | | | | | | |
| Homework 5 | | | | | | |
| Total SWL (h/sem) | | | | | | |
| الحمل الدراسي الكلي للطالب خلال الفصل | | | | | | |

| Module Evaluation | | | | | | | | |
|---|-----------------|-----|----------|-------------------|------------------------------------|--|--|--|
| تقييم المادة الدراسية | | | | | | | | |
| Time/Number Weight (Marks) Week Due Outcome | | | | | | | | |
| | Quizzes | 4 | 20% (20) | 3, 6,10 and 13 | LO #1, 2, 4, 5, 8, 9, 11 and 12 | | | |
| Formative assessment | Assignments | 4 | 10% (10) | 2, 5, 9 and 12 | LO #1, 4, 10 and 11 | | | |
| | Projects / Lab. | 15 | 10% (10) | Continuous | All | | | |
| | Seminars | | | | | | | |
| Summative | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - #7 | | | |

| assessment | Final Exam | 3hr | 50% (50) | 16 | All |
|----------------|------------|-----|------------------|----|-----|
| Total assessme | ent | | 100% (100 Marks) | | |

| Delivery Plan (Weekly Syllabus) | | | | | | | | |
|---------------------------------|--|--|--|--|--|--|--|--|
| | المنهاج الاسبوعي النظري | | | | | | | |
| | Material Covered | | | | | | | |
| Week 1 | Principles of thermodynamics, heat, work, types of systems, Zeroth, first law of thermodynamics, Thermochemistry, Effect of temperature on the reaction enthalpy, | | | | | | | |
| Week 2 | Types of thermodynamic processes, isothermal process, adiabatic process | | | | | | | |
| Week 3 | Second and Third laws of thermodynamics, Heat engines | | | | | | | |
| Week 4 | Entropy, Calculation of entropy, Gibbs free energy and Helmholtz energy, effect of pressure on the Gibbs energy | | | | | | | |
| Week 5 | Chemical Equilibrium, Relation between Kc and Kp, | | | | | | | |
| Week 6 | Relation between Kp and Kx, Effect of inert gas on equilibrium | | | | | | | |
| Week 7 | Mid-term Exam - | | | | | | | |
| Week 8 | Phase Equilibrium, Phase rule, phase diagram of one component system, Phase diagram of two components system, Equilibrium between two phases, Clausius-Clapeyron equation. | | | | | | | |
| Week 9 | Henry's law, Raoult's Law, Applications of osmosis, calculating Molar Mass from Colligative Properties | | | | | | | |
| Week 10 | Applications of phase rule, Types of solutions, ideal and non- ideal solutions, intermolecular forces, Solubility of gases in liquids | | | | | | | |
| Week 11 | Electrochemical cells Electromotive force (EMF) of a cell, The polarity of electrodes, Electrode potential, The cell reaction and reversible cells, | | | | | | | |
| Week 12 | Types of half EMF, Standard electrode potentials, Thermodynamic data from cell E.M.F.S., | | | | | | | |
| Week 13 | Oxidation reduction cells, Determination of pH, Concentration cells, the temperature dependence of the EMF | | | | | | | |
| Week 14 | Kinetic reaction, Application of chemical kinetics, The rate of chemical reactions, order of reaction, theories of chemical reactions | | | | | | | |
| Week 15 | Rate constant zero, first, second and third order rate equations, Determination of the reactions order, Effect of temperature on reaction rate, Activation energy and Arrhenius equation | | | | | | | |
| Week 16 | Final Exam | | | | | | | |

| | Delivery Plan (Weekly Lab. Syllabus) | | | | | | |
|--------|---|--|--|--|--|--|--|
| | المنهاج الاسبوعي للمختبر | | | | | | |
| | Material Covered | | | | | | |
| Week 1 | Determine the density of a liquid by pyknometry | | | | | | |
| Week 2 | Distribution of acetic acid between toluene and water | | | | | | |
| Week 3 | The solubility of solids in liquids | | | | | | |
| Week 4 | Determine the viscosity of a pure liquid(nitrobenzene) and its variation with temperature | | | | | | |
| Week 5 | Determine the surface tension of a liquid by the double capillary method at different | | | | | | |
| | temperatures | | | | | | |

| Week 6 | Adsorption of acetic acid with activated carbon particles |
|--------|--|
| Week 7 | Find the solubility between phenol and water |
| Week 8 | Calculate the molecular weights of the freezing point depression |
| Week 9 | Study the variation of the vapour pressure of a pure liquid with temperature |
| Week10 | Determine of the relative molecular mass of chloroform by Victor Meyer's method |
| Week11 | Determine the solubility of benzoic acid over a range of temperatures and hence calculate its heat of solution |
| Week12 | Determine the adsorption isotherm of acetic acid from aqueous solution by charcoal |
| Week13 | Study the Kinetics of the hydrolysis of methyl acetate catalyzed by hydrochloric acid |
| Week14 | Determine the rate constant of the hydrolysis of ethyl acetate by sodium hydroxide |
| Week15 | Determine the rate constant and the energy of activation of the reaction between hydrogen peroxide and hydrogen iodide |

| Learning and Teaching Resources | | | | | | | | | |
|---------------------------------|--|---------------------------|--|--|--|--|--|--|--|
| | مصادر التعلم والتدريس | | | | | | | | |
| | Text | Available in the Library? | | | | | | | |
| Required Texts | 1.Robert A. Alberty "Physical Chemistry " seven edition | Yes | | | | | | | |
| Recommended Texts | 1.K.K. Sharma "Physical Chemistry" 2.Arun Bahl, B.S.Bahl, G.D.Tuli "Essentials of Physical Chemistry "S.Chand and Company ltd 2008 | Yes | | | | | | | |
| Websites | | | | | | | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدر اسية | | | | | | | |
|--|-------------|----------------------|------------------------|-----|--------------------------------|--------------|---------------|
| Module Title | Petroleum | and Natural gas | | | Modu | ıle Delivery | |
| Module Type | Core | | | | Theory Lecture Tutorial | | |
| Module Code | CHE 205 | | | | | | |
| ECTS Credits | 5 | | | | | Lab | |
| SWL (hr/sem) | 79 | 79 | | | | | |
| Module Level | | 2 | Semester (s) offered 3 | | | 3 | |
| Administering De | partment | Chemical Engineering | College | En | Engineering | | |
| Module Leader | Dr. Ahmed | Abbas Fadhil | e-mail | Ał | Ahmed.a.fadhil@uodiyala.edu.iq | | diyala.edu.iq |
| Module Leader's | Acad. Title | Lecturer | Module Le | ade | r's Qua | lification | Ph.D. |
| Module Tutor None | | | e-mail | | | | |
| Peer Reviewer Name None | | | e-mail | | | | |
| Review Committee Approval | | Version Nu | umb | er | | | |

| Relation with Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | | | |
|--|----------|----------|---|--|--|--|--|
| Prerequisite module | | Semester | 1 | | | | |
| Co-requisites module | Semester | | | | | | |
| Module Aims, Learning Outcomes, Indicative Contents and Brief Description أهداف المادة الدر اسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | | | | |
| *Petroleum* 1. *Understanding Petroleum:* - Define petroleum and explain its origins Identify and describe the chemical composition of crude oil Classify different types of crude oil based on their properties and composition. | | | | | | | |

- 2. *Preparation of Crude Oil for Refining/Distillation: * - Explain the processes involved in crude oil degassing and stabilization. - Discuss methods for sulfur recovery from petroleum gases. - Describe techniques for treating emulsions and dehydration in crude oil. - Explain the desalting process in crude oil preparation. 3. *Refining/Distillation of Crude Oil: * - Understand the operation of simple fractional distillation columns. - Describe the process and importance of atmospheric complex fractional distillation. - Explain vacuum complex fractional distillation and its applications. - Introduce azeotropic distillation and its significance in refining. 4. *Petroleum Products: * - Identify various petroleum products including LPG, gasoline, kerosene, gas oil, lubricating oils, wax, and asphalt. - Discuss the uses and applications of different petroleum products. 5. *Physical Properties of Petroleum Products: * - Understand mechanical properties such as density, specific gravity, viscosity and viscosity index, and surface tension. - Study thermal properties including cloud point, pour point, freezing point, and melting point. - Examine optical properties like boiling point, flash point, fire point, autoignition point, octane number, aniline point, diesel index, specific heat, volatility, and refractive index. *Natural Gas* 1. *Understanding Natural Gas: * - Define natural gas and describe its chemical composition. - Discuss the various uses of natural gas. - Differentiate between types of gas reservoirs and classify associated and nonassociated gases. - Study the properties of natural gas. 2. *Natural Gas Processing: * - Explain the methods for condensate and water removal. - Describe acid gas removal processes and the sweetening of sour natural gas. - Discuss sulfur recovery using the Claus process. - Understand dehydration (moisture removal) and mercury removal techniques. - Explain nitrogen rejection, NGL recovery, and fractionation of natural gas liquids (NGL). 3. *Natural Gas Transmission: * - Discuss the principles and practices of liquefied natural gas (LNG) and compressed natural gas (CNG) transmission.
 - Explain the concept and importance of the heating value (HV) of fuel in natural gas transmission.

Module Learning Outcomes

Petroleum

- 1. *Definition and Origin of Petroleum*
 - Understand and define petroleum and its origins.
 - Explain the chemical composition and classification of crude oil.

مخرجات التعلم للمادة الدراسية

- 2. *Preparation of Crude Oil for Refining/Distillation*
 - Describe the processes involved in crude oil degassing and stabilization.
 - Explain the techniques for sulfur recovery from petroleum gases.

- Understand emulsion treatment, dehydration, and desalting processes in crude oil preparation.
- 3. *Refining/Distillation of Crude Oil*
- Describe the operation of a simple fractional distillation column.
- Explain the functions and processes within an atmospheric complex fractional distillation column.
- Understand the principles of vacuum complex fractional distillation and azeotropic distillation.
- 4. *Petroleum Products and Their Uses*
- Identify various petroleum products including LPG, gasoline, kerosene, gas oil, lubricating oils, wax, and asphalt.
 - Explain the various uses of crude oil and its products.
- 5. *Physical Properties of Petroleum Products*
- Define and measure the mechanical properties such as density, specific gravity, viscosity and viscosity index, and surface tension.
- Explain the thermal properties including cloud point, pour point, freezing point, and melting point.
- Understand the optical properties like boiling point, flash point, fire point, auto-ignition point, octane number, aniline point, diesel index, specific heat, volatility, and refractive index.

Natural Gas

- 1. *Introduction to Natural Gas*
 - Define natural gas and explain its chemical composition.
- Understand the various uses of natural gas and differentiate between types of gas reservoirs.
 - Explain the differences between associated and non-associated gases.
- 2. *Natural Gas Properties*
 - Understand the physical and chemical properties of natural gas.
- 3. *Natural Gas Processing*
 - Explain the processes involved in condensate and water removal.
- Understand acid gas removal (sweetening of sour natural gas) and sulfur recovery using the Claus process.
- Describe the methods for dehydration (moisture removal), mercury removal, and nitrogen rejection.
 - Understand NGL recovery and the fractionation of natural gas liquids.
- 4. *Natural Gas Transmission*
- Describe the processes and techniques involved in the transmission of natural gas.
- Understand the differences and uses of liquefied natural gas (LNG) and compressed natural gas (CNG).
 - Explain the heating value (HV) of fuel and its importance in natural gas applications.

4 ...

Petroleum

Indicative Contents المحتويات الإرشادية

- 1. *Introduction to Petroleum*
 - Definition of petroleum
- Origin and formation of petroleum
- Chemical composition of crude oil
- Classification of crude oil (paraffinic, naphthenic, aromatic, and mixed-base)
- 2. *Preparation of Crude Oil for Refining/Distillation*

- Crude oil degassing: methods and importance
- Crude oil stabilization: principles and techniques
- Sulfur recovery from petroleum gases: processes (Claus process, etc.)
- Emulsion treatment and dehydration: methods (chemical, electrical, and thermal)
 - Desalting of crude oil: single-stage and multi-stage desalting processes

3. *Refining/Distillation of Crude Oil*

- Simple fractional distillation: principles, equipment, and operation
- Atmospheric complex fractional distillation: design and operation
- Vacuum complex fractional distillation: purpose, equipment, and processes
- Azeotropic distillation: principles, applications, and examples

4. *Petroleum Products and Their Uses*

- LPG: production, properties, and uses
- Gasoline: composition, properties, and uses
- Kerosene: properties and uses
- Gas oil (diesel): production, properties, and applications
- Lubricating oils: types, properties, and uses
- Wax: production, properties, and uses
- Asphalt: production, properties, and applications

5. *Physical Properties of Petroleum Products*

- Mechanical properties: density, specific gravity, viscosity, viscosity index, surface tension
 - Thermal properties: cloud point, pour point, freezing point, melting point
- Optical properties: boiling point, flash point, fire point, auto-ignition point, octane number, aniline point, diesel index, specific heat, volatility, refractive index

Natural Gas

- 1. *Introduction to Natural Gas*
 - Definition and formation of natural gas
 - Chemical composition of natural gas
 - Uses of natural gas in various industries
 - Types of gas reservoirs: conventional, unconventional
 - Associated vs. non-associated gases: definitions and examples

2. *Properties of Natural Gas*

- Physical properties: phase behavior, density, viscosity
- Chemical properties: methane content, impurities

3. *Natural Gas Processing*

- Condensate and water removal: methods and equipment
- Acid gas removal: techniques (amine gas treating, membrane separation)
- Sulfur recovery: Claus process, SuperClaus process, and other methods
- Dehvdration: methods (glvcol dehvdration, molecular sieves)
- Mercury removal: techniques and importance
- Nitrogen rejection: cryogenic distillation, adsorption methods
- NGL recovery: processes and equipment
- Fractionation of natural gas liquids: distillation columns, separation techniques

4. *Natural Gas Transmission*

- Liquefied Natural Gas (LNG): production, transportation, and storage
- Compressed Natural Gas (CNG): production, transportation, and applications
- Heating value (HV) of fuel: definition, measurement, and significance

Course Description

This course provides an in-depth exploration of the petroleum and natural gas industries, focusing on their origins, compositions, processing methods, and

applications. Students will begin by understanding the definition and formation of petroleum and natural gas, followed by a detailed study of the chemical composition and classification of crude oil and natural gas.

The course delves into the preparation of crude oil for refining and distillation, covering essential processes such as degassing, stabilization, sulfur recovery, emulsion treatment, dehydration, and desalting. Students will learn about various refining and distillation techniques, including simple and complex fractional distillation, vacuum distillation, and azeotropic distillation.

An examination of petroleum products such as LPG, gasoline, kerosene, gas oil, lubricating oils, wax, and asphalt will be conducted, along with their specific uses and applications. The course will also cover the physical properties of these petroleum products, including mechanical, thermal, and optical properties.

In the natural gas section, students will study the types of gas reservoirs, associated and non-associated gases, and the various physical and chemical properties of natural gas. The course will cover natural gas processing techniques such as condensate and water removal, acid gas removal, sulfur recovery, dehydration, mercury removal, nitrogen rejection, NGL recovery, and fractionation.

Learning and Teaching Strategies استراتيجيات التعلم والتعليم

- 1. *Lectures and Interactive Sessions*
- *Overview: * Deliver structured lectures to introduce key concepts, theories, and processes.
- *Interactive Component: * Incorporate Q&A sessions, real-time polling, and discussion prompts to encourage student participation and engagement.
- 2. *Hands-On Laboratory Exercises*
- *Overview: * Provide practical experience with laboratory equipment and simulations related to petroleum and natural gas processing.
- *Activities: * Perform experiments on crude oil stabilization, distillation processes, and natural gas sweetening, and measure physical properties of petroleum products.
- 3. *Case Studies and Industry Examples*
- *Overview: * Utilize real-world case studies to illustrate the application of theoretical concepts in industry settings.
- *Application: * Analyze successful and challenging projects in petroleum and natural gas industries, examining methods used and lessons learned.
- 4. *Field Trips and Industry Visits*
- *Overview: * Arrange visits to refineries, natural gas processing plants, and research facilities.
- *Exposure: * Provide students with firsthand observation of industrial processes and the opportunity to interact with professionals in the field.
- 5. *Guest Lectures and Expert Panels*
- *Overview: * Invite industry experts and researchers to share their experiences and insights.
- *Integration: * Schedule guest lectures on specialized topics and organize panel discussions to address current trends and challenges in the industry.
- 6. *Group Projects and Presentations*
- *Overview: * Encourage collaborative learning through group projects focused on specific aspects of petroleum and natural gas processing.
- *Deliverables: * Have student groups present their findings and recommendations, fostering teamwork and communication skills.

Strategies

- 7. *Problem-Based Learning (PBL)*
- *Overview: * Engage students in solving real-world problems related to petroleum and natural gas industries.
- *Method: * Present scenarios and challenges for students to analyze and develop solutions, promoting critical thinking and application of knowledge.
- 8. *Simulation Software and Digital Tools*
- *Overview:* Integrate the use of simulation software to model petroleum refining and natural gas processing.
- *Tools:* Use software like Aspen HYSYS, ChemCAD, or similar tools to simulate processes, enhancing understanding through virtual experimentation.

| Student Workload (SWL) الحمل الدر اسي للطالب | | | | | | |
|--|---------------|-----|--|-----|--|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | 4 5 | | Structured SMI (b/vv) | | | |
| In class lectures In class tests Lab | 45 4 30 | 79 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 2 | | |
| Final Exam | 3 | | | | | |
| | 0 80 12 | 72 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 4.8 | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 151 | | | | |

| Module Evaluation | | | | | | | | |
|-----------------------|---|---|----------|------------|--------------------|--|--|--|
| تقييم المادة الدراسية | | | | | | | | |
| | Time/ Weight (Marks) Week Due Relevant Learning Outcome | | | | | | | |
| | Quizzes | 2 | 10% (10) | 3 and 6 | LO #1, 2, 6, and 7 | | | |
| Formative | Semen | 1 | 10% (10) | 2 and 14 | LO # 1,2,3,5 and 6 | | | |
| assessment | Projects / Lab. | 1 | 10% (10) | Continuous | All | | | |
| | Home work | 2 | 10% (10) | 3 and 10 | LO # 4-9 | | | |
| Summative | Midterm Exam | 1 | 10% (10) | 7 | LO # 1-5 | | | |
| assessment | Final Exam | 1 | 50% (50) | 16 | All | | | |
| Total assessm | Total assessment 100% (100Marks) | | | | | | | |

| Delivery Plan (Weekly Syllabus) |
|---------------------------------|
| المنهاج الاسبوعي النظري |

Material Covered

Week Delivery Plan (Weekly Syllabus for 3 Hours per Week)

Week 1: Introduction to Petroleum and Natural Gas

- *Topics: *

- Definition and origin of petroleum and natural gas
- Chemical composition of crude oil and natural gas
- Classification of crude oil
- *Activities: *
- Lecture and introductory discussion (1.5 hours)
- Overview of course structure and expectations (30 minutes)
- Reading assignment: Introductory chapters on petroleum and natural gas (1 hour)

Week 2: Chemical Composition and Classification of Crude Oil

- *Topics: *
- Detailed chemical composition of crude oil
- Classification of crude oil: paraffinic, naphthenic, aromatic, mixed-base
- *Activities: *
- Lecture and discussion (1.5 hours)
- Group activity: Classifying samples of crude oil (1.5 hours)

Week 3: Crude Oil Degassing and Stabilization

- *Topics: *
- Crude oil degassing: methods and importance
- Crude oil stabilization: principles and techniques
- *Activities: *
- Lecture and demonstration (1.5 hours)
- Lab exercise: Degassing and stabilization techniques (1.5 hours)

Week 4: Sulfur Recovery and Emulsion Treatment

- *Topics: *
- Sulfur recovery from petroleum gases: Claus process
- Emulsion treatment and dehydration methods
- *Activities: *
- Lecture and discussion (1.5 hours)
- Lab exercise: Emulsion breaking techniques (1.5 hours)

Week 5: Desalting of Crude Oil

- *Topics: *
- Desalting processes: single-stage and multi-stage desalting
- *Activities: *
- Lecture and demonstration (1.5 hours)
- Lab exercise: Desalting process simulation (1.5 hours)

Week 6: Simple Fractional Distillation

- *Topics:*
- Principles of simple fractional distillation
- Equipment and operation
- *Activities: *
- Lecture and demonstration (1.5 hours)
- Lab exercise: Simple distillation process (1.5 hours)

Week 7: Atmospheric and Vacuum Distillation

- *Topics: *
- Atmospheric complex fractional distillation
- Vacuum complex fractional distillation
- *Activities: *
- Lecture and discussion (1.5 hours)
- Lab exercise: Atmospheric and vacuum distillation simulation (1.5 hours)

Week 8: Mid-term Exam and Review

- *Activities: *
- Mid-term exam covering Weeks 1-7 content (1.5 hours)
- Review session to discuss exam results and key concepts (1.5 hours)

Week 9: Azeotropic Distillation

- *Topics: *
- Principles and applications of azeotropic distillation
- *Activities: *
- Lecture and demonstration (1.5 hours)
- Lab exercise: Azeotropic distillation process (1.5 hours)

Week 10: Petroleum Products and Their Uses

- *Topics: *
- LPG, gasoline, kerosene, gas oil, lubricating oils, wax, asphalt
- Uses and applications of petroleum products
- *Activities: *
- Lecture and discussion (1.5 hours)
- Group activity: Research and presentation on petroleum products (1.5 hours)

Week 11: Physical Properties of Petroleum Products (Part 1)

- *Topics: *
- Mechanical properties: density, specific gravity, viscosity, viscosity index, surface tension
- *Activities: *
- Lecture and demonstration (1.5 hours)
- Lab exercise: Measuring mechanical properties (1.5 hours)

Week 12: Physical Properties of Petroleum Products (Part 2)

- *Topics: *
- Thermal properties: cloud point, pour point, freezing point, melting point
- Optical properties: boiling point, flash point, fire point, auto-ignition point, octane number, aniline point, diesel index, specific heat, volatility, refractive index
- *Activities: *
- Lecture and demonstration (1.5 hours)
- Lab exercise: Measuring thermal and optical properties (1.5 hours)

Week 13: Natural Gas Properties and Processing

- *Topics: *
- Physical and chemical properties of natural gas
- Condensate and water removal, acid gas removal
- *Activities: *
- Lecture and discussion (1.5 hours)
- Lab exercise: Natural gas processing techniques (1.5 hours)

Week 14: Sulfur Recovery, Dehydration, and Mercury Removal

- *Topics: *
- Sulfur recovery: Claus process
- Dehydration methods: glycol dehydration, molecular sieves
- Mercury removal techniques
- *Activities: *
- Lecture and discussion (1.5 hours)
- Lab exercise: Sulfur recovery and dehydration techniques (1.5 hours)

Week 15: Nitrogen Rejection, NGL Recovery, and Fractionation / Natural Gas Transmission and Capstone Project Presentation

- *Topics: *
- Nitrogen rejection: cryogenic distillation, adsorption methods
- NGL recovery and fractionation processes
- LNG and CNG production, transportation, and storage
- Heating value (HV) of fuel
- *Activities: *
 - Lecture and discussion (1.5 hours)
- Capstone project presentations by student groups (1.5 hours)

Week 16

- Final exam

This 15-week syllabus with 3 hours per week ensures a comprehensive understanding of the petroleum and natural gas, blending theoretical knowledge with practical skills insights.

| Learning and Teaching Resources | | | | | | | | |
|---------------------------------|---|----------|--|--|--|--|--|--|
| | مصادر التعلم والتدريس | | | | | | | |
| | Available in the | | | | | | | |
| | | Library? | | | | | | |
| | 1. W.L Nelson, Petroleum Refinery Engineering, 4th edition, 1985. | | | | | | | |
| Required Texts | 2. William D. | Yes | | | | | | |
| | McCain, The Properties of Petroleum Fluids, 1990. | | | | | | | |
| | - 1. Saeid Mokhatab, William A.Poe and James G.Speight. "Hand | | | | | | | |
| | book of Natural Gas Transmission and Processing" 2006. | | | | | | | |
| | | | | | | | | |
| Recommended Texts | 2. Arthur J. Kidnay & William R. Parrish "Fundamentals of Natural | No | | | | | | |
| | Gas Processing" 2006. | | | | | | | |
| | | | | | | | | |
| | 3. Xiuli Wang XGAS "Advanced Natural Gas Engineering" 2009. | | | | | | | |
| Websites | Nil | | | | | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | | | |
|---|-------------|-------------------------|------------|------------------------------|-----------------|-----|--|--|
| Module Title | Computer P | rogramming I | | Modu | Module Delivery | | | |
| Module Type | Core | | | | - Theory | | | |
| Module Code | CHE 206 | | | \ \ \ \ | Lecture Lab | | | |
| ECTS Credits | 4 | | | | Tutorial | | | |
| SWL (hr/sem) | √ Seminar | | | | | | | |
| Module Level | | 2 | Semester o | f Delivery 1 | | | | |
| Administering De | epartment | Chemical Engineering | College | Engineering | | | | |
| Module Leader | Sura Fahmy | Yousif | e-mail | Sura.fahmy@uodiyala.edu.iq | | | | |
| Module Leader's | Acad. Title | Assistant Prof. | Module Le | odule Leader's Qualification | | MSc | | |
| Module Tutor None | | | e-mail | - | | | | |
| Peer Reviewer Name | | None | e-mail | - | | | | |
| Scientific Committee Approval Date | | | Version Nu | ımber | 1.0 | | | |

| Relation with other Modules | | | |
|-----------------------------------|------|----------|---|
| العلاقة مع المواد الدراسية الأخرى | | | |
| Prerequisite module | None | Semester | - |
| Co-requisites module | None | Semester | - |

| Module Aims, Learning Outcomes and Indicative Contents | | |
|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | |
| Module Objectives | 1. Teaching the student during the semester about MATLAB program specifications, | |
| أهداف المادة الدراسية | algorithm design and development, coding, and testing using a modern software | |

| | 11 |
|--|---|
| | development environment. Teaching the student how to write programs in a high-level object-oriented programming language including algorithms, diagrams, problem solving, programming concepts, methods, control structures, arrays, and strings. Providing the students with problem-solving skills and their application to solve computer problems. Developing the student's mind and enabling him to write any program in the MATLAB language through the application of different programs in all fields. Clarifying the importance of computer science for students at present and in the future, and keeping pace with the development of this science in various disciplines. |
| Module Learning Outcomes | Enabling the student to know the concepts of information technology by learning the basics of MATLAB language. Enabling the student to know MATLAB instructions and how to use each instruction inside the program. Enabling the student how to write functions and use them repeatedly within each program. Enabling the student how to use MATLAB language to draw diagrams and models. Enabling the student how to write programs in MATLAB language for any |
| مخرجات التعلم للمادة الدراسية | required mathematical equation or diagram. Enabling the student to think about the importance of programming using the MATLAB language in facilitating contemporary life. Enabling the student to think about the importance of the impact of the MATLAB language on the development of scientific research methods. Enabling the student to think and follow the rapid development of the MATLAB language. |
| Indicative Contents المحتويات الإرشادية | Indicative content includes the following: A brief history importance of MATLAB [4 hrs]. Explaining the MATLAB language using simplified programs [8 hrs]. Types of data, variables, arithmetic functions, numbers, and operations in MATLAB using subprograms [8 hrs]. Solving differential equations, derivative, and integration in MATLAB [12 hrs]. Plot functions using applied examples in MATLAB [8 hrs]. Arrays, matrices, and their arithmetic operations in MATLAB [12 hrs]. Basic control and conditional functions with examples in MATLAB [12 hrs]. |
| Course Description | This course provides students an idea of fundamental principles and applications of programming using MATLAB language. Types of data, constants, variables, characters, numbers, operations, instructions and mathematical functions in MATLAB. Solving algebraic equations, differential equations, differentiation, limited integration, unlimited integration, and limits in MATLAB. Control flow, relational, logical operators, and loops in MATLAB. Basic plotting including 2D and 3D drawing, adding titles, axis labels, annotations, specifying line styles and colors of the graph, subplots in MATLAB. Generation of arrays and matrices and their arithmetic operations, solving linear equations using matrices in MATLAB language. |

| Learning and Teaching Strategies | | | | |
|----------------------------------|--|--|--|--|
| | استراتيجيات التعلم والتعليم | | | |
| Strategies | The teacher prepares lectures on the subject in paper and electronic form and presents them to the students. The teacher explains the theoretical and practical lectures using modern teaching and presentation methods. The teacher explains lectures in detail. The teacher requests periodic reports and homework assignments on the basic topics of the subject. The teacher encourages students to participate in discussions to clear up any misconceptions. The teacher asks questions and gives exercises that require analysis and conclusions related to lectures. The teacher uses different examples to connect the theoretical and practical sides. The teacher encourages students to do the practical part and work in groups. | | | |

| Student Workload (SWL) | | | | |
|---|---|--|-----|--|
| ىبوعا | الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | |
| Structured SWL (h/sem) | | | | |
| الحمل الدراسي المنتظم للطالب خلال الفصل | | | | |
| In class lectures 13 | 64 | Structured SWL (h/w) | 4.2 | |
| In class tests 4 | 04 | الحمل الدراسي المنتظم للطالب أسبوعيا | 7.2 | |
| Lab 43 | | | | |
| Final Exam 4 | | | | |
| Unstructured SWL (h/sem) | | | | |
| الحمل الدراسي غير المنتظم للطالب خلال الفصل | | The second secon | | |
| Libraray, dorm, home, memorize 14 | 36 | Unstructured SWL (h/w) | 2.4 | |
| Preparation for tests 14 | | الحمل الدراسي غير المنتظم للطالب أسبوعيا | | |
| Homework 8 | | | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 100 | | |
| <u> </u> | | | | |

| Module Evaluation تقييم المادة الدراسية | | | | | |
|---|-------------|-------------|----------------|---------------------------|--|
| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome |
| Formative | Quizzes | 4 | 10% (10) | 3, 6, 9 and 12 | LO #1, 2, 3, 4, 5, 7, 8, 9, 10 and 11 |
| assessment | Assignments | 6 | 10% (10) | 3, 5, 8, 11, 12 and 13 | LO #1, 4, 10 and 11 |

| | Projects / Lab. | 1 | 10% (10) | Continuous | All |
|------------------|-----------------|-----------|----------|------------|------------------|
| | Report | 1 | 10% (10) | 13 | LO #3, #4 and #6 |
| Summative | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - #8 |
| assessment | Final Exam | 3hr | 50% (50) | 16 | All |
| Total assessment | | 100% (100 | | | |
| | | Marks) | | | |

| Delivery Plan (Weekly Syllabus) | | | |
|---------------------------------|---|--|--|
| | المنهاج الاسبوعي النظري | | |
| | Material Covered | | |
| Week 1 | A brief history importance of MATLAB | | |
| Week 2 | Explains the MATLAB program using simplified programs | | |
| Week 3 | Explains the MATLAB program using subprograms | | |
| Week 4 | Types of data, variables, arithmetic functions, numbers, and operations in MATLAB | | |
| Week 5 | Write and solve algebraic equations in MATLAB | | |
| Week 6 | Write and solve differential equations in MATLAB | | |
| Week 7 | Write and solve differentiation in MATLAB | | |
| Week 8 | Write and solve limited integration, unlimited integration, and limits in MATLAB | | |
| Week 9 | Plots 2D functions using applied examples in MATLAB | | |
| Week 10 | Plots 3D functions using applied examples in MATLAB | | |
| Week 11 | Adding titles, axis labels, annotations, specifying line styles and colors of the graph, subplots in MATLAB | | |
| Week 12 | Arrays and their arithmetic operations in MATLAB | | |
| Week 13 | Matrices and their arithmetic operations in MATLAB | | |
| Week 14 | Basic programs about control functions in MATLAB | | |
| Week 15 | Basic programs about conditional functions in MATLAB | | |
| Week 16 | Final Exam | | |

| Delivery Plan (Weekly Lab. Syllabus) | | | |
|--------------------------------------|--|--|--|
| | المنهاج الاسبوعي للمختبر | | |
| | Material Covered | | |
| Week 1 | Introduction to the lab and get started with use of MATLAB program on the computer | | |
| Week 2 | Explains the MATLAB program using simplified programs | | |
| Week 3 | Explains the MATLAB program using subprograms | | |
| Week 4 | Types of data, variables, arithmetic functions, numbers, and operations in MATLAB | | |
| Week 5 | Write and solve algebraic equations in MATLAB | | |
| Week 6 | Write and solve differential equations in MATLAB | | |

| Week 7 | Write and solve differentiation in MATLAB |
|--------|---|
| Week 8 | Write and solve limited integration, unlimited integration, and limits in MATLAB |
| Week 9 | Plots 2D functions using applied examples in MATLAB |
| Week10 | Plots 3D functions using applied examples in MATLAB |
| Week11 | Adding titles, axis labels, annotations, specifying line styles and colors of the graph, subplots in MATLAB |
| Week12 | Arrays and their arithmetic operations in MATLAB |
| Week13 | Matrices and their arithmetic operations in MATLAB |
| Week14 | Basic programs about control functions in MATLAB |
| Week15 | Basic programs about conditional functions in MATLAB |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|---|---|--|--|
| | Text Available in the Library? | | | |
| | Chemical Engineering Computation with MATLAB®, 1st | | | |
| | Edition, and Professor Yeong Koo Yeo teaches chemical | | | |
| | engineering at the College of Engineering of Hanyang University, | | | |
| | Seoul, South Korea. | | | |
| Required Texts | Essential MATLAB for Engineers and Scientists, Fourth Edition, | - | | |
| | Brian H. Hahn, Daniel T. Valentine | | | |
| | • Essential MATLAB for Engineers and Scientists, Fifth Edition, | | | |
| | Brian H. Hahn, Daniel T. Valentine | | | |
| Recommended | All solid scientific magazines and periodicals related to programming | | | |
| Texts | in the MATLAB language. | | | |
| Websites | 1. http://www.cprogramming.com/ | | | |
| TT CUSICES | 2. http://www.mathworks.com/ | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|---------------------------------------|-------------------|-----------------------------------|-----------|--------------------------|------|
| Module Title | ىث في العراق Extinct Ba'ath Party) | | جرائم نظام البعث ف (rty Crimes | Modu | ıle Delivery | |
| Module Type | Basic learning | g activities | | | ☑ Theory | |
| Module Code | UD05 | | | | ☑ Lecture □ Lab | |
| ECTS Credits | 2 | | | | ☐ L Tutorial | |
| SWL (hr/sem) | r/sem) 50 | | | | □ Practical 図 Seminar | |
| Module Level | | 2 | Semester of | f Deliver | ·y | |
| Administering De | partment | جميع اقسام الكلية | College | College | e of | |
| Module Leader | | | e-mail | | | |
| Module Leader's Acad. Title | | | Module Le | ader's Q | ualification | MSc. |
| Module Tutor | | | e-mail | | | |
| Peer Reviewer Name | | | e-mail | | | |
| Scientific Committee Approval 9/09/20 | | 9/09/2024 | Version Nu | mber | 1.0 | |

| Relation with other Modules | | | | |
|-----------------------------------|------|----------|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | None | Semester | | |
| Co-requisites module | None | Semester | | |

| Module Aims, Learning Outcomes and Indicative Contents أهداف المادة الدر اسية ونتائج التعلم والمحتويات الإرشادية | | | |
|---|--|--|--|
| التعرف على ماهية الجريمة لغة واصطلاحاً وماهية أقسام الجرائم. التعرف على جرائم نظام البعث وفق قانون المحكمة الجنائية العراقية العليا عام 2005م. تنمية وعي الطلب بجرائم نظام البعث وفق توثيق قانون المحكمة الجنائية العراقية العليا لسنة 2005م. | | | |
| | | | |

| | دراسة الجرائم التي ارتكبها نظام البعث على مدى سنوات طويلة واثار ها النفسية والاجتماعية. |
|----------------------------------|--|
| | التعرف على صور انتهاكات حقوق الانسان وجرائم السلطة والتعرف على الجرائم البيئية لنظام البعث في العراق. |
| | 6. تعزيز الوعي بحقيقة ما جرى من مآسي المقابر الجماعية المرتكبة من النظام البعثي في العراق. |
| | تمكين الطالب من معرفة المفاهيم النظرية للجرائم وأركان الجرئم . |
| | 2. تمكين الطالب من معرفة أقسام الجرائم. |
| | تمكين الطالب من معرفة قانون المحكمة الجنائية العراقية العليا لسنة 2005. |
| | 4. فهم تشكيل المحكمة الجنائية العراقية العليا لسنة 2005 والتعرف على تشكيل المحكمة أجراءات التقاضي امام المحكمة. |
| Module Learning | يتعلم الطالب أنواع الجرائم الدولية على وفق النظام الاساسي للمحكمة الجنائية الدولية. |
| Outcomes | معرفة الطالب بألاثار النفسية والاجتماعية لجرائم نظام البعث. |
| مخرجات التعلم للمادة الدراسية | يتمكن الطالب من فهم موقف النظام البعثي من الدين من خلال فهم عقيدة النظام السياسي سبيلاً لفهم موقف النظام من الدين. |
| | 8. يتمكن الطالب من التعرف على صور أنتهاكات القوانين العراقية وأنتهاكات حقوق الانسان وجرائم السلطة. |
| | 9. تمكين الطالب من التعرف على بعض قرارات الانتهاكات السياسية والعسكرية لنظام البعث. |
| | 10- يتعرف الطالب على أماكن السجون والاحتجاز لنظام البعث. 11- معرفة الطالب بالجرائم البيئية وبأثار الجرائم البيئية لنظام البعث، ويتعرف جرائم المقابر الجماعية. |
| | الجزء الاول: جرائم نظام البعث وفق قاتون المحكمة الجنائية العراقية العليا لعام 2005م، والجرائم النفسية والاجتماعية وآثارها وابرز انتهاكات النظام البعثي في العراق: |
| | التعريف بالجريمة لغة وأصطلاحاً، اركان واقسام الجريمة (2 ساعة). جرائم نظام البعث وفق قانون |
| | المحكمة الجنائية العراقية العليا عام 2005م: أنواع الجرائم الدولية، القرارات الصادرة من المحكمة الجنائية العليا (2 ساعة). وابرز القضايا التي نظرت فيها المحكمة (2 ساعة).الجرائم النفسية والاجتماعية وآثارها |
| | وابرز انتهاكات النظام البعثي في العراق: الجرائم النفسية، اليات الجرائم النفسية (2 ساعة). اثار الجرائم |
| Indicative Contents | النفسية ، الجرائم الاجتماعية (2 ساعة) . عسكرة المجتمع، موقف النظام البعثي من الدين (2 ساعة) . أنتهاكات القوانين العراقية، صور أنتهاكات حقوق الانسان (2 ساعة) . جرائم السلطة، بعض قرارات |
| المحتويات | الانتهاكات السياسية والعسكرية لنظام البعث، أماكن السجون والاحتجاز لنظام البعث (2 ساعة). |
| الإرشادية | الجزء الثاني: الجرائم البيئية لنظام البعث في العراق، جرائم المقابر الجماعية: الجرائم البيئية لنظام البعث في العراق: التلوث الحربي والإشعاعي – أستعمال الاسلحة المحرمة دولياً |
| | ومخاطر الالغام. (2 ساعة). التلوث بالمواد المشعة، أثار أستخدام الاسلحة المحرمة دولياً (2 ساعة). تدمير المدن والقرى (سياسة الارض المحروقة): قصف المدن، قصف العتبات المقدسة والمساجد |
| | والحسينيات، معركة نهر جاسم ، حرق آبار النفط (2 ساعة). تجفيف الأهوارو أثارها البيئية والاجتماعية |
| | والاقتصادية (2 ساعة). ، تجريف بساتين النخيل والاشجار والمزروعات (2 ساعة). جرائم المقابر الجماعية وموقف الامم المتحدة منها (2 ساعة). احداث المقابر الجماعية المرتكبة من النظام البعثي في |
| | العراق، التصنيف الزمني لمقابر ابادة الجماعية في العراق للمدة 1963- 2003 (2 ساعة). |
| | Learning and Teaching Strategies استراتيجيات التعلم والتعليم |
| | 1- زيادة وعي الطالب بالجرائم التي ارتكبها نظام البعث في العراق وحقيقة ما جرى من مآسي وويلات بحق |
| Strategies | الشعب العراقي. 2- اكتساب الطالب ثقافة عامة بماهية الجرائم واركانها واقسامها وموقف المشرع العراقي منها. |
| Su atogios | 2- المسلوب المعالب بموقف القانون الدولي والمحاكم الجنائية الدولية من الجرائم والانتهاكات التي ترتكبها الانظمة السلطوية. |

Student Workload (SWL)
الحمل الدراسي للطالب محسوب لـ ١٥ أسبوعا

| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | 33 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 2.2 |
|--|----|---|-----|
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال | 17 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 1.1 |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 50 | |

| Module Evaluation تقييم المادة الدر اسية | | | | | | |
|---|----------------------|-----------------|------------------|------------|---------------------------|--|
| | | Time/Nu mber | Weight (Marks) | Week Due | Relevant Learning Outcome | |
| | Quizzes | 2 | 10% (10) | 5, 10 | LO #1, 2, 10 and 11 | |
| Formative | Assignments | 2 | 10% (10) | 2, 12 | LO # 3, 4, 6 and 7 | |
| assessment | Projects / tutorial. | 1 | 10% (10) | Continuous | | |
| | Report | 1 | 10% (10) | 13 | LO # 5, 8 and 10 | |
| Summative | Midterm Exam | 1 hr | 10% (10) | 7 | LO # 1-7 | |
| assessment | Final Exam | 2hr | 50% (50) | 16 | All | |
| Total assessment 100% | | | 100% (100 Marks) | | | |

| | Delivery Plan (Weekly Syllabus) |
|---------|---|
| | المنهاج الاسبوعي النظري |
| | Material Covered |
| Week 1 | محاضرة تعريفية عن المادة واهميتها. |
| Week 2 | التعريف بالجريمة لغة واصطلاحا، أقسام الجريمة، جرائم نظام البعث وفق قانون المحكمة الجنائية العراقية العليا عام 2005م، أنواع الجرائم الدولية. |
| Week 3 | القرارات الصادرة من المحكمة الجنائية العليا، وأبرز القضايا التي نظرت فيها المحكمة. |
| Week 4 | الجرائم النفسية، اليات الجرائم النفسية. |
| Week5 | اثار الجرائم النفسية، الجرائم الاجتماعية |
| Week 6 | عسكرة المجتمع، موقف النظام البعثي من الدين. |
| Week 7 | انتهاكات القوانين العراقية، صور انتهاكات حقوق الانسان، جرائم السلطة. |
| Week 8 | بعض قرارات الانتهاكات السياسية والعسكرية لنظام البعث، أماكن السجون والاحتجاز لنظام البعث. |
| Week 9 | الجرائم البيئية لنظام البعث في العراق: التلوث الحربي والاشعاعي – استعمال الاسلحة المحرمة دولياً ومخاطر الالغام. |
| Week 10 | التلوث بالمواد المشعة، أثار أستخدام الاسلحة المحرمة دولياً |
| Week 11 | تدمير المدن والقرى (سياسة الارض المحروقة). |
| Week 12 | تجفيف الأهوار أثارها البيئية والاجتماعية والاقتصادية . |

| Week 13 | تجريف بساتين النخيل والاشجار والمزروعات. |
|---------|---|
| Week 14 | جرائم المقابر الجماعية، أحداث المقابر الجماعية المرتكبة من النظام البعثي في العراق. |
| Week 15 | التصنيف الزمني لمقابر الابادة الجماعية في العراق للمدة 19632003 |
| Week 16 | الامتحان النهائي |

Learning and Teaching Resources

مصادر التعلم والتدريس

| | Text | Available in the Library? |
|----------------------|---|---------------------------|
| Required Texts | المنهج المقر الدراسي للجامعات الحكومية و الأهلية كافة كتاب وزارة التعليم والبحث العلمي ذي العدد (ت م 3/ 7588 في 2023/10/19) | نعم |
| Recommended Texts | | У |
| Websites | | |

SECOND YEAR (SEMESTER 4)





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|----------------|----------------------------|----------------------------|--------------|--------------------|---------|
| Module Title | Engineering | Engineering Mathematics II | | | ıle Delivery | |
| Module Type | Core | Core | | | ⊠ Theory | |
| Module Code | CHE 207 | | | | ⊠ Lecture □ Lab | |
| ECTS Credits | 4 | 4 | | | ☐ Practical | |
| SWL (hr/sem) | 100 | | | | ☐ Seminar | |
| Module Level | | UGII | Semester of Delivery Three | | Three | |
| Administering De | partment | Chemical engineering | College | College | of Engineering | |
| Module Leader | Ali I. Abdalla | 1 | e-mail | alialnua | immy@uodiyala | .edu.iq |
| Module Leader's Acad. Title Lecturer | | Module Le | ader's Q | ualification | MSc. | |
| Module Tutor | | | e-mail | | | |
| Peer Reviewer Name | | | e-mail | | | |
| Scientific Committee Approval Date | | | Version Nu | ımber | 1.0 | |

| | Relation with other Modules | | | | |
|-----------------------------------|-----------------------------|----------|-----|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | | |
| Prerequisite module | CHE 201 | Semester | Two | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | | |
|--|---|--|--|--|
| | أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | |
| Module Objectives أهداف المادة الدر اسية | 1-Provide a Strong Foundation in Mathematical Concepts Introduce fundamentals of calculus, linear algebra, and other core mathematical tools Develop logical understanding and mathematical skills to solve engineering problems 2- Develop Problem-Solving Skills Equip students with techniques to interpret objectives, mathematize, solve, and communicate results Cultivate critical thinking and creativity in applying mathematics to engineering problems 3- Facilitate Active Learning Use a mix of lectures, tutorials, discussions, and hands-on exercises for active learning Provide guided practice with relevant engineering problems 4- Teach Applications to Engineering Fields Relate mathematical concepts to real-world engineering scenarios and applications Integrate engineering examples and case studies throughout the curriculum Demonstrate the importance and symbiosis between mathematics and engineering The overall goal is to develop students' mathematical maturity and ability to apply advanced concepts to solve complex engineering problems. This requires a combination of solid foundational knowledge, practice with relevant applications, and explicit connections between theory and practice. | | | |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | Module Learning Outcomes 1. Demonstrate an understanding the Double integral, changing double integral from Cartesian to polar form. 2. Finding the area using double integral, area using polar coordinates, moments and center of mass 3. Demonstrate an understanding the triple integral, volume using triple integral 4. Demonstrate an understanding of infinite sequence and series, geometric series, alternating series, tests of infinite series. 5-Representing function as series, Taylor series, McLaurin series, power series, Taylor's theorem with remainder 6- Types of matrix, matrix algebra, Inverse of matrix, Grammar's rule, Gaussian elimination method. 7- Understanding and finding Eigen value and Eigen vector of a matrix | | | |
| Course Description | Course Description Engineering Mathematics I is a foundational course that introduces students to the key mathematical concepts and techniques essential for success in various engineering disciplines. The course typically covers the following topics: 1- Multiple Integration Calculus • Double integral, changing double integral from Cartesian to polar form • Applications of double integration (Area, Volume, mass, center of mass, | | | |

average value, etc.)

• Triple integration and its applications

2-Linear Algebra

- Systems of linear equations
- Gaussian elimination
- Matrix algebra
- Transpose
- Determinants and Cramer's Rule
- Inverse matrices
- Eigen value and Eigen vector of matrix

3- Sequence and Series

- Convergence and Divergence of Series (Tests for convergence: ratio test, root test, integral test, comparison test, etc.)
- Geometric series, Harmonic series, Alternating series, Power series, Taylor Series, Fourier series.
- Radius and interval of convergence
- Approximation of functions

The course is typically delivered through a combination of lectures, tutorials, classwork, and homework. Assessment includes a final exam worth 50%, mid-term exam worth 10% and other coursework such as homework assignments, classwork and tests worth 40%.

Learning and Teaching Strategies

استراتيجيات التعلم والتعليم

1- Integrate Lectures and Tutorials for Active Learning

- Mix direct instruction in lectures with guided practice in tutorials for active learning
- Include a variety of learning activities like problem-solving, discussions, and hands-on exercises

2- Use Relevant Examples and Applications

- Use examples and problems with engineering contexts embedded throughout the course
- Include some engineering guest lectures to explain real-world applications of the mathematical concepts
- Relate the mathematics to specific engineering disciplines like fluid mechanics, structures, etc

3- Establish High Expectations and Develop Critical Thinking

- Establish high expectations for students' mathematical understanding
- Develop critical thinking and problem-solving skills
- Cultivate creativity in applying mathematics to engineering problems

The overall goal is to help students develop a strong conceptual understanding of the mathematics that is directly applicable to their engineering studies. This requires a combination of solid foundational knowledge, practice with relevant

Strategies

problems, and explicit connections to engineering applications. Both students and instructors play an active role in employing effective learning and teaching strategies to achieve this goal.

| | Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | |
|---|--|---|------|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures 41 In class tests 4 Tutorial 15 Final Exam 3 | 63 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 4.2 |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Library, dorm, home memorizing 12 Preparation for tests 13 Homework 12 | 37 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2.47 |
| Total SWL (h/sem) 100 الحمل الدر اسي الكلي للطالب خلال الفصل | | | |

| | Module Evaluation | | | | | |
|----------------------|-----------------------|---------------------|----------------|----------|---------------------------|--|
| | تقييم المادة الدراسية | | | | | |
| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome | |
| | Quizzes | 2 | 20% (20) | 3 and 10 | LO #1, 2, 3, 4,5 and 7 | |
| Formative assessment | Assignments | 4 | 10% (10) | 5 and 11 | LO #1, 2, 3, 4,5 and 7 | |
| | homework | 4 | 10%(10) | 4 and 12 | LO #1, 2, 3, 4,5 and 7 | |
| Summative | Midterm Exam | 2 hr | 10% (10) | 8 | LO #1 - #4 | |
| assessment | Final Exam | 3 hr | 50 % (40) | 16 | All | |
| Total assessment | | 100% (100 Marks) | | | | |

| Delivery Plan (Weekly Syllabus) | | |
|---------------------------------|--|--|
| المنهاج الاسبوعي النظري | | |
| | Material Covered | |
| Week 1 | Double integral, changing double integral from Cartesian to polar form | |
| Week 2 | Area using double integral, area using polar coordinates | |
| Week 3 | Moments and center of mass | |

| Week 4 | Triple integral, volume using triple integral |
|---------|--|
| Week 5 | Introduction, infinite series, geometric series |
| Week 6 | Tests of infinite series (Integral test, comparison test, ratio test, root test) |
| Week 7 | Alternating series, tests of alternating series |
| Week 8 | Midterm Exam |
| Week 9 | Taylor series, McLaurin series |
| Week 10 | power series |
| Week 11 | Taylor`s theorem with remainder |
| Week 12 | introduction, types of matrix, matrix addition, subtraction and multiplication |
| Week 13 | Determinant, Grammar's rule |
| Week 14 | Inverse of matrix, solution of system of linear equations by matrix |
| Week 15 | Eigen value and Eigen vector |
| Week 16 | Final Exam |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|---|---------------------------|--|--|
| | Text | Available in the Library? | | |
| Required Texts | 3. George B. Thomas and Ross L. Finney, "Calculus and Analytic Geometry, Addison- Wesley, ISBN:0201531747. 4. Erwin Kreyszig, Advanced Engineering Mathematics, John Wiley | Yes | | |
| Recommended Texts | George F. Simmons, "Calculus with Analytic Geometry", McGraw-Hill, ISBN: 0070576424 | yes | | |
| Websites | https://www.geogebra.org/calculator https://www.desmos.com/calculator https://www.symbolab.com/solver/integral-calculator | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | | |
|---|--------------|----------------------|------------|-----------------------------------|------------------------|---------------|--|
| Module Title | Fluid Flow I | I | | Modu | ıle Delivery | | |
| Module Type | Core | | | | ☑ Theory | | |
| Module Code | CHE 208 | | | ☑ Lecture | | | |
| ECTS Credits | 5 | | | | - ⊠ Lab □ Practical | | |
| SWL (hr/sem) | 125 | | | ☐ Seminar | | | |
| Module Level | | 2 | Semester o | of Delivery 2 | | 2 | |
| Administering De | partment | Chemical engineering | College | College of Engineering | | | |
| Module Leader | Walaa abid m | nahmood | e-mail | Whalaa_alkhaisi76@uodiyala.edu.iq | | liyala.edu.iq | |
| Module Leader's | Acad. Title | Assistant Lecturer | Module Le | ule Leader's Qualification MSo | | MSc. | |
| Module Tutor | | | e-mail | | | | |
| Peer Reviewer Name | | | e-mail | | | | |
| Scientific Committee Approval Date | | | Version Nu | ımber | 1.0 | | |

| Relation with other Modules | | | | | |
|-----------------------------------|------|----------|--|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | | |
| Prerequisite module | None | Semester | | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | | | |
|--|--|--|--|--|--|
| | أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | |
| Module Objectives | This semester is a continuation of the first semester, where the student | | | | |
| أهداف المادة الدر اسية | understands the applications of the Bernoulli and Venturi equations in the | | | | |

| | operation of pumps |
|-------------------------------|--|
| | As well as learning about the types of Newtonian fluids and how to deal with |
| | them, and also learning about the devices for the flow meter. |
| | Compressed fluids and their applications are also studied. |
| | The student must be able to work with an integrated team and be able to use the skills he has acquired and apply them in the field of work. The student should be able to apply Bernoulli's equation in his practical |
| Module Learning | reality to control the flow under the conditions imposed on him |
| Outcomes | Student distinguish between Newtonians fluid and Non-Newtonians fluid |
| مخرجات التعلم للمادة الدراسية | The student should distinguish between the different methods of dealing |
| | with fluids. Acquiring the skill of studying the types of flow by studying |
| | compressible and non-compressive fluids. • |
| Indicative Contents | Introducing the student to the basic properties of fluids and the phenomena related to them |
| المحتويات الإرشادية | The nature and behavior of fluids at rest |
| . 3, .3 | The nature of fluids in the state of flow of all types and the forces exchanged between them and the surfaces containing them |
| | Designing pipe connection systems, starting from the process of pumping with pumps and connecting to pipes and their connectors, to connecting to various places. |
| | Introducing the student to the basic principles of compressive flow to enable him to design nozzles and their types |
| | Familiarize the student with the basics of mixing fluids to enable him to know the properties of the mixture. |

| Learning and Teaching Strategies | | | | | | |
|----------------------------------|--|--|--|--|--|--|
| | استراتيجيات التعلم والتعليم | | | | | |
| Strategies | Daily discussion to determine the extent of students' understanding of the material and to evaluate the daily contributions. Daily exams with various short scientific questions to understand the extent of their understanding of the material. | | | | | |
| | Giving part of each chapter's grade to homework assignments. Daily exams (quizzes) and monthly exams for the curriculum and the final exam. | | | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | |
|--|---|---|-----|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 4 | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل | 121 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 5.4 | |
| Total SWL (h/sem) الحمل الدر اسي الكلي للطالب خلال الفصل | | | | |

| Module Evaluation | | | | | | | |
|-------------------|-----------------------|-------------|----------------|---------|-------------------|--|--|
| | تقييم المادة الدراسية | | | | | | |
| | | Time/Number | Weight (Marks) | Week | Relevant Learning | | |
| | | | | Due | Outcome | | |
| | Quizzes | 2 | 10% (10) | 7 and 9 | LO #2 and #5 | | |
| Formative | Assignments 2 | 2 | 20% (20) | 11 and | LO #4 and #5 | | |
| assessment | | _ | | 13 | LO 114 and #3 | | |
| assessment | Projects / Lab. | 1 | 10% (10) | | | | |
| | Report | | | | | | |
| Summative | Midterm Exam | 2 hr | 10% (10) | 7and 9 | LO #1 - #3 | | |
| assessment | Final Exam | 3 hr | 50% (50) | 16 | All | | |
| Total assessm | Total assessment | | 100% (100 | | | | |
| Total assessment | | | Marks) | | | | |

| Delivery Plan (Weekly Syllabus) | | | |
|---------------------------------|---|--|--|
| | المنهاج الاسبوعي النظري | | |
| | Material Covered | | |
| Week 1 | Flow Measurement (Practical Application of Bernoulli's Equation) Pitot tubes, orifice meter, | | |
| Week 2 | Venturi meter with its applications | | |
| Week 3 | Nozzle meter, Rotameters other types of flow meters, Flow in open channels and weirs. | | |
| Week 4 | Tutorial, Midterm one | | |
| Week 5 | Pumping of Liquids Total heads, NPSH, Horse Power and cost consumption, Pumping Efficiencies, | | |
| Week 6 | Types of the pumps, Selection of Pumps, Centrifugal pump relations, homologous centrifugal | | |
| vveek 6 | pump Characteristics curves, Centrifugal pumps in series and in parallel. | | |
| Week 7 | Tutorial +Quiz | | |
| Week 8 | Non-Newtonian Fluids in Pipes Definition, types of fluid depended on time, | | |
| Week 9 | Calculation of friction and pressure drop for general time independent in laminar and turbulent flow. +Quiz | | |
| Week 10 | Flow of Compressible Fluid General equation, equation of state, sonic velocity in fluids, , | | |
| Week 10 | Converging-diverging nozzle for gas flow,. | | |
| Week 11 | Isothermal, Non-isothermal and Adiabatic flow of an ideal gas in horizontal pipe | | |
| Week 12 | Type of Compressor, Gas compression and compressors work and efficiency. Two Phase Flow, | | |
| Week 12 | Flow regime, Pressure Momentum, and Energy relations | | |
| Week 13 | Liquid Mixing Stirring and mixing, Stirred vessels (power consumption, power curve, scaledup), | | |
| Week 13 | equipment. Motion of Particles in a Fluid Drag force on a particle, terminal falling velocities , | | |
| Week 14 | Sedimentation of fine and coarse Particles. Flow of Fluid through Granular Bed | | |
| Week 15 | Midterm Exam. | | |
| | | | |

| Week 16 | Preparatory week before the final Exam. |
|---------|---|
| Week 16 | Preparatory week before the final Exam. |

| Delivery Plan (Weekly Lab. Syllabus) | | | | |
|--------------------------------------|--|--|--|--|
| المنهاج الاسبوعي للمختبر | | | | |
| | Material Covered | | | |
| Week 1 | Experiment #1: OSBORNE REYNOLDS' DEMONSTRATION | | | |
| Week 2 | EXPERIMENT #2: BERNOULLI'S THEOREM DEMONSTRATION | | | |
| Week 3 | Experiment #3: Orifice experiment | | | |
| Week 4 | Experiment #4 Venturi meter | | | |
| Week 5 | Daily Exam and discussion students | | | |
| Week 6 | E EXPERIMENT #5: ENERGY LOSS IN PIPE FITTINGS | | | |
| Week 7 | EXPERIMENT #6: HYDROSTATIC PRESSURE | | | |
| Week 8 | Preparing lab project | | | |
| Week 9 | Experiment #7:Calibration | | | |
| Week 10 | Experiment #8: IMPACT OF A JET | | | |
| Week 11 | EXPERIMENT #9: ENERGY LOSS IN PIPES | | | |
| Week 12 | Daily Exam and dissection students | | | |
| Week 13 | Experiment #10: FLOW OVER WEIRS | | | |
| Week 14 | EXPERIMENT #11: PUMPS | | | |
| Week 15 | lab. Projects. | | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|---|---------------------------|--|--|
| | Text | Available in the Library? | | |
| Required Texts | R. W. Fox and A. T. McDonald, "Introduction to fluid mechanics", 5th. Ed., | Yes | | |
| Recommended Texts | John Wiley & Sons, 1998. 2. R. M. Holland, "Fluid Flow", John Wiley & Sons, 1982. W. L. McCabe, J. Smith and P. Harriot, "Unit Operations of Chemical Engineering", 6th Ed., McGraw – Hill, International Edition, 2001 | yes | | |
| Websites | http://www.efluids.com | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|--------------------------------|-----------------------|-------------------------|-----------------|-------------|-------|
| Module Title | Pollution | and Industrial Safety | | Modi | ule Deliver | y |
| Module Type | Core | | | | Theory | |
| Module Code | CHE 209 | | | | Lecture | |
| ECTS Credits | 5 | 5 Tutorial Practical | | | | |
| SWL (hr/sem) | 125 | 125 Seminar | | | | |
| Module Level | | 2 | Semester | r (s) offered 2 | | 2 |
| Administering Departme | nt | Chemical Engineering | College | Engineering | | |
| Module Leader | Yaser I. Jase | em | e-mail | Yaser_ij(| @uodiyala.e | du.iq |
| Module Leader's Acad. Title | | Assist. Proff. | Module Lo Qualificat | | | MS.c. |
| Module Tutor | None | | e-mail | | | |
| Peer Reviewer Name | Peer Reviewer Name None e-mail | | | | | |
| Review Committee Appr | oval | | Version N | umber | 1.0 | |

| Relation With Other Modules العلاقة مع المواد الدراسية الأخرى | | | | | |
|--|--|----------|---|--|--|
| Prerequisite module | None | Semester | - | | |
| Co-requisites module | None | Semester | - | | |
| | Module Aims, Learning Outcomes, Indicative Contents and Brief Description | | | | |
| خنصر | أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية مع وصف مختصر | | | | |
| Module Aims | a) Introducing pollutants in general, whether they are air, water, or so | | | | |
| pollutants, especially those resulting from various industries. | | | | | |
| اهداف المداد الدر اسيد | b) Introducing the harms resulting from these pollutants and their negative | | | | |
| | impact on human life and the surrounding soil, air, water, and all living organisms. | | | | |

| | c) Thus, finding the best appropriate methods (taking into account Cost factor) to | |
|-------------------------------|---|--|
| | | |
| | reduce these pollutants or reduce their impact through designing the necessary | |
| | equipment to reduce these pollutants or through the correct management of | |
| | various wastes in a way that ensures the best environment with the least amount | |
| | of pollutants possible to preserve life. | |
| | d) Introducing food fortification technology and fortification requirements. | |
| | e) Ensuring the necessary protection for all production teams and reducing risk | |
| | factors, especially with regard to the human factor. | |
| | f) Reducing the incidence of occupational diseases such as poisoning and others. | |
| | g) Reducing the possibility of injuries and accidents at work. | |
| | h) Preventing and extinguishing fires and searching for preventive factors. | |
| | At the end of this course the students will be able to: | |
| | 1- Knowledge of environmental phenomena that have a global impact, such as | |
| | global warming and the erosion of the ozone layer. | |
| | 2- Knowing how to classify different pollutants, whether they are air, water, or | |
| | solid materials. | |
| | 3- Designing various equipment to reduce pollutants in all their forms. | |
| Module Learning | 4- Optimal management of solid waste, ensuring the reduction of waste of raw | |
| Outcomes | materials and reducing damage to the minimum possible extent. | |
| | 5- Taking care of the environment of which we are a part and spreading | |
| مخرجات التعلم للمادة الدراسية | environmental awareness. | |
| | 6- Enables the student to ensure the safety of workers and the facility in the event | |
| | of an accident or fire. | |
| | 7- Reduce the factors causing accidents and achieve the highest level of safety A | |
| | person in the establishment, | |
| | 8- As well as limiting the consequences of fires as much as possible and | |
| | minimizing losses. | |
| | Indicative content includes the following. | |
| | Identifying global environmental phenomena, their effects and | |
| | management. [4 hrs] | |
| Indicative Contents | Types of Hazards. [6 hrs] Airmallation airmallation plansification [0 haz] | |
| المحتويات الإرشادية | Air pollution, air pollution classification. [9 hrs] The risks of pressure, heat, radiation, fire and explosion. [6 hrs] | |
| المحلويات الإرسادية | Water and wastewater treatment. [9 hrs] | |
| | Working at heights and maintenance. [6 hrs] | |
| | Sold waste management. [4 hrs] | |
| | Management of drilling and cutting fluids for wells. [4 hrs] | |
| | Identifying global environmental phenomena, their effects, and ways to reduce | |
| | them and reduce their effects. Air pollution, air pollution classification, | |
| | Gravitational settling chambers, Cyclone separators, Baghouses, Electrostatic | |
| | precipitator, Wet scrubber and their calculations. Water and wastewater | |
| | treatment, Primary and Secondary treatment. Tertiary treatment. Solid Waste | |
| Course Description | Management. | |
| | The safety training course trains the youth is various sectors such as construction, | |
| | infrastructure, fire, industrial hazards, and various risk assessment safety | |
| | measures. This provides a detailed overview and empowers the youth into a | |
| | skilled individual helping them meet the demands of a career in the following field. | |
| | Learning and Teaching Strategies | |
| | | |
| | استراتيجيات التعلم والتعليم | |
| Strategies | Begin to establish a strong conceptual understanding of the environmental issues | |
| Ü | and Safety training. Use the reality of work examples to help students relate | |

abstract concepts to environmental issues. Encourage discussions and questions to clear up any misconceptions. In addition, provide students with the tool to deal with environmental issues and Safety problems. Encourage active participation and group discussions to enhance critical thinking and problem-solving skills. Guide students through the problem-solving process and provide constructive feedback.

| Student Workload (SWL) الحمل الدراسي للطالب | | | | | |
|---|-----|---|-----|--|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم الطالب خلال الفصل In class lectures 38 Tutorial 2 In class tests 4 Final Exam 4 | 48 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 3.2 | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم الطالب خلال الفصل Library, dorm, home memorizing 50 Preparation for tests 14 Homework 13 | 77 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 5.1 | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | 125 | | | | |

| Module Evaluation تقييم المادة الدر اسية | | | | | | |
|---|---|---|----------|----------------|---|--|
| | Time/Nu Weight (Marks) Week Due Outcome | | | | | |
| Formative | Quizzes | 4 | 15% (15) | 2, 5, 8 and 11 | LO #1, 2, 3,7,8 and LO# 4,5,6,9,10 | |
| assessment | Assignments | 4 | 5% (5) | 3, 6, 9 and 12 | LO # 7, 8 and LO# 9,10 | |
| | Homework | 2 | 20% (20) | Continuous | | |
| Summative assessment | Midterm Exam | 2 | 10% (10) | 7 | LO #1, 2, 3,4,7,8 and LO# 4,5,6,9,10 | |
| assessment | Final Exam | 3 | 50% (50) | 16 | All | |
| Total assessr | Total assessment 100% (100 Marks) | | | | | |

| | Delivery Plan (Weekly Syllabus) المنهاج الاسبوعي النظري | | |
|-------------|--|--|--|
| | Material Covered | | |
| Week 1 | Identifying global environmental phenomena, their effects, and ways to reduce them and reduce their effects. | | |
| Week 2&3 | Air pollution, air pollution classification (primary, secondary and air borne particulate), Gravitational settling chambers. | | |

| Week 4 | Air pollution, cyclones, wet scrubbers. |
|---------|---|
| Week 5 | Water and wastewater treatment, types of pollutant their sources and effect, primary treatment. |
| Week 6 | Water and wastewater treatment, secondary treatment. |
| Week 7 | Water and wastewater treatment, tertiary treatment. |
| Week 7 | water and wastewater treatment, tertiary treatment. |
| Week 8 | Solids waste management. |
| Week 9 | Types of Hazards Definitions of Hazards Visible, Hidden and Developed Hazards with Examples Hazard Awareness |
| Week 10 | Understanding Hazardous Energy Forms of Energy: Sources and Developments How Energy Becomes Hazardous Controlling Hazardous Energy: Procedures Personal Protective Equipment (PPE): Eye and Face Protection (Common Eye and Face Hazards), Head Protection (Hard Hats: Types and Quality), Foot and Leg Protection, Hand Protection, Body Protection, Hearing Protection (Permissible Noise Exposures), Respiratory Protective Equipment (RPE), and Fall Protection (Training and Requirements) |
| Week 11 | General Fire and Explosion Considerations Fire Triangle: Diagram Elements of Fire: Energy, Oxygen and Fuel Critical Risk Factors: Chemical and Physical Factors Control for Elements of Fire: Energy, Oxygen and Fuel Risk Assessment: Conditions, Level of Risk, and Procedures of Treatment Required Fire and Explosion Prevention Supervisor Roles: Training, Preparing Hazard ID and Controls, Reporting and Evaluating Unsafe Works Worker Roles Training Requirements Fire Classes Safety Measures Types of Fire Tenders Basic Precautions for Fire Prevention Oil & Gas Well Drilling Risks Hot Work, Fire, and Explosive Hazards: Possible Solutions |
| Week 12 | Pressure, Temperature and Radiation Hazards Examples of Pressure Hazards Includes Using Compressed Air for Cleaning Tools Temperature Hazards Radiation Hazards: Types of Radiation Norm in Oil Industry Norm Regulations |
| Week 13 | Maintenance Definitions and Descriptions Maintenance Hazards Basic Rules for Safe Maintenance |
| Week 14 | Working area and height hazard. |
| Week 15 | Health, Safety, Environment and Waste Management of Drilling Fluids |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|--|------------------------------|--|--|
| | Text | Available in the Library? | | |
| Required Texts | C.S.Rao , "Environmental Pollution Control Engineering", 2nd eddition , New Age International(P) Limited, Published, 2006, Reprint 2007. M. Grawford, "Air Pollution Control Theory", McGraw-Hill, New York, 1976. M. M. Gilbert, "Introduction To Environmental Engineering And Science", 2nd edition, Hall, Inc, New Jersey, 1998. Introduction to Industrial and Systems Engineering (Prentice-Hall International Series in Industrial and Systems) 2nd Edition Wayne C. Turner (Author) Prentice Hall International Series in Industrial & Systems Engineering: Systems Engineering and Analysis, Benjamin Blanchard | Yes | | |
| Recommended Texts | 1. R. K. Sinnott, Chemical Engineering Design, Vol. 6. 4 th edition, Chemical Engineering Design, 2005, pp. 450-457. | Yes | | |
| Websites | | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|-------------|-------------------------|------------|----------------------|----------|--|
| Module Title | Engineering | Statistic | | Module Delivery | | |
| Module Type | Basic | | | Theory | Theory | |
| Module Code | CHE 210 | | | ✓ Lecture Lab | | |
| ECTS Credits | 3 | | | ✓ Tutorial | | |
| SWL (hr/sem) | 75 | 75 Seminar | | | _ | |
| Module Level 2 | | 2 | Semester o | ester of Delivery 2 | | |
| Administering Denartment | | Chemical Engineering | College | Engineering | | |
| Module Leader | Sura Fahmy | Yousif | e-mail | Sura.fahmy@uodiyal | a.edu.iq | |
| Module Leader's Acad. Title Assistant Prof. | | Assistant Prof. | Module Le | ader's Qualification | MSc | |
| Module Tutor | r None | | e-mail | 1 | | |
| Peer Reviewer Name None | | e-mail | - | | | |
| Scientific Committee Approval Date | | | Version Nu | umber 1.0 | | |

| Relation with other Modules | | | | |
|-----------------------------------|------|----------|---|--|
| العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | None | Semester | - | |
| Co-requisites module | None | Semester | - | |

| Module Aims, Learning Outcomes and Indicative Contents | | |
|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | |
| Module Objectives | 6. Giving the student an overview of the principles and concepts of statistics | |

| أهداف المادة الدر اسية | and probability |
|---|--|
| العدادة | and probability. Make the student able to represent data in the form of tables and graphics. Teach the student to differentiate between quantitative and qualitative data and how they are represented. Knowing the types of descriptive measures such as arithmetic mean, frequency, range, amount of change and standard deviation. Knowledge of the principles of probability and its types and the laws of multiplication, addition and continuity in addition to the use of the laws of permutations and combinations in finding probability. Make the student able to know the types of random variables. Make the student able to know the probability distribution and types of discrete random variable. Make the student able to know the probability distribution and types of continuous random variable. Make the student able to know the difference between population proportion and sample proportion. Teach the student the properties and conditions of central limit theorem. |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | Enabling the student to differentiate between quantitative and qualitative data. Enabling the student to differentiate between the methods of representing quantitative and qualitative data, whether in the form of tables or graphics. Enabling the student of finding the types of descriptive metrics. Enabling the student of finding the probability for each random variable. Enabling the student of finding the types of distribution for the discrete and continuous random variables. Enabling the student of finding the population proportion and sample proportion. Enabling the student of finding the shape of the sampling distribution of the sample mean and the central limit theorem. Enabling the student of knowing the methods of curve fitting and its applications. |
| Indicative Contents المحتويات الإرشادية | Indicative content includes the following: Principles of statistics, quantitative and qualitative data (Types of quantitative and qualitative data and the way to represent them numerically and graphically) [7 hrs]. Descriptive metrics (Descriptive measures such as mean, frequency, median value, amount of change, standard deviation, and range) [7]. Probability (Introduction to probability and its types and the laws of multiplication and addition using Venn forms) [7]. Permutations and combinations (Use permutations and combinations theorems to find probability values) [4]. Types of probability distribution (Introduction to the types of probability distribution, such as normal distribution, Poisson distribution, Binomial distribution, Standard distribution) [4]. Central limit theorem and Curve fitting methods [4]. |
| Course Description | This course provides students an idea of basic principles of statistic and probability, types of data, represent qualitative and quantitative data in tables |

and graphs, descriptive measures for grouped and ungrouped data, mean, median, mode, range, variance, standard deviation, percentiles, quartiles, deciles. Experiment, outcomes, sample space, types of events, tree diagram, types of probability, how to use permutations and combinations in probability problems. Types and probability distribution of discrete and continuous random variables. Sampling distribution of sample mean, sampling distribution of the population proportion, sampling distribution of the sample proportion, central limit theorem and ways of curve fitting.

| Learning and Teaching Strategies | | | | | | |
|----------------------------------|--|--|--|--|--|--|
| | استراتيجيات التعلم والتعليم | | | | | |
| Strategies | Explaining the theoretical lectures by subject teachers using modern teaching and presentation methods. Explaining the philosophy of statistical laws and the purpose of their use. Practical application of solving statistical problems using applied examples and solving them. Encouraging students by collecting statistical data and preparing individual and group reports, as well as preparing presentations. Encouraging students to contribute in discussions and questions to correct misconceptions. Giving exercises that require analysis and conclusions related to subject lectures. Proper assessment and evaluative feedback. Summarizing and reviewing of lectures. | | | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | |
|--|----|---|-----|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم الطالب خلال الفصل In class lectures 25 In class tests 5 Final Exam 3 | 33 | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 2.2 | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Libraray, dorm, home, memorize 20 Preparation for tests 14 Homework 8 | 42 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2.8 | |
| Total SWL (h/sem) 75 الحمل الدر اسي الكلي للطالب خلال الفصل | | | | |

| Module Evaluation تقييم المادة الدر اسية | | | | | |
|---|-----------------|---------------------|----------------|---------------------------|---------------------------------------|
| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome |
| Formativo | Quizzes | 4 | 20% (20) | 3, 6, 9 and 12 | LO #1, 2, 3, 4, 5, 7, 8, 9, 10 and 11 |
| Formative assessment | Assignments | 6 | 10% (10) | 3, 5, 8, 11, 12 and 13 | LO #1, 4, 10 and 11 |
| | Report | 1 | 10% (10) | 13 | LO #3, #4 and #6 |
| Summative assessment | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - #8 |
| assessment | Final Exam | 3hr | 50% (50) | 16 | All |
| Total assessment | | 100% (100 Marks) | | | |

| Delivery Plan (Weekly Syllabus) | | | | | | |
|---------------------------------|--|--|--|--|--|--|
| | المنهاج الاسبوعي النظري | | | | | |
| | Material Covered | | | | | |
| Week 1 | Introduction to statistics | | | | | |
| Week 2 | Types of quantitative data and the way to represent them numerically and graphically | | | | | |
| Week 3 | Types of qualitative data and the way to represent them numerically and graphically | | | | | |
| Week 4 | Descriptive measures such as mean, frequency, and median value | | | | | |
| Week 5 | Descriptive measures such as amount of change, standard deviation, and range | | | | | |
| Week 6 | Introduction to probability and its types | | | | | |
| Week 7 | Introduction to probability and its laws of multiplication and addition using Venn forms | | | | | |
| Week 8 | Permutations and combinations | | | | | |
| Week 9 | Use permutations and combinations theorems to find probability values | | | | | |
| Week 10 | Introduction to the types of probability distribution such as Binomial distribution | | | | | |
| Week 11 | Introduction to the types of probability distribution such Poisson distribution | | | | | |
| Week 12 | Introduction to the types of probability distribution such as uniform distribution | | | | | |
| Week 13 | Introduction to the types of probability distribution such as normal distribution | | | | | |
| Week 14 | Introduction to the types of probability distribution such as standard normal distribution | | | | | |
| Week 15 | central limit theorem and curve fitting methods | | | | | |
| Week 16 | Final Exam | | | | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|---|-----|--|--|
| | Available in the Library? | | | |
| Required Texts | Douglus C. Montgomery, G. C. Runger, Applied Statistics and Probability for Engineers, John Wiley and Sons, 2003. | Yes | | |
| Recommended Texts | P. G. Hoel, S. C. Port and C. J. Stone, Introduction to Statistical Theory, Houghton Miffin, 1971. | Yes | | |
| Websites | | | | |

APPENDIX:

| GRADING SCHEME مخطط الدرجات | | | | | |
|--------------------------------|-------------------------|-------------|-----------|---------------------------------------|--|
| Group | Grade | التقدير | Marks (%) | Definition | |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | |
| g G | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors | |
| Success Group (50 - 100) | C - Good | جيد | 70 - 79 | Sound work with notable errors | |
| | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | |
| | | | | | |
| Note: | | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.





MODULE DESCRIPTION FORM

نموذج وصف المادة الدراسية

| Module Information معلومات المادة الدراسية | | | | | |
|---|---------------|-------------------------|-------------|----------------------------|-----|
| Module Title | Computer P | rogramming II | | Module Delivery | |
| Module Type | Core | | | Theory | |
| Module Code | CHE 206 | | | ✓ Lecture ✓ Lab | |
| ECTS Credits | 4 | | | Tutorial | |
| SWL (hr/sem) | 100 | | | ✓ Seminar | |
| Module Level | 2 | | Semester of | of Delivery 2 | |
| Administering De | epartment | Chemical Engineering | College | Engineering | |
| Module Leader | Sura Fahmy | Yousif | e-mail | Sura.fahmy@uodiyala.edu.iq | |
| Module Leader's | Acad. Title | Assistant Prof. | Module Le | ader's Qualification | MSc |
| Module Tutor | None | | e-mail | - | |
| Peer Reviewer Name None | | e-mail | - | | |
| Scientific Commi Date | ttee Approval | | Version Nu | umber 1.0 | |

| Relation with other Modules | | | | | |
|-----------------------------|-----------------------------------|----------|---|--|--|
| | العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | None | Semester | - | | |
| Co-requisites module | None | Semester | - | | |

Module Aims, Learning Outcomes and Indicative Contents

أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية

| Module Objectives أهداف المادة الدر اسية | Teaching the student during the semester about Simulink in MATLAB program specifications, algorithm design and development, and testing using a modern software development environment. Teaching the student how to design models in a high-level object-oriented programming language including algorithms, diagrams, problem solving, programming concepts, methods, control structures, arrays, and strings. Providing the students with problem-solving skills and their application to solve computer problems using MATLAB Simulink models. Developing the student's mind and enabling him to design any model in the MATLAB Simulink through the application of different programs in all fields. Teaching the student about Python program specifications to solve challenging problems of chemical engineering applications. Providing the students with modeling concepts and techniques from different disciplines to solve computer problems using Python language. Teaching the student how to incorporate AI-based modeling with first-principles-based models using Python language. Combine the traditional symbolic AI with the more recent data-driven AI using Python language. Teaching the students how to incorporate AI-based modeling in combination with first-principles-based models derived from the understanding of the physics and |
|---|--|
| Module Learning Outcomes | chemistry (and biology) of our products, processes, and systems. 17. Enabling the student to know the concepts of information technology by learning the basics of Simulink in MATLAB language. 18. Enabling the student to know MATLAB Simulink blocks and how to use each block inside the model. 19. Enabling the student how to use MATLAB Simulink language to design diagrams and models. 20. Enabling the student to think about the importance of the impact of the MATLAB Simulink language on the development of scientific research methods. 21. Enabling the student to know the concepts of information technology by learning the basics of Python programming language. 22. Enabling the student to solve particular problem in a scientific manner and present it clearly and correctly using Python programming language. 23. Enabling the student to think about the importance of programming using the Python language in facilitating contemporary life. 24. Enabling the student to think and follow the rapid development of the Python language. |
| Indicative Contents المحتويات الإرشادية | Indicative content includes the following: A brief history importance of Simulink in MATLAB, explaining the Simulink in MATLAB language using simplified programs [4 hrs]. Solving differential equations, derivative, integration, types of relational and logical operator's examples in MATLAB Simulink [8 hrs]. Plot functions using applied examples in MATLAB Simulink [4 hrs]. Arrays, matrices, and their arithmetic operations in MATLAB Simulink [8 hrs]. A brief history importance of Python language including basic principles like variables, arithmetic operations, strings, conditional statements, recursive phrases, functions, database design, and graphical interfaces [14 hrs]. Chemical engineering modeling approaches, knowledge representation, symbolic reasoning and inference, knowledge-based systems, and statistical data analysis [14 hrs]. Machine learning methods such as clustering, neural networks, random forests, Bayesian networks, and directed evolution [12 hrs]. |
| Course Description | This course provides students an idea of fundamental principles and applications of designing using MATLAB Simulink language. Types of blocks in MATLAB Simulink. Solving algebraic equations, evaluating polynomials, differentiation, limited integration, and unlimited integration in MATLAB Simulink. Relational and logical operators in MATLAB Simulink. Basic plotting including 2D and 3D drawing, adding titles, axis labels, annotations, specifying line styles and colors of the graph, subplots in MATLAB Simulink. Generation of arrays and matrices and their |

arithmetic operations, solving linear equations using matrices in MATLAB Simulink language.

Also, this course provides students an idea of fundamental principles and applications of Python language. Types of variables, numbers, arithmetic operations, strings, arrays, data objects, conditional statements, recursive phrases, Built-in functions, basic math functions, trigonometric functions, polynomials, database design, and graphical interfaces. For and while loops, visual studio code, subplots, write data to a file, read data from a file, and logging data to file. Chemical engineering applications, modeling approaches, machine learning methods, clustering, neural networks, random forests, and directed evolution in Python language.

| Learning and Teaching Strategies | | | | | |
|----------------------------------|--|--|--|--|--|
| | استراتيجيات التعلم والتعليم | | | | |
| Strategies | The teacher prepares lectures on the subject in paper and electronic form and presents them to the students. The teacher explains the theoretical and practical lectures using modern teaching and presentation methods. The teacher explains lectures in detail. The teacher requests periodic reports and homework assignments on the basic topics of the subject. The teacher encourages students to participate in discussions to clear up any misconceptions. The teacher asks questions and gives exercises that require analysis and conclusions related to lectures. The teacher uses different examples to connect the theoretical and practical sides. The teacher encourages students to do the practical part and work in groups. | | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | | |
|--|----|---|-----|--|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures 13 In class tests 4 Lab 43 Final Exam 4 Structured SWL (h/w) 4.2 | | | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل Libraray, dorm, home, memorize 14 Preparation for tests 14 Homework 8 | 36 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 2.4 | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 100 | | | |

Module Evaluation

تقييم المادة الدراسية

| | | Time/Number | Weight (Marks) | Week Due | Relevant Learning Outcome |
|----------------------|-----------------|---------------------|----------------|---------------------------|--|
| | Quizzes | 4 | 10% (10) | 3, 6, 9 and 12 | LO #1, 2, 3, 4, 5, 7, 8, 9, 10 and 11 |
| Formative assessment | Assignments | 6 | 10% (10) | 3, 5, 8, 11, 12 and 13 | LO #1, 4, 10 and 11 |
| | Projects / Lab. | 1 | 10% (10) | Continuous | All |
| | Report | 1 | 10% (10) | 13 | LO #3, #4 and #6 |
| Summative | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - #8 |
| assessment | Final Exam | 3hr | 50% (50) | 16 | All |
| Total assessment | | 100% (100 Marks) | | | |

| Delivery Plan (Weekly Syllabus) | | | | | |
|---------------------------------|--|--|--|--|--|
| | المنهاج الاسبوعي النظري | | | | |
| | Material Covered | | | | |
| Week 1 | A brief history importance of Simulink in MATLAB, types of blocks in MATLAB Simulink | | | | |
| Week 2 | Explaining the Simulink in MATLAB language using simplified programs | | | | |
| Week 3 | Solve algebraic equations, evaluate polynomials in MATLAB Simulink | | | | |
| Week 4 | Solve differentiation, limited integration, unlimited integration, relational and logical operators in MATLAB Simulink | | | | |
| Week 5 | Plots 2D functions, plots 3D functions using applied examples, adding titles, axis labels, annotations, specifying line styles and colors of the graph, subplots in MATLAB Simulink | | | | |
| Week 6 | Arrays, matrices and their arithmetic operations, solve linear equations using matrices in MATLAB Simulink | | | | |
| Week 7 | A brief history importance of Python language including basic principles like variables, arithmetic operations, strings, conditional statements, recursive phrases, functions, database design, and graphical interfaces | | | | |
| Week 8 | For and while loops, visual studio code, subplots, write data to a file, read data from a file, and logging data to file in Python language | | | | |
| Week 9 | Chemical engineering modeling approaches, knowledge representation in Python language | | | | |
| Week 10 | Symbolic reasoning and inference, knowledge-based systems in Python language | | | | |
| Week 11 | Statistical data analysis, machine learning methods: clustering in Python language | | | | |
| Week 12 | Machine learning methods: neural networks in Python language | | | | |
| Week 13 | Machine learning methods: random forests in Python language | | | | |
| Week 14 | Machine learning methods: Bayesian networks in Python language | | | | |
| Week 15 | Machine learning methods: directed evolution in Python language | | | | |
| Week 16 | Final Exam | | | | |

| | Delivery Plan (Weekly Lab. Syllabus) المنهاج الاسبوعي للمختبر | | |
|--------|--|--|--|
| | Material Covered | | |
| Week 1 | A brief history importance of Simulink in MATLAB, types of blocks in MATLAB Simulink | | |
| Week 2 | Explaining the Simulink in MATLAB language using simplified programs | | |
| Week 3 | Solve algebraic equations, evaluate polynomials in MATLAB Simulink | | |
| Week 4 | Solve differentiation, limited integration, unlimited integration, relational and logical operators in MATLAB Simulink | | |
| Week 5 | Plots 2D functions, plots 3D functions using applied examples, adding titles, axis labels, annotations, specifying line styles and colors of the graph, subplots in MATLAB Simulink | | |
| Week 6 | Arrays, matrices and their arithmetic operations, solve linear equations using matrices in MATLAB Simulink | | |
| Week 7 | A brief history importance of Python language including basic principles like variables, arithmetic operations, strings, conditional statements, recursive phrases, functions, database design, and graphical interfaces | | |
| Week 8 | For and while loops, visual studio code, subplots, write data to a file, read data from a file, and logging data to file in Python language | | |
| Week 9 | Chemical engineering modeling approaches, knowledge representation in Python language | | |
| Week10 | Symbolic reasoning and inference, knowledge-based systems in Python language | | |
| Week11 | Statistical data analysis, machine learning methods: clustering in Python language | | |
| Week12 | Machine learning methods: neural networks in Python language | | |
| Week13 | Machine learning methods: random forests in Python language | | |
| Week14 | Machine learning methods: Bayesian networks in Python language | | |
| Week15 | Machine learning methods: directed evolution in Python language | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|---|---------------------------|--|--|
| | Text | Available in the Library? | | |
| Required Texts | Chemical Engineering Computation with MATLAB®, 1st Edition, and Professor Yeong Koo Yeo teaches chemical engineering at the College of Engineering of Hanyang University, Seoul, South Korea. Essential MATLAB for Engineers and Scientists, Fourth Edition, Brian H. Hahn, Daniel T. Valentine. Essential MATLAB for Engineers and Scientists, Fifth Edition, Brian H. Hahn, Daniel T. Valentine. Venkatasubramanian, V., Artificial Intelligence in Process Engineering: Experiences from a Graduate Course. Chem. Eng. Educ., Fall Issue, 188–192, 1986. Venkatasubramanian, V., The promise of artificial intelligence in chemical engineering: Is it here, finally? AIChE J., 65: 466-478, 2019. | - | | |
| Recommended Texts | All solid scientific magazines and periodicals related to programming in the MATLAB Simulink and Python languages. | - | | |
| Websites | 3. http://www.cprogramming.com/4. http://www.mathworks.com/ | | | |





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|--------------|-------------------------|-------------|---------------------------|-------------------|--------|
| Module Title | Materials E | igineering | | Modu | ıle Delivery | |
| Module Type | Core | | | | Theory | |
| Module Code | CHE 212 | | | | Lecture | |
| ECTS Credits | 5 | | | | - Lab Tutorial | |
| SWL (hr/sem) | 125 | | | Seminar | | |
| Module Level | | 2 | Semester of | er of Delivery 2 | | 2 |
| Administering Department | | Chemical Engineering | College | Engine | eering | |
| Module Leader | Khalid Al Do | olaimy | e-mail | khalideng@uodiyala.edu.iq | | edu.iq |
| Module Leader's Acad. Title | | Lecturer | Module Le | ader's Q | ualification | MSc |
| Module Tutor None | | | e-mail | | | |
| Peer Reviewer Name | | None | e-mail | | | |
| Scientific Committee Approval Date | | | Version Nu | mber | 1.0 | |

| Relation with other Modules | | | | | |
|-----------------------------------|------|----------|--|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | | |
| Prerequisite module | None | Semester | | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | | | |
|--|--|--|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | | |
| Module Objectives أهداف المادة الدر اسية | This course aims to teach the students to learn and recognize the engineering materials and its properties. The student learns an idea about the followings; 1. Learn and understand Structure of materials-Various types of bonds, Crystalline | | | | |
| | Structure of Solids- concepts of unit cell and space lattice, and packing factor | | | | |

| 2. Learn and understand the X-ray diffraction for determining crystal structure; 3. Learn and understand Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-clastic and visco-clastic behavior of materials; 4. Learn and understand the Inorganic & organic amorphous materials and their structural & property characteristics; 5. Learn and understand the plastic materials and polymers, properties and uses of poly chylene, 6. Learn and understand the plastic materials and properties, metallurgical aspects of materials, Significance of microstructural features, 10. Learn and understand the Classification of materials and properties, metallurgical aspects of materials, Significance of microstructural features, 10. Learn and understand the effect of cooling and heating rates and ageing materials for mechanical load bearing applications, 11. Learn and understand the corrosion resistant materials, 12. Learn and understand the corrosion resistant materials, 12. Learn and understand the corrosion resistant materials, 13. Learn and understand the corrosion resistant materials. 14. Learn and understand the corrosion resistant materials, 15. Learn and understand the corrosion resistant materials. 16. Learn and understand the corrosion resistant materials. 17. Learn and understand the corrosion resistant materials. 18. Learn and understand the corrosion resistant materials. 19. Learn and understand the corrosion resistant materials. 10. Learn and understand the corrosion resistant materials. 10. Learn and understand the corrosion resistant materials. 11. Learn and understand the corrosion resistant materials. 12. Learn and understand the corrosion resistant materials. 12. Learn and understand the corrosion resistant materials. 13. Indicative content includes the following: 12. Learn and understand the corrosion resistant necessary and the properties of the materials of the properties of the materials properties of the properties of the properties of the properties o | | |
|---|--------------------|--|
| 1. Understanding and teaching students the concepts of Materials Engineering. 2. Enabling students to obtain knowledge and understanding about the properties of the materials and its applications. Indicative content includes the following: The Structure of materials-Various types of bonds [3 hrs] Crystalline Structure of Solids- concepts of unit cell [3 hrs] Space lattice and packing factor and X-ray diffraction for determining crystal structure [3 hrs] Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials [3 hrs] Crystals structure and phase diagrams [3 hrs] Metallurgical aspects of materials and significance of microstructural features [3 hrs] Classification of materials and properties [3 hrs] Classification of materials and properties and uses of poly ethylene Ceramic materials and polymers, properties and uses of poly ethylene Ceramic materials [3 hrs] Optical fibers, optical & optoelectronic properties of material Inorganic & organic amorphous materials and their structural & property characteristics [3 hrs] Civil, biomedical material and its applications [3 hrs] Course Description This course provides students an idea about the engineering materials, its | | Learn and understand Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials; Learn and understand the Inorganic & organic amorphous materials and their structural & property characteristics; Learn and understand Optical fibers. Learn and understand Crystals structure, phase diagrams, Learn about copper, aluminum and iron alloys, Learn and understand the plastic materials and polymers, properties and uses of poly ethylene, Learn and understand the Classification of materials and properties, metallurgical aspects of materials, Significance of microstructural features, Learn and understand the effect of cooling and heating rates and ageing materials for mechanical load bearing applications, Learn and understand the corrosion resistant materials, Learn about the Electrical, Electronic, Optical & Optoelectronic, civil, biomedical |
| Dutcomes 2. Enabling students to obtain knowledge and understanding about the properties of the materials and its applications. Indicative content includes the following: The Structure of materials-Various types of bonds [3 hrs] Crystalline Structure of Solids- concepts of unit cell [3 hrs] Space lattice and packing factor and X-ray diffraction for determining crystal structure [3 hrs] Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials [3 hrs] Crystals structure and phase diagrams [3 hrs] Metallurgical aspects of materials and significance of microstructural features [3 hrs] Classification of materials and properties [3 hrs] Copper, aluminum and iron alloys [3 hrs] Effect of cooling and heating rates and ageing materials for mechanical load bearing applications [3 hrs] Plastic materials and polymers, properties and uses of poly ethylene Ceramic materials [3 hrs] Optical fibers, optical & optoelectronic properties of material Inorganic & organic amorphous materials and their structural & property characteristics [3 hrs] Civil, biomedical material and its applications [3 hrs] Corrosion resistant, electrical and electronic, materials and its applications [3 hrs] This course provides students an idea about the engineering materials, its | Module Learning | |
| roperties of the materials and its applications. Indicative content includes the following: The Structure of materials-Various types of bonds [3 hrs] Crystalline Structure of Solids- concepts of unit cell [3 hrs] Space lattice and packing factor and X-ray diffraction for determining crystal structure [3 hrs] Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials [3 hrs] Crystals structure and phase diagrams [3 hrs] Metallurgical aspects of materials and significance of microstructural features [3 hrs] Classification of materials and properties [3 hrs] Classification of materials and properties and ageing materials for mechanical load bearing applications [3 hrs] Plastic materials and polymers, properties and uses of poly ethylene Ceramic materials [3 hrs] Optical fibers, optical & optoelectronic properties of material Inorganic & organic amorphous materials and their structural & property characteristics [3 hrs] Civil, biomedical material and its applications [3 hrs] Course Description This course provides students an idea about the engineering materials, its | _ | |
| Indicative content includes the following: The Structure of materials-Various types of bonds [3 hrs] Crystalline Structure of Solids- concepts of unit cell [3 hrs] Space lattice and packing factor and X-ray diffraction for determining crystal structure [3 hrs] Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials [3 hrs] Crystals structure and phase diagrams [3 hrs] Metallurgical aspects of materials and significance of microstructural features [3 hrs] Classification of materials and properties [3 hrs] Copper, aluminum and iron alloys [3 hrs] Effect of cooling and heating rates and ageing materials for mechanical load bearing applications [3 hrs] Plastic materials and polymers, properties and uses of poly ethylene Ceramic materials [3 hrs] Optical fibers, optical & optoelectronic properties of material Inorganic & organic amorphous materials and their structural & property characteristics [3 hrs] Civil, biomedical material and its applications [3 hrs] Course Description This course provides students an idea about the engineering materials, its | | |
| • The Structure of materials-Various types of bonds [3 hrs] • Crystalline Structure of Solids- concepts of unit cell [3 hrs] • Space lattice and packing factor and X-ray diffraction for determining crystal structure [3 hrs] • Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials [3 hrs] • Crystals structure and phase diagrams [3 hrs] • Metallurgical aspects of materials and significance of microstructural features [3 hrs] • Classification of materials and properties [3 hrs] • Copper, aluminum and iron alloys [3 hrs] • Effect of cooling and heating rates and ageing materials for mechanical load bearing applications [3 hrs] • Plastic materials and polymers, properties and uses of poly ethylene • Ceramic materials [3 hrs] • Optical fibers, optical & optoelectronic properties of material • Inorganic & organic amorphous materials and their structural & property characteristics [3 hrs] • Civil, biomedical material and its applications [3 hrs] • Corrosion resistant, electrical and electronic, materials and its applications [3 hrs] This course provides students an idea about the engineering materials, its | <u>.</u> , , | ^^ |
| Course Description | | The Structure of materials-Various types of bonds [3 hrs] Crystalline Structure of Solids- concepts of unit cell [3 hrs] Space lattice and packing factor and X-ray diffraction for determining crystal structure [3 hrs] Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials [3 hrs] Crystals structure and phase diagrams [3 hrs] Metallurgical aspects of materials and significance of microstructural features [3 hrs] Classification of materials and properties [3 hrs] Copper, aluminum and iron alloys [3 hrs] Effect of cooling and heating rates and ageing materials for mechanical load bearing applications [3 hrs] Plastic materials and polymers, properties and uses of poly ethylene Ceramic materials [3 hrs] Optical fibers, optical & optoelectronic properties of material Inorganic & organic amorphous materials and their structural & property characteristics [3 hrs] Civil, biomedical material and its applications [3 hrs] Corrosion resistant, electrical and electronic, materials and its applications [3 hrs] |
| | Course Description | |

| Learning and Teaching Strategies | | | | |
|----------------------------------|--|--|--|--|
| استراتيجيات التعلم والتعليم | | | | |
| | 1. Explain of Materials Engineering and topics by subject specialists. | | | |
| Strategies | 2. Present the topics which are related to Materials Engineering properties. | | | |
| | 3. Present the calculation of equilibrium diagram. | | | |

4. Explain and clarify some applications of the engineering materials.

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | |
|---|----|---|---|--|
| Structured SWL (h/sem) الحمل الدراسي المنتظم للطالب خلال الفصل In class lectures 45 Tutorial 15 Lab 30 | | Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا | 5 | |
| Final Exam 4 | | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل الحمل الدراسي غير المنتظم للطالب خلال الفصل Libraray, dorm, home, memorize 11 Preparation for tests 10 Homework 10 | 31 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 3 | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | _ | 125 | | |

| Module Evaluation تقييم المادة الدر اسية | | | | | |
|---|-----------------|---------------------|-------------------|-----------------|---------------------------|
| | | | Weight (Marks) | Week Due | Relevant Learning Outcome |
| | Quizzes | 2 | 10% (10) | 2, 5, 9 and 12 | LO #1-14 |
| Formative assessment | Assignments | 2 | 10% (10) | 3, 6, 10 and 13 | LO # 1-14 |
| | Projects / Lab. | 1 | 10% (10) | Continuous | All |
| | Seminars | 1 | 10% (10) | 14 | LO # 1 – 14 |
| Summative assessment | Midterm Exam | 2hr | 10% (10) | 7 | LO #1 - 7 |
| assessment | Final Exam | 4hr | 50% (50) | 16 | All |
| Total assessment | | 100% (100 Marks) | | | |

| Delivery Plan (Weekly Syllabus) | | | |
|---------------------------------|--|--|--|
| المنهاج الاسبوعي النظري | | | |
| | Material Covered | | |
| Week 1 | Structure of materials-Various types of bonds | | |
| Week 2 | Crystalline Structure of Solids- concepts of unit cell | | |
| Week 3 | Space lattice and packing factor and X-ray diffraction for determining crystal structure | | |

| Week 4 | Mechanical properties: Strength, hardness, toughness, ductility, brittleness of Engineering Materials; Elastic, an-elastic and visco-elastic behavior of materials | | |
|---|--|--|--|
| Week5 | Crystals structure and phase diagrams | | |
| Week 6 | Metallurgical aspects of materials and significance of microstructural features, | | |
| Week 7 | Mid-term Exam, Classification of materials and properties | | |
| Week 8 | Copper, aluminum and iron alloys | | |
| Week 9 | Effect of cooling and heating rates and ageing materials for mechanical load bearing applications | | |
| Week 10 | Plastic materials and polymers, properties and uses of poly ethylene | | |
| Week 11 | Ceramic materials | | |
| Week 12 | Optical fibers, optical & optoelectronic properties of material | | |
| Week 13 | Inorganic & organic amorphous materials and their structural & property characteristics | | |
| Week 14 | Civil, biomedical material and its applications. | | |
| Week 15 | Corrosion resistant, electrical and electronic, materials and its applications | | |
| Week 16 | Final Exam | | |
| Delivery Plan (Weekly Lab. Syllabus) المنهاج الاسبو عي للمختبر | | | |
| | Material Covered | | |
| Week 1 | Lab 1: Experiment of preparing a specimen to examine the microscopic structure of the metal | | |
| Week 2 | Lab 2: Experiment of preparing a specimen to examine the microscopic structure of the metal | | |
| Week 3 | Lab 3: Experiment of preparing a specimen to examine the microscopic structure of the metal | | |
| Week 4 | Lab 4: Experiment of preparing Microscopy sample support | | |
| Week 5 | Lab 5: Experiment of preparing Microscopy sample support | | |
| Week 6 | Lab 6: Experiment of preparing Microscopy sample support | | |
| Week 7 | Lab 7: Mid-term Exam | | |
| Week 8 | Lab 8: Hardness test experiment | | |
| Week 9 | Lab 9: Hardness test experiment | | |
| Week 10 | Lab 10: Hardness test experiment | | |
| Week 11 | Lab 11: Tensile test for metals experiment | | |
| Week 12 | Lab 12: Tensile test for metals experiment | | |
| Week 13 | Lab 13: Compression test for metals experiment | | |
| Week 14 | Lab 14: Compression test for metals experiment | | |
| Week 15 | Lab 15: Final test | | |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | | |
|--|--|-----|--|--|
| | Available in the Library? | | | |
| Required Texts | w.Bolton "Engineering Material Technology" 3ed Butterwoth-Heinemann, Oxford (1998) | Yes | | |
| Recommended Boxs | Brian S. Mitchell An Introduction To Material Engineering and Science For Chemical and Materials Engineers A. John Wiley & Sons Inc. Publication | Yes | | |
| Websites | http://www.kutub.info/library | | | |

APPENDIX:

| GRADING SCHEME مخطط الدرجات | | | | | |
|--------------------------------|-------------------------|-------------|-----------|---------------------------------------|--|
| Group | Grade | التقدير | Marks (%) | Definition | |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | |
| | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors | |
| Success Group (50 - 100) | C - Good | جيد | 70 - 79 | Sound work with notable errors | |
| (30 - 100) | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | |
| | | | | | |
| Note: | | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.





MODULE DESCRIPTION FORM

| Module Information معلومات المادة الدراسية | | | | | | |
|---|--------------------|--------------|----------------------|--------------------|-------------------------|--|
| Module Title | English Language 2 | | | Modu | ıle Delivery | |
| Module Type | Basic learning | g activities | | ⊠Theory | | |
| Module Code | UD21 | | | | □Lecture □Lab | |
| ECTS Credits | 2 | | | | ☐Tutorial ☐Practical | |
| SWL (hr/sem) | 50 | | | | □Seminar | |
| Module Level | | UGII | Semester of Delivery | | | |
| Administering D | epartment | All | College | College All | | |
| Module Leader | | | e-mail | | | |
| Module Leader's | Acad. Title | | Module Lea | der's Qu | alification | |
| Module Tutor Name(if available) | | ble) | e-mail | E-mail | | |
| Peer Reviewer Name | | Name | e-mail | E-mail | | |
| Scientific Committee Approval Date | | 3/11/2024 | Version Nu | mber | 1.0 | |

| Relation with other Modules | | | | | |
|-----------------------------------|------|----------|--|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | | |
| Prerequisite module None Semester | | | | | |
| Co-requisites module | None | Semester | | | |

| Module Aims, Learning Outcomes and Indicative Contents أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | |
|--|---|--|--|--|
| Module Objectives أهداف المادة الدر اسية | The module aims at enabling students to learn and understand the written and spoken form of English. It also aims at teaching functional English to learners and honing their reading, writing and listening skills> | | | |
| | Read and understand simple texts in English. | | | |
| Module Learning | 2. Answer simple comprehension questions and match sentences about texts. | | | |
| Outcomes | 3. Reconstruct texts by reordering sentences. | | | |
| | 4. Understand the main idea of a text. | | | |
| مخرجات التعلم للمادة الدراسية | 5. Identify specific information in a text. | | | |
| · | Writing and paraphrasing paragraphs. | | | |
| | Indicative content includes the following. | | | |
| | i) Grammar has a core place in language teaching and learning. | | | |
| Indicative Contents | ii) A wide variety of practice tasks in all the four skills are essential to language | | | |
| المحتويات الإرشادية | learning. | | | |
| | iii) Everyday expressions, particularly of spoken English, also need a place in the | | | |
| | syllabus. These can be functional, social, situational or idiomatic. | | | |

| Learning and Teaching Strategies | | | | |
|----------------------------------|--|--|--|--|
| استر اتيجيات التعلم والتعليم | | | | |
| | Headway's trusted methodology combines solid grammar and practice, vocabulary | | | |
| | development, and integrated skills with communicative role-plays and | | | |
| Strategies | personalization. | | | |
| | Authentic material from a variety of sources enables students to see new language in | | | |
| | context, and a range of comprehension tasks, language and vocabulary exercises, and | | | |
| | extension activities practice the four skills. 'Everyday English' and 'Spoken grammar' | | | |
| | sections practice real-world speaking skills, and a writing section for each unit at the | | | |
| | back of the book provides models for students to analyze and imitate. | | | |

| Student Workload (SWL) الحمل الدراسي للطالب محسوب لـ ١٥ اسبوعا | | | | | |
|--|----|---|-----|--|--|
| Structured SWL (h/sem) Structured SWL (h/w) الحمل الدراسي المنتظم للطالب أسبوعيا الحمل الدراسي المنتظم للطالب خلال الغصل | | | | | |
| Unstructured SWL (h/sem) الحمل الدراسي غير المنتظم للطالب خلال الفصل | 17 | Unstructured SWL (h/w) الحمل الدراسي غير المنتظم للطالب أسبوعيا | 1.1 | | |
| Total SWL (h/sem) الحمل الدراسي الكلي للطالب خلال الفصل | | 50 | | | |

Module Evaluation

تقييم المادة الدراسية

| | | | Time/Number Weight (Marks) | Week Due | Relevant Learning |
|------------------|--------------|----------------------------|----------------------------|------------|-------------------------------|
| | | Time/Number weight (Marks) | | Week Due | Outcome |
| Formative | Quizzes | 4 | 20% (5) | 3 , 6,8,11 | LO #1#3, #4#6, #7, #9 #11 |
| assessment | Assignments | 2 | 10% (5) | 5 and 12 | LO #3, #4 and #6, #7 |
| assessifient | Projects / | | | | |
| | Report | 1 | 10% (10) | 13 | LO #5, #8 and #10 |
| Summative | Midterm Exam | 1hr | 10% (10) | 7 | LO #1 - #7 |
| assessment | Final Exam | 2hr | 50% (50) | 15 | All |
| Total assessment | | | 100% (100 Marks) | | |

Delivery Plan (Weekly Syllabus)

المنهاج الاسبوعي النظري

| | Material Covered |
|--------|---|
| Week 1 | Course Introduction (Course material and objectives, learning outcomes, lessons and assessment discussed with the learners). |
| Week 2 | Unit 1. Getting to know you Reading: People, the great communicators' - the many ways we communicate Listening: Neighbours - Steve and Mrs Snell talk about each other as neighbours Speaking: Information gap - Joy Darling Writing: Informal letters - A letter to a pen friend Grammar: Tenses: present, past and future |
| Week 3 | Unit 2. The way we live Reading: 'Living in the USA' - three people talk about their experiences Listening: You drive me mad (but I love you)!' - what annoys you about the people in your life? Speaking: Information gap - people's Lifestyles. Exchanging information about immigrants to the USA Writing: Linking words, but, however. Describing a person Grammar: Present tense, present continuous |
| Week 4 | Unit 3. It all went wrong Reading: 'The burglars' friend'. Newspaper stories. A short story - 'The perfect crime Listening: A radio drama - 'The perfect crime Speaking: Information gap - Zoe's party. Telling stories Writing: Linking words: while, during, and for. Writing a story Grammar: Past tenses: Past Simple and past continuous |
| Week 5 | Unit 4. Let's go shopping! Reading: 'The best shopping street in the world' Listening: 'My uncle's a shopkeeper', Buying things Speaking: Town survey - the good things and bad things about living in your town, Discussion - attitudes to shopping p Writing: Filling in forms Grammar: Quantity, Articles |
| Week 6 | Assessment Test 1. Feedback and Remedial Work |

| | VI 1/2 200 |
|----------|---|
| | Unit 5. What do you want to do? |
| | Reading: 'Hollywood kids - growing up in Los Angeles ain't easy |
| Week 7 | Listening: A song - You've got a friend |
| | Speaking: What are your plans and ambitions? Being a teenager |
| | Writing: Writing a postcard |
| | Grammar: Verb patterns 1, future intentions |
| | Unit 6. Tell me! What's it like? |
| | Reading: A tale of two millionaires' - one was mean and one was generous |
| Week 8 | Listening: Living in another country — an interview with a girl who went to live in Sweden |
| | Speaking: Information gap – comparing cities |
| | Writing: Relative clauses 1 who/that/which/where. Describing a place |
| | Grammar: Comparative and superlative adjectives big, bigger, biggest, good, better, best |
| | Unit 7. Famous couples |
| | Reading: Celebrity interview from Hi! Magazine with the pop star and the footballer who are in love |
| Week 9 | Listening : An interview with the band Style |
| WEEK 5 | Speaking: Mingle - Find someone who Role play - interviewing a band |
| | Writing: Relative clauses 2 who/ which/ that as the object. Writing a biography |
| | Grammar: Present Perfect and Past Simple |
| | Unit 8. Do's and don'ts |
| | Reading: Problems and suggestions |
| W. J. 40 | Listening: Holidays in January - three people's advice on what to do in their country in January |
| Week 10 | Speaking: Jobs - a game. Discussion - house rules, Asking questions about place |
| | Writing: Writing letters- Formal letters |
| | Grammar: have (got) to, should, must |
| Week 11 | Assessment Test 2. Feedback and Remedial Work |
| Week 11 | |
| | Unit 9. Going places |
| | Reading: The world's first megalopolis - a city of 40 million people |
| Week 12 | Listening: Life in 2050 - an interview with Michio Kaku, Professor of Theoretical Physics |
| | Speaking: What will you do? Discussion - what will life be like in the 21st century? |
| | |
| | Writing: Linking words 2, Advantages and disadvantages |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses |
| | Writing: Linking words 2, Advantages and disadvantages |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world |
| Week 13 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants Writing: Writing a review of a book or film |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants Writing: Writing a review of a book or film Grammar: Passive |
| Week 14 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants Writing: Writing a review of a book or film Grammar: Passive Unit 12. Dreams and reality Reading: The vicar who's a ghostbuster |
| | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants Writing: Writing a review of a book or film Grammar: Passive Unit 12. Dreams and reality Reading: The vicar who's a ghostbuster Listening: An interview with a woman who heard voices |
| Week 14 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants Writing: Writing a review of a book or film Grammar: Passive Unit 12. Dreams and reality Reading: The vicar who's a ghostbuster Listening: An interview with a woman who heard voices Speaking: Giving advice - If I were you, I'd |
| Week 14 | Writing: Linking words 2, Advantages and disadvantages Grammar: Time and conditional clauses Unit 10. Scared to death Reading: 'Don't look down' - walking on a dangerous footpath, 'Into the wild' Listening: When I was young p80 It was just a joke - a boy called Jamie kidnapped his friend Speaking: 'When I was young' - talking about your childhood Writing: Writing letters Formal and informal letters 1 Grammar: Verb patterns 2 manage to do, used to do, go walking, Infinitives, Purpose Unit 11. Things that changed the world Reading: Three plants that changed the world - tobacco, sugar and cotton Listening: The world's most common habit: chewing gum Speaking: Exchanging information about three plants Writing: Writing a review of a book or film Grammar: Passive Unit 12. Dreams and reality Reading: The vicar who's a ghostbuster Listening: An interview with a woman who heard voices |

| Learning and Teaching Resources | | | | |
|---------------------------------|--|---------------------------|--|--|
| مصادر التعلم والتدريس | | | | |
| | Text | Available in the Library? | | |
| Required Texts | New Headway Pre-Intermediate by:John and Liz Soars. Oxford University Press | Yes | | |
| Recommended Texts | None | | | |
| Websites | https://apoyanblog.wordpress.com/wp-content/uploads/2016, intermediate-students-book.pdf | /09/new-headway-pre- | | |

APPENDIX:

| GRADING SCHEME مخطط الدرجات | | | | | | |
|--------------------------------|-------------------------|-------------|-----------|---------------------------------------|--|--|
| Group | Grade | التقدير | Marks (%) | Definition | | |
| | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | | |
| | B - Very Good | جيد جدا | 80 - 89 | Above average with some errors | | |
| Success Group (50 - 100) | C - Good | جيد | 70 - 79 | Sound work with notable errors | | |
| | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | | |
| | | | | | | |
| Note: | | | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.





MODULE DESCRIPTION FORM نموذج وصف المادة الدراسية

| Module Information | | | | | | |
|---------------------------------------|--------------------|-----------|-----------------------------|----------------------|-------------------------|-------|
| معلومات المادة الدراسية | | | | | | |
| Module Title | | | Module Delivery | | | |
| Module Type | Ba | S | ⊠Theory □Lecture □Lab | | | |
| Module Code | | | | | | |
| ECTS Credits | 2 | | | | □Tutorial □Practical | |
| SWL (hr/sem) | 50 | | | □Seminar | | |
| Module Level | UGII Se | | Semester o | f Delivery | | |
| Administering Department | | All | College | All | | |
| Module Leader | | | e-mail | | | |
| Module Leader's Acad. Title | | Lecturer | Module Lea | ader's Qualification | | Ph.D. |
| Module Tutor | Name(if available) | | e-mail | E-mail | | |
| Peer Reviewer Name | | Name | e-mail | E-mail | E-mail | |
| Scientific Committee Approval Date | | 3/11/2024 | Version Number 1.0 | | | |

| Relation with other Modules | | | | |
|-----------------------------------|------|----------|--|--|
| العلاقة مع المواد الدراسية الأخرى | | | | |
| Prerequisite module | None | Semester | | |
| Co-requisites module | None | Semester | | |

| Module Aims, Learning Outcomes and Indicative Contents | | | | |
|--|--|--|--|--|
| أهداف المادة الدراسية ونتائج التعلم والمحتويات الإرشادية | | | | |
| Module Objectives أهداف المادة الدر اسية | الهدف الأساس في تدريس اللغة العربية للأقسام العلمية: اخذ الطالب الى روعة بيان القران الكريم ، وادراكه يقينا ان التعبير القراني تعبير فني مقصود كل لفظة ، وكل حرف وضع وضعا فنياً مقصوداً تقويم اللسان العربي . واعتماد العربية الفصحى في الحديث والكتابة ، ولزيادة رصيد الطالب من ادب وتراث ، على تناول النصوص المختارة من العصور الأدبية المختلفة ، لزيادة رصيد ادب لتراث والادب المعاصر . التأكيد على دور الطالب في المتابعة واثراء المعرفة باللغة العربية وفنونها بجهد خاص ، ذا ما وضعنا مفاتيح المنهاج الدراسي لتقع على الطالب بعد ذلك مهمة فتح الأبواب والنوافذ ما وضعنا مفاتيح المعرفة الواسعة . في جعل العربية الفصيحة تحتل موضع الصدارة وتجاوز العامية ، خدمة الى لغتنا العربية المقدسة. وحفاظا على قوتها وجمالها . ومن نافلة القول في اهداف تدريس اللغة العربية : هي الجانب المحقق للوحدة وتلزمنا دوافع الوفاء بالحرص عليها والمحافظة على جوهرها . | | | |
| Module Learning Outcomes مخرجات التعلم للمادة الدراسية | يمكن ايجاز مخرجات العلم لمادة اللغة العربية العامة لاقسام غير الاختصاص بالنسبة لمرحلة الدراسة الأولى بالآي :- * علمه اليقين بان القران الكريم قد اعطى للغة قيمة عليا ، ومنزلة رفيعة . واجبه الحفاظ عليها وصيانتها من اللحن والخطا ، ومن كل ما يشوبها . * تمكين الطالب من قراءة النص القرائي بنفسه وتمكنه من معرفة ما في الايات الكريمة من امور لغوية وبلاغية ومعنوية ، وتمكنه من فحصها فحصاً دقيقاً . * سيتمكن الطالب من معرفة ما في لغته العربية من دقة في التعبير واحكام في الفن والعلو في الصنعة . * إظهار القواعد العلمية ، والأسس الفنية ، التي يقوم عليها العمل الأدبي في جزئياته وكلياته ولبيان الروح الجمالية والإبداعية التي تتجلى في النص الأدبي . * فهم طبيعة اللغة من حيث اعرابها (علم النحو) ، من الفعل والفاعل الذي يقع عليه فعل الفاعل ، والحال والتمييز والاوجه الاعرابية لما بعد التمييز . * تمكين الطالب من التفريق بين الافعال الثلاثية والرباعية والخماسية ، وتمكينه من التمييز بين الأفعال ومصادر الافعال ، فالفعل : لفظ يدل على حدث الى جانب دلالته على الزمن ، مصدره لا يدل على زمنه أي زمن فعله . * التعريف بالمفهوم الزمني الفني : للحداثة والمعاصرة في الأدب ، وإعطاء صورة لتطور أدب وصولا الى أدب العصر الحديث . * معرفة الطالب لمفهوم النثر العربي , وكيفية تطور النثر من القديم وكيفية تطور من القديم الى العصر الحديث . * تعريف الطالب بمفهوم الشعر الحر او " شعر التفعيلة " ورواد مدرسة الشعر الحر ، نازك الملائكة وبدر شاكر السياب ، وكيف تفجرت حركة الشعر كلون من الوان الاحتجاج على الواقع الفني العربي . | | | |
| Indicative Contents المحتويات الإرشادية | 1- الاهتمام بالكلام ، ومحاولة التعمق في دراسة قواعده واصوله وتاريخه . 2- الاهتمام بقواعد اللغة تاصيلا وتقعيدا ، وتاريخ ادبها تسجيلا وتدوينا ، ونقد نصوصها تفسيرا وتاويلا . | | | |

| 3- الفهم الدقيق لاصوات اللغة العربية ، ومعرفة مخارج الاصوات ، ومعرفة مخرج كل صوت من |
|--|
| اصوات العربية بطريقة مبسطة .وعلاقة الدال بالمدلول . |
| البحث في نشاة اللغة ، وعلاقتها بغيرها ، وخصائص اصواتها ، وابنية مفرداتها وتراكيبها . |
| 5- البحث في عناصر لهجاتها وتطور دلالالتها ، والعوامل التي اثرت فيها ، والقوانين التي تحكم الصلة |
| بين الفاظها . |
| 6- الاهتمام بالكلام ومحاولة التعمق في دراسة قواعده واصوله وتاريخه . |
| 7 توليد القدرة على تذوق النصوص ، وفتح نافذة القدرة على التحليل والتاويل ، ومعرفة ما يريد ان |
| يقوله النص الادبي . |
| 8- دراسة اساسيات النحو العربي وقواعد الجملة العربية . |
| 9- التعرف على تراكيب الجملة الفعلية والاسمية والمركبة . |
| |

| Learning and Teaching Resources مصادر التعلم والتدريس | | | |
|--|---|---------------------------|--|
| | Text | Available in the Library? | |
| | 1- التعبير ألقراني – الدكتور : فاضل السامرئي | | |
| | 2- شرح ابن عقيل على الفية ابن مالك – لابن عقيل | | |
| Required Texts | 3- علم اساليب البيان – غازي يموت. | | |
| | 4- اللغة العربية لاقسام غير الاختصاص مجموعة من اساتذة اللغة | | |
| | العربية . | | |
| Recommended | | | |
| Texts | | | |
| Websites | | | |

APPENDIX:

| GRADING SCHEME مخطط الدر جات | | | | | |
|---------------------------------|-------------------------|-------------|--|---------------------------------------|--|
| Group | Grade | التقدير | Marks (%) | Definition | |
| Success Group (50 - 100) | A - Excellent | امتياز | 90 - 100 | Outstanding Performance | |
| | B - Very Good | جيد جدا | 80 - 89 Above average with some errors | | |
| | C - Good | جيد | 70 - 79 | Sound work with notable errors | |
| | D - Satisfactory | متوسط | 60 - 69 | Fair but with major shortcomings | |
| | E - Sufficient | مقبول | 50 - 59 | Work meets minimum criteria | |
| Fail Group | FX – Fail | مقبول بقرار | (45-49) | More work required but credit awarded | |
| (0-49) | F – Fail | راسب | (0-44) | Considerable amount of work required | |
| | | | | | |
| Note: | | | | | |

NB Decimal places above or below 0.5 will be rounded to the higher or lower full mark (for example a mark of 54.5 will be rounded to 55, whereas a mark of 54.4 will be rounded to 54. The University has a policy NOT to condone "near-pass fails" so the only adjustment to marks awarded by the original marker(s) will be the automatic rounding outlined above.